



भारत 2023 INDIA  
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OFFICE OF THE DIRECTOR/PRINCIPAL  
GOVT. COLLEGE OF PHARMACY, ROHRU  
Tehsil Rohru, District Shimla, Himachal Pradesh- 171 207  
(NAAC B++ Accredited and ISO 9001:2015 Certified Institute)  
Recognized by PCI; Approved by AICTE, New Delhi; Affiliated to HPTU, Hamirpur  
Ph. No. 01781 - 241306; E-mail: [gcprohru@gmail.com](mailto:gcprohru@gmail.com); web: [www.gcprohru.ac.in](http://www.gcprohru.ac.in)

# Govt. College of Pharmacy, Rohru

## District Shimla, Himachal Pradesh 171207



(An ISO 9001:2015 Certified Institute)

Recognized by PCI/AICTE, New Delhi  
Affiliated to Himachal Pradesh Technical University, Hamirpur  
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# CURRICULUM AND COURSE CATALOGUE

## ACADEMIC YEAR: 2023-2024

  
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Govt. College of Pharmacy Rohru  
Disstt. Shimla HP-171207



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## Institute at Glance

The Government College of Pharmacy (GCP), Rohru was established in the year 2005 at Rohru, District Shimla, Himachal Pradesh. This college offers a four-year B. Pharmacy program since its inception. GCP, Rohru is currently affiliated with Himachal Pradesh Technical University, Hamirpur, Himachal Pradesh. The ongoing B. Pharmacy programme at the GCP, Rohru is approved by the All India Council for Technical Education (AICTE), New Delhi, and Pharmacy Council of India (PCI), New Delhi with a sanctioned intake of 40 students till the academic year 2021-22. The intake has been revised with the consent of AICTE and now the institute has an intake capacity of 60 students from the academic year 2022-23. The GCP, Rohru is the first Pharmacy College in the state of Himachal Pradesh that was established to impart Technical Education in the area of Pharmacy to aspirants, especially those who belong to the remote rural areas, keeping in view the immense employment opportunities in various sectors like academics, administration, corporate and Pharmaceutical Industries. The institute is experiencing fast academic growth and imparts quality-based pharmacy education. The success of the college is amply reflected by the academic results and excellent placement record. Every year nearly 25-40% of students in the final year qualify GPAT exam.

GCP, Rohru is having vast infrastructure, which includes modern buildings, labs, a library, an auditorium, and lush landscaping around the campus. It is focused on the all-round development of its students, which includes intellectual, physical, and mental health, along with personality development and improving communication skills. The college has well functional NSS unit and the students have been participating and representing the institute at the state and national levels.

## About the Programme

Pharmaceutical sciences offer an unparalleled opportunity to experience a science-driven course to equip a future leader in the multidisciplinary provision of health care. GCP Rohru is committed to making our graduates competent professional pharmacists to excel in the field of academics, research, industry, pharmacovigilance, drug discovery, and entrepreneurship.

  
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Pharmacy is recognized as an integral part of the healthcare system and this discipline combines a wide range of scientific concepts from various disciplines along with core pharmacy, which includes biochemistry, microbiology, communication skills, mathematics, biology, etc., which are critical for the development and discovery of new drugs, drug designing, and therapeutics.

GCP, Rohru is the oldest and one of the best pharmacy institutes in the state of Himachal Pradesh. The institute is not just focused on providing pharmacy education to its students but is also committed to providing research-oriented teaching pedagogy, and global scientific exposure to our students with scientific, and research capabilities, essential to becoming world-class entrepreneurs, pharmacists, and scientists.

The GCP Rohru follows the syllabus of PCI and the academic calendar of the Himachal Pradesh Technical University, Hamirpur. We have designed a curriculum for the undergraduate students pursuing Bachelor of Pharmacy in such a way that, along with achieving academic excellence, it provides an ample opportunity for the students to take part in extracurricular activities, especially under the NSS Unit, NCC Unit, and Red Ribbon Club of the college, for their overall personality development. This is reflected by the excellent academic results (> 97% every year), annual GPAT qualifiers (on average 25-40% of students every year), and state, national and international representations by our students in various camps of NCC/ NSS, awards won at scientific conferences, appreciation received from local and state bodies for excellent community work, etc.

## Vision

***“To emerge as a center of excellence in the field of pharmaceutical sciences by providing globally acceptable theoretical, practical and moral learning to develop excellent health care professionals, entrepreneurs, and researchers”***

Our aim here at the Govt. College of Pharmacy, Rohru is to emerge as a center of excellence in the field of pharmaceutical sciences in India and to achieve recognition at the national level for excellence in quality Pharmacy Education and Research by 2025. The institute is committed

  
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to delivering high-quality pharmacy education, experimental training, research exposure, and opportunities for physical, mental, and overall personality development to its students and faculty. Govt. College of Pharmacy, Rohru is devoted to creating a human resource of exceptional leadership quality, moral intellect, and professional thinking to serve national needs.

## Quality Policy

**M1:** To sincerely adopt the curriculum of the Himachal Pradesh technical University, Hamirpur (PCI and CBCS) while incorporating necessary amendments as per the dynamic changes and requirements of industry, academics, and research.

**M2:** To provide a constructive environment to students for research and development in different disciplines of Pharmaceutical Sciences and technology.

**M3:** To empower students with the necessary skills for becoming skilled and trained professionals in the pharmacy discipline.

**M4:** To enable students for intellectual achievements in terms of academics, research, innovation, and product development.

**M5:** To instill moral, personal, and professional ethics in our students.

## Institutional Core Values

- **Ethics:** To maintain the highest levels of ethical standards in teaching, learning, research, and service to humanity.
- **Creativity:** To develop a human resource with the highest level of creative potential by nurturing critical thinking, and problem-solving skills within social, ethical, and emotional capacities.
- **Sensitivity:** To remain sensitive to the key stakeholders such as institute, industry, Government, society, and country.

  
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- **Accountability:** To take accountability and responsibility for achieving common goals and objectives of the Institute and Government.
- **Awareness:** To develop social and moral awareness with a prime focus on serving humanity by contributing towards developing and maintaining a healthy environment, serving social causes, etc.
- **Responsibility:** To take up the responsibility to fulfill the regional mandate and requirements of the institute/ local community and work for the overall upliftment of the rural populace especially the weaker section of the society to enable them to escape poverty.

  
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## Programme Educational Objectives (PEOs)

PEO1	To develop graduates as competent professionals, who will be capable of utilizing and practicing professional aspects of pharmacy in academics, research institutions, self-practice, hospitals, government and non-government organizations, and the corporate sectors.
PEO2	To develop graduates with the capabilities to integrate the knowledge of basic sciences and pharmaceuticals to modify treatment approaches that reflect the breadth and scope of pharmacy practice and demonstrate clinical competency in evaluation, treatment planning, and implementation.
PEO3	To develop graduates with high morals and ethics with excellent leadership skills to sustain continual professional development through lifelong learning activities and to have the knowledge to support their endeavors.

## Programme Specific Outcomes (PSOs)

PSO1	Perform research on various disease pathophysiologies, herbal and synthetic medical aspects, and implement the Pharmaceutical knowledge in the drug discovery process.
PSO2	Graduates would be able to handle prescriptions, perform patient counseling, and study the effect of drugs on biological systems along with compounding, dispensing, selling, and marketing pharmaceutical drugs/products.
PSO3	Graduates would be able to use their expertise in various sectors of pharmaceutical product management like production, QA, QC, regulatory affairs, etc.

  
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## Programme Outcomes (PO)

PO	KEY CONCEPT	EXPLANATION
PO1	Pharmacy Knowledge	Possess knowledge and comprehension of the core and basic knowledge associated with the profession of pharmacy, including biomedical sciences; pharmaceutical sciences; behavioral, social, and administrative pharmacy sciences; and manufacturing practices.
PO2	Planning Abilities	Demonstrate effective planning abilities including time management, resource management, delegation skills, and organizational skills. Develop and implement plans and organize work to meet deadlines.
PO3	Problem analysis	Utilize the principles of scientific inquiry, thinking analytically, clearly, and critically, while solving problems and making decisions during daily practice. Find, analyze, evaluate and apply information systematically and shall make defensible decisions.
PO4	Modern tool usage	Learn, select, and apply appropriate methods and procedures, resources, and modern pharmacy-related computing tools with an understanding of the limitations.
PO5	Leadership skills	Understand and consider the human reaction to change, motivation issues, leadership, and team-building when planning changes required for fulfillment of practice, professional and societal responsibilities. Assume participatory roles as responsible citizens or leadership roles when appropriate to facilitate improvement in health and well-being.
PO6	Professional Identity	Understand, analyze and communicate the value of their professional roles in society (e.g. health care professionals, promoters of health, educators, managers, employers, and employees).
PO7	Pharmaceutical Ethics	Honor personal values and apply ethical principles in professional and social contexts. Demonstrate behavior that recognizes cultural and personal variability in values, communication, and lifestyles.


  
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		Use ethical frameworks; apply ethical principles while making decisions and take responsibility for the outcomes associated with the decisions.
PO8	Communication	Communicate effectively with the pharmacy community and with society at large, such as being able to comprehend and write effective reports, make effective presentations and documentation, and give and receive clear instructions.
PO9	The Pharmacist and society	Apply reasoning informed by the contextual knowledge to assess societal, health, safety, and legal issues and the consequent responsibilities relevant to the professional pharmacy practice.
PO10	Environment and sustainability	Understand the impact of professional pharmacy solutions in societal and environmental contexts, and demonstrate the knowledge of, and need for sustainable development.
PO11	Life-long learning	Recognize the need for, and have the preparation and ability to engage in independent and lifelong learning in the broadest context of technological change. Self-assess and use feedback effectively from others to identify learning needs and satisfy these needs on an ongoing basis.

  
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## Major Features of the Curriculum

At present, the institute is following the curriculum designed by the PCI. We have modified the curriculum in such a way that core aspects of PCI are met without any deviation and also incorporated special aspects so that the students are provided with ample opportunities for overall personal and professional development. Some major features of our curriculum are as follows:

- Curriculum comprising of all the courses and curriculum notified by PCI as well as the components for the overall holistic development of the candidate.
- Students are provided with special provisions for enrolment in the NSS unit to raise their sense of social responsibilities and need for working towards the betterment of society by keeping their personal interests at the secondary level with a motto of “Not Me But You”.
- Students are provided with special provisions for enrolment in the NCC unit to raise their sense of patriotism and work for the nation’s integrity and sovereignty.
- Students are provided with special provisions of funding for participating in conferences and workshops within and outside the state for providing them concrete exposure to research and global standards.
- Students are provided with opportunities with continuous motivations for industrial visits, hospital training, and educational tours, which have become an integral part of our curriculum.
- To impart high competency to the students, the curriculum offers distinct ability enhancement and value-added courses.
- Students are provided with opportunities to interact with professionals from academics, research, industry, defense services, social workers, etc. to uplift their overall personal and professional development.

  
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## Course Structure

### Semester wise credits distribution

Semester	Credit Points
I	27/29 <sup>§</sup> /30 <sup>#</sup>
II	29
III	26
IV	28
V	26
VI	26
VII	24
VIII	22
Extracurricular/ Co curricular activities	01*
<b>Total credit points for the program</b>	<b>209/211<sup>§</sup>/212<sup>#</sup></b>

\* The credit points assigned for extracurricular and or co-curricular activities shall be given by the Principals of the colleges and the same shall be submitted to the University. The criteria to acquire this credit point shall be defined by the colleges from time to time.

<sup>§</sup>Applicable ONLY for the students studied Physics / Chemistry / Botany / Zoology at HSC and appearing for Remedial Mathematics course.

<sup>#</sup>Applicable ONLY for the students studied Mathematics / Physics / Chemistry at HSC and appearing for Remedial Biology course.

  
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## 1<sup>st</sup> Year (Semester I)

### PCI Syllabus

S. N.	Category	Paper Code	Subject	L	T	P/D	Credits	Evaluation Scheme				
								Internal Assessment			ESE	Subject Total
								CT	TA	Total		
<b>Theory:</b>												
1	PC	BP101 T	Human Anatomy and Physiology I – Theory	3	1	-	4	15	10	25	75	100
2	PC	BP102 T	BP102T Pharmaceutical Analysis I – Theory	3	1	-	4	15	10	25	75	100
3	PC	BP103 T	BP103T Pharmaceutics I – Theory	3	1	-	4	15	10	25	75	100
4	PC	BP104 T	Pharmaceutical Inorganic Chemistry – Theory	3	1	-	4	15	10	25	75	100
5	MC	BP105 T	BP105T Communication skills – Theory *	2	-	-	2	10	5	15	35	50
6	MC	BP106 RBT BP106 RMT	Remedial Biology/ Remedial Mathematics – Theory*	2	-	-	2	10	5	15	35	50
<b>Labs:</b>												
1	PC	BP107 P	Human Anatomy and Physiology – Practical	-	-	4	2	10	5	15	35	50
2	P C	BP108 P	BP108P Pharmaceutical Analysis I – Practical	-	-	4	2	10	5	15	35	50
3	P C	BP109 P	BP109P Pharmaceutics I – Practical	-	-	4	2	10	5	15	35	50
4	PC	BP110 P	Pharmaceutical Inorganic Chemistry – Practical	-	-	4	2	10	5	15	35	50
	MC	BP111 P	Communication skills – Practical*	-	-	2	1	5	5	10	15	25
	MC	BP112 RBP	Remedial Biology – Practical*	-	-	2	1	5	5	10	15	25
<b>Total</b>				<b>16</b>	<b>4</b>	<b>18/20</b>	<b>29/30</b>					<b>675/725</b> S/ 750#
<b>Total work Load=27 Hrs.</b>							<b>Total Credit = 27/295/30#</b>					

#Applicable ONLY for the students who have studied Mathematics / Physics / Chemistry at HSC and appearing for Remedial Biology (RB)course.

SApplicable ONLY for the students who have studied Physics / Chemistry / Botany / Zoology at HSC and appearing for Remedial Mathematics (RM)course.

\* Non University Examination (NUE)

#### Legend:

L	Lecture	ESE	End Semester Examination
T	Tutorial	PC	Program Core Courses
P	Practical	FC	Foundation Courses
CT	Class Test	HS	Humanities and social science
TA	Teacher's Assessment	MC	Mandatory Course

  
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## 1<sup>st</sup> Year (Semester II)

### PCI Syllabus

S. N.	Category	Paper Code	Subject	L	T	P/D	Credits	Evaluation Scheme				
								Internal Assessment			ESE	Subject Total
								CT	TA	Total		
<b>Theory:</b>												
1	MC	BP201 T	Human Anatomy and Physiology II – Theory	3	1	-	4	15	10	25	75	100
2	PC	BP202 T	Pharmaceutical Organic Chemistry I – Theory	3	1	-	4	15	10	25	75	100
3	PC	BP203 T	Biochemistry – Theory	3	1	-	4	15	10	25	75	100
4	PC	BP204 T	Pathophysiology – Theory	3	1	-	4	15	10	25	75	100
5	PC	BP205 T	Computer Applications in Pharmacy – Theory *	3	-	-	3	15	10	25	75	100
6	FC	BP206 T	Environmental sciences – Theory *	3	-	-	3	15	10	25	75	100
<b>Labs:</b>												
1	PC	BP207 P	Human Anatomy and Physiology II – Practical		-	4	2	10	5	15	35	50
2	PC	BP208 P	Pharmaceutical Organic Chemistry I- Practical		-	4	2	10	5	15	35	50
3	PC	BP209 P	Biochemistry – Practical		-	4	2	10	5	15	35	50
4	PC	BP210 P	Computer Applications in Pharmacy – Practical*		-	2	1	5	5	10	15	25
<b>Total</b>				<b>18</b>	<b>0</b>	<b>14</b>	<b>29</b>					<b>725</b>
<b>Total Work Load 32 Hrs.</b>				<b>Total Credit 29</b>								

\* The subject experts at college level shall conduct examinations.

#### Legend:

L	Lecture	ESE	End Semester Examination
T	Tutorial	PC	Program Core Courses
P	Practical	FC	Foundation Courses
CT	Class Test	HS	Humanities and social science
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## 2<sup>nd</sup> Year (Semester III)

### PCI Syllabus

S. N.	Category	Paper Code	Subject	L	T	P/D	Credits	Evaluation Scheme				
								Internal Assessment			ESE	Subject Total
								CT	TA	Total		
<b>Theory:</b>												
1	PC	BP301 T	Pharmaceutical Organic Chemistry II – Theory	3	1		4	15	10	25	75	100
2	PC	BP302 T	Physical Pharmaceutics I – Theory	3	1		4	15	10	25	75	100
3	PC	BP303 T	Pharmaceutical Microbiology – Theory	3	1		4	15	10	25	75	100
4	PC	BP304 T	Pharmaceutical Engineering – Theory	3	1		4	15	10	25	75	100
<b>Labs:</b>												
1	PC	BP305 P	Pharmaceutical Organic Chemistry II – Practical			4	2	10	5	15	35	50
2	PC	BP306 P	Physical Pharmaceutics I – Practical			4	2	10	5	15	35	50
3	PC	BP307 P	Pharmaceutical Microbiology – Practical			4	2	10	5	15	35	50
4	PC	BP308P	Pharmaceutical Engineering – Practical			4	2	10	5	15	35	50
<b>Total</b>				<b>12</b>	<b>4</b>	<b>16</b>	<b>24</b>					<b>600</b>
<b>Total Work Load=32 Hrs.</b>						<b>Total Credit 24</b>						

#### Legend:

L	Lecture	ESE	End Semester Examination
T	Tutorial	PC	Program Core Courses
P	Practical	FC	Foundation Courses
CT	Class Test	HS	Humanities and social science
TA	Teacher's Assessment	MC	Mandatory Course

  
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## 2<sup>nd</sup> Year (Semester IV)

### PCI Syllabus

S. N.	Category	Paper Code	Subject	L	T	P/D	Credits	Evaluation Scheme				
								Internal Assessment			ESE	Subject Total
								CT	TA	Total		
<b>Theory:</b>												
1	FC	BP401 T	Pharmaceutical Organic Chemistry III – Theory	3	1		4	15	10	25	75	100
2	PC	BP402 T	Medicinal Chemistry I – Theory	3	1		4	15	10	25	75	100
3	PC	BP403 T	Physical Pharmaceutics II – Theory	3	1		4	15	10	25	75	100
4	PC	BP404 T	Pharmacology I – Theory	3	1		4	15	10	25	75	100
5	PC	BP405 T	Pharmacognosy and Phytochemistry I – Theory	3	1		4	15	10	25	75	100
<b>Labs:</b>												
1	CC	BP406 P	Medicinal Chemistry I – Practical			4	2	10	5	15	35	50
2	PC	BP407 P	Physical Pharmaceutics II – Practical			4	2	10	5	15	35	50
3	PC	BP408 P	Pharmacology I – Practical			4	2	10	5	15	35	50
4	PC	BP409 P	Pharmacognosy and Phytochemistry I – Practical			4	2	10	5	15	35	50
<b>Total</b>				<b>15</b>	<b>5</b>	<b>16</b>	<b>28</b>					<b>700</b>
<b>Total Working = 36 Hrs.</b>							<b>Total Credit = 28</b>					

#### Legend:

L	Lecture	ESE	End Semester Examination
T	Tutorial	PC	Program Core Courses
P	Practical	FC	Foundation Courses
CT	Class Test	HS	Humanities and social science
TA	Teacher's Assessment	MC	Mandatory Course

  
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### 3<sup>rd</sup> Year (Semester V)

### PCI Syllabus

S. N.	Category	Paper Code	Subject	L	T	P/D	Credits	Evaluation Scheme				
								Internal Assessment			ESE	Subject Total
								CT	TA	Total		
<b>Theory:</b>												
1	PC	BP 501T	Medicinal Chemistry II – Theory	3	1	-	4	15	10	25	75	100
2	PC	BP 502T	Industrial Pharmacy I- Theory	3	1	-	4	15	10	25	75	100
3	PC	BP 503T	Pharmacology II – Theory	3	1	-	4	15	10	25	75	100
4	PC	BP 504T	Pharmacognosy and Phytochemistry II- Theory	3	1	-	4	15	10	25	75	100
5	PC	BP 505T	Pharmaceutical Jurisprudence – Theory	3	1	-	4	15	10	25	75	100
<b>Labs:</b>												
1	PC	BP 506P	Industrial Pharmacy I – Practical	-	-	4	2	10	5	15	35	50
2	PC	BP 507P	Pharmacology II – Practical	-	-	4	2	10	5	15	35	50
3	PC	BP 508P	Pharmacognosy and Phytochemistry II – Practical	-	-	4	2	10	5	15	35	50
<b>Total</b>				<b>15</b>	<b>5</b>	<b>12</b>	<b>26</b>					<b>650</b>
<b>Total Work Load=32 Hrs.</b>							<b>Total Credit 26</b>					

#### Legend:

L	Lecture	ESE	End Semester Examination
T	Tutorial	PC	Program Core Courses
P	Practical	FC	Foundation Courses
CT	Class Test	HS	Humanities and social science
TA	Teacher's Assessment	MC	Mandatory Course

  
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### 3<sup>rd</sup> Year (Semester VI)

### PCI Syllabus

S. N.	Paper Code	Subject	L	T	P/D	Credits	Evaluation Scheme					
							Internal Assessment			ESE	Subject Total	
							CT	TA	Total			
<b>Theory:</b>												
1	PC	BP 601T	Medicinal Chemistry III – Theory	3	1	-	4	15	10	25	75	100
2	PC	BP 602T	Pharmacology III – Theory	3	1	-	4	15	10	25	75	100
3	PC	BP 603T	Herbal Drug Technology – Theory	3	1	-	4	15	10	25	75	100
4	PC	BP 604T	Biopharmaceutics and Pharmacokinetics – Theory	3	1	-	4	15	10	25	75	100
5	PC	BP 605T	Pharmaceutical Biotechnology – Theory	3	1	-	4	15	10	25	75	100
6	PC	BP 606T	Quality Assurance – Theory	3	1	-	4	15	10	25	75	100
<b>Labs:</b>												
1	PC	BP 607P	Medicinal chemistry III – Practical	-	-	4	2	10	5	15	35	50
2	PC	BP 608P	Pharmacology III – Practical	-	-	4	2	10	5	15	35	50
3	PC	BP 609P	Herbal Drug Technology – Practical	-	-	4	2	10	5	15	35	50
<b>Total</b>				<b>18</b>	<b>6</b>	<b>12</b>	<b>30</b>					<b>750</b>
<b>Total Work Load=36 Hrs.</b>							<b>Total Credit 30</b>					

Note

Legend:

L	Lecture	ESE	End Semester Examination
T	Tutorial	PC	Program Core Courses
P	Practical	FC	Foundation Courses
CT	Class Test	HS	Humanities and social science
TA	Teacher's Assessment	MC	Mandatory Course

  
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### 4<sup>th</sup> Year (Semester VII)

### PCI Syllabus

S. N.	Category	Paper Code	Subject	L	T	P/ D	Credits	Evaluation Scheme				
								Internal Assessment			ESE	Subject Total
								CT	TA	Total		
<b>Theory:</b>												
1	PC	BP 701T	Instrumental Methods of Analysis – Theory	3	1	-	4	15	10	25	75	100
2	PC	BP 702T	Industrial PharmacyII – Theory	3	1	-	4	15	10	25	75	100
3	PC	BP 703T	Pharmacy Practice – Theory	3	1	-	4	15	10	25	75	100
4	PC	BP704 T	Novel Drug Delivery System – Theory	3	1	-	4	15	10	25	75	100
<b>Labs:</b>												
1	PC	BP 705P	Instrumental Methods of Analysis – Practical			4	2	10	5	15	35	50
2	PC	BP 706P	Practice School*			12	6	20	30	50	50	100
<b>Total</b>				<b>12</b>	<b>4</b>	<b>16</b>	<b>24</b>					<b>600</b>
<b>Total Work Load=32 Hrs.</b>							<b>Total Credit 24</b>					

\* Non University Examination (NUE)

  
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**4<sup>th</sup> Year (Semester VIII)**

### PCI Syllabus

S. N.	Category	Paper Code	Subject	L	T	P/D	Credits	Evaluation Scheme								
								Internal Assessment			ESE	Subject Total				
								CT	TA	Total						
<b>Theory:</b>																
1	PC	BP 801T	Biostatistics and Research Methodology	3	1	-	4	15	10	25	75	100				
2	PC	BP 802T	Social and Preventive Pharmacy	3	1	-	4	15	10	25	75	100				
3	PC	BP 803ET	Pharma Marketing Management	3 +3 =6	1 +1 =2	-	4+4=8	15	10	25	75	100				
4	PC	BP 804ET	Pharmaceutical Regulatory Science					15	10	25	75	100				
5	PC	BP 805ET	Pharmacovigilance					15	10	25	75	100				
6	PC	BP 806ET	Quality Control and Standardization of Herbals					15	10	25	75	100				
7	PC	BP 807ET	Computer Aided Drug Design					15	10	25	75	100				
8	PC	BP 808ET	Cell and Molecular Biology					15	10	25	75	100				
9	PC	BP 809ET	Cosmetic Science					15	10	25	75	100				
10	PC	BP 810ET	Experimental Pharmacology					15	10	25	75	100				
11	PC	BP 811ET	Advanced Instrumentation Techniques					15	10	25	75	100				
12	PC	BP 812ET	Dietary Supplements and Nutraceuticals					15	10	25	75	100				
<b>Labs:</b>																
1	PC	BP 813 PW	Project Work					-	-	12	6	-	-	-	150	150
<b>Total</b>				<b>12</b>	<b>4</b>	<b>12</b>	<b>22</b>					<b>550</b>				
<b>Total Work Load=28</b>							<b>Total Credit 22</b>									

**Legend:**

L	Lecture	ESE	End Semester Examination
T	Tutorial	PC	Program Core Courses
P	Practical	FC	Foundation Courses

  
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### Value Added Courses

S. No.	Type	Remarks
1	NSS Unit Course	<ul style="list-style-type: none"><li>Institute has a dedicated NSS Unit with a sanctioned intake of 50 volunteers every year from Himachal Pradesh University, Shimla.</li><li>The minimum duration is of three years during which volunteers participate and learn various aspects of community service and take part in certified national and state camps.</li></ul>
2	NCC Unit Course	<ul style="list-style-type: none"><li>Institute has a dedicated NCC Unit with a sanctioned intake of 56 cadets from 8 HP Bn NCC Rampur.</li><li>The minimum duration is of three years for a C certification.</li><li>During these three years, cadets are provided with lessons on basic military drills and other aspects and take part in certified national and state camps.</li></ul>

  
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## ACADEMIC CALANDER FOR THE ACADEMIC YEAR 2023-2024

### HP TECHNICAL UNIVERSITY, HAMIRPUR 177 001, HP Academic Calendar for the Session 2023-24

#### Odd Semester: 2023-24

SN	Event(s)	Date(s)
1	Reporting of Faculty in respective Colleges	27-07-2023
2	Industrial/Institutional Training:	
	After 2 <sup>nd</sup> Semester: MBA	19-07-23 to 20-08-23
	After 6 <sup>th</sup> semester: B.Tech.	27-07-23 to 07-09-23
3	*Registration [except 3 <sup>rd</sup> Sem MBA & 7 <sup>th</sup> Sem B. Tech.]	01-08-23 to 03-08-23
4	Commencement of classes [except 3 <sup>rd</sup> Sem MBA & 7 <sup>th</sup> Sem B. Tech.]	02-08-23
5	*Registration for 3 <sup>rd</sup> Sem (MBA)	21-08-23 to 23-08-23
6	Commencement of classes of 3 <sup>rd</sup> Sem (MBA)	21-08-23
7	*Registration for 7 <sup>th</sup> Sem (B. Tech.)	08-09-23 to 11-09-23
8	Commencement of classes of 7 <sup>th</sup> Sem (B.Tech.)	08-09-23
9	Induction programme for 1 <sup>st</sup> year students	02-08-23 to 16-08-23
10	Mid-Sem Exams-I (1 <sup>st</sup> Periodical Exams) except 3 <sup>rd</sup> Sem MBA & 7 <sup>th</sup> Sem B.Tech.	18-09-23 to 21-09-23
11	HPTU Youth Festival	06-10-23 to 08-10-23
12	Mid-Sem Exams-I (1 <sup>st</sup> Periodical Exams) for 3 <sup>rd</sup> Sem (MBA)	27-09-23 to 30-09-23
13	Mid-Sem Exams-I (1 <sup>st</sup> Periodical Exams) for 7 <sup>th</sup> Sem (B.Tech.)	11-10-23 to 13-10-23
14	Diwali Vacations	09-11-23 to 13-11-23
15	Mid-Semester Exams-I (2 <sup>nd</sup> Periodical Exams) for all semesters	15-11-23 to 18-11-23
16	End of Teaching Work for all semesters	01-12-23
17	<ul style="list-style-type: none"> <li>Reporting of Shortage of attendance cases, and</li> <li>Display of internal assessment awards</li> </ul>	02-12-23
18	End Semester Practical Examinations	04-12-23 to 07-12-23
19	End semester Theory Examinations	08-12-23 to 24-12-23
20	Vacations	25-12-23 to 13-01-2024
21	Reporting of Faculty in Respective Colleges	15-01-24

#### Even Semester: 2023-24

SN	Event(s)	Date(s)
1	Reporting of Faculty in respective Colleges	15-01-2024
2	*Registration of UG and PG Courses	16-01-24 to 18-01-24
3	Commencement of classes	16-01-24
4	Mid-Semester Examinations-I (1 <sup>st</sup> Periodical Examinations)	26-02-24 to 29-02-24
5	HPTU Sports Meet	15-03-24 to 17-03-24
6	Mid-Semester Examinations-I (2 <sup>nd</sup> Periodical Examinations)	03-04-24 to 06-04-24
7	End of Teaching work	30-04-24
8	<ul style="list-style-type: none"> <li>Reporting of Shortage of attendance cases, and</li> <li>Display of internal assessment awards</li> </ul>	01-05-24
9	End Semester Practical Examinations	02-05-24 to 06-05-24
10	End semester Theory Examinations	08-05-24 to 22-05-24
11	Vacations	26-05-24 to 27-06-24
12	<b>Declaration of results of last semester of all courses</b>	<b>Last Week of July 2024</b>
13	Industrial/Institutional Training of Students: 4 to 6 weeks (as applicable)	24-05-24 to 10-07-24
14	Reporting of Faculty in respective Colleges	28-06-24

\* Fine @ Rs. 50/- per day, upto 10 days from the last date of registration, is chargeable on account of late registration as per

  
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07.03.2024

HP TECHNICAL UNIVERSITY, HAMIRPUR 177 001, HP

Academic Calendar: 2023-24

Odd Semester: 2023-24 session (i.e., 1<sup>st</sup> semester of 2023-24)

SN	Event(s)	Date(s)
1	Reporting of Faculty in respective Colleges	27-07-2023
2	Industrial/Institutional/Practical Training:	
	After 2 <sup>nd</sup> Semester: MBA	19-07-23 to 20-08-23
	After 4 <sup>th</sup> Semester: B. Tech. (CSE) & 6 <sup>th</sup> Semester: B. Pharm.	27-07-23 to 27-08-23
	After 6 <sup>th</sup> semester: B. Tech. (All Branches)	27-07-23 to 07-09-23
3	Registration of all students for all courses; <b>except</b> 3 <sup>rd</sup> Sem MBA, 5 <sup>th</sup> sem B. Tech.-CSE, 7 <sup>th</sup> sem B. Tech. & 7 <sup>th</sup> sem B.Pharm.	01-08-23 to 03-08-23
4	Commencement of all classes; <b>except</b> 3 <sup>rd</sup> Sem MBA, 5 <sup>th</sup> sem B. Tech.-CSE, 7 <sup>th</sup> sem B. Tech. & 7 <sup>th</sup> sem B. Pharm.]	02-08-23
5	Induction programme for 1 <sup>st</sup> year students	02-08-23 to 11-08-23
6	*Registration for 3 <sup>rd</sup> Sem (MBA)	21-08-23 to 23-08-23
7	Commencement of classes of 3 <sup>rd</sup> Sem (MBA)	21-08-23
8	*Registration for 5 <sup>th</sup> Sem B. Tech. (CSE) & 7 <sup>th</sup> Sem B. Pharm.	28-08-23 to 30-08-23
9	Commencement of classes of 5 <sup>th</sup> Sem B.Tech. (CSE) & 7 <sup>th</sup> Sem B. Pharm.	28-08-23
10	*Registration for 7 <sup>th</sup> Sem (B. Tech.)	08-09-23 to 11-09-23
11	Commencement of classes of 7 <sup>th</sup> Sem (B.Tech.)	08-09-23
12	Mid-Sem Tests-I (1 <sup>st</sup> Periodical Exams) of all; <b>except</b> 3 <sup>rd</sup> Sem MBA, 5 <sup>th</sup> Sem B. Tech.-CSE and 7 <sup>th</sup> Sem of B. Tech. & B. Pharm.	18-09-23 to 21-09-23
13	HPTU Youth Festival	06-10-23 to 08-10-23
14	Mid-Sem Tests-I (1 <sup>st</sup> Periodical Exams) of 3 <sup>rd</sup> Sem (MBA)	27-09-23 to 30-09-23
15	Mid-Sem Tests-I (1 <sup>st</sup> Periodical Exams) of 5 <sup>th</sup> Sem (B.Tech.-CSE) & 7 <sup>th</sup> Sem B. Pharm.	03-10-23 to 06-10-23
16	Mid-Sem Tests-I (1 <sup>st</sup> Periodical Exams) of 7 <sup>th</sup> Sem (B.Tech.)	11-10-23 to 14-10-23
17	Diwali Vacations	09-11-23 to 13-11-23
18	Mid-Semester Tests-II (2 <sup>nd</sup> Periodical Exams) of all courses & semesters	15-11-23 to 18-11-23
19	End of Teaching Work for all semesters	30-11-23
20	• Reporting of Shortage of attendance cases, and • Display of internal assessment awards	02-12-23
21	End Semester Practical Examinations	04-12-23 to 07-12-23
22	End semester Theory Examinations	08-12-23 to 24-12-23
23	Vacations	25-12-23 to 13-01-2024
24	Reporting of Faculty in Respective Colleges	15-01-24

Revised for Even Semester of 2023-24 session (i.e., 2<sup>nd</sup> semester of 2023-24)

SN	Event(s)	Earlier Date(s)	Revised Date(s)
1	Reporting of Faculty in respective Colleges	15-01-2024	No change
2	Registration of UG and PG Courses	16-01-24 to 18-01-24	-do-
3	Commencement of classes	16-01-24	-do-
4	Mid-Semester Tests-I (1 <sup>st</sup> Periodical Examinations)	26-02-24 to 29-02-24	-do-
5	HPTU Sports Meet	15-03-24 to 17-03-24	28-03-24 to 30-03-24
6	Mid-Semester Tests-II (2 <sup>nd</sup> Periodical Examinations)	03-04-24 to 06-04-24	06-04-24 to 10-04-24
7	End of Teaching work	29-04-24	04-05-24
8	• Reporting of Shortage of attendance cases, and • Display of internal assessment awards	01-05-24	06-05-24
9	End Semester Practical Examinations	02-05-24 to 06-05-24	06-05-24 to 09-05-24
10	End semester Theory Examinations	08-05-24 to 22-05-24	13-05-24 to 31-05-24
11	Vacations	26-05-24 to 27-06-24	10-06-24 to 12-07-24
12	Declaration of results of last semester of all courses	Last Week of July, 2024	First Week of August, 2024
13	Industrial/Institutional Training of Students: 4 to 6 weeks (wherever applicable)	24-05-24 to 10-07-24	02-06-24 to 22-07-24
14	Reporting of Faculty in respective Colleges	28-06-24	15-07-24

Dean (Academic) 07.03.24

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**TIME TABLE FOR THE ACADEMIC YEAR 2022-2023**

GOVT. COLLEGE OF PHARMACY, ROHRU DISTT. SHIMLA (H.P.)		NOTICE		Date: .....	
TIME TABLE FOR B. Pharm. (To be effective from 7 AUG., 2023)					
First Semester (PCI)					
Day	Time	Subject	Remarks	Day	Time
MON	09:00-10:00	Ph. Anal-1 (HK)	Lib/Revision	MON	09:00-10:00
TUE	10:00-11:00	Ph. Anal-2 (HK)	NS/NCC	TUE	10:00-11:00
WED	11:00-12:00	Ph. Anal-3 (HK)	Ph.C (C)	WED	11:00-12:00
THU	12:00-1:00	Ph. Anal-4 (HK)	Comm. Skills (*)	THU	12:00-1:00
FRI	1:00-2:00	Ph. Anal-5 (HK)	Comm. Skills (*)	FRI	1:00-2:00
SAT	2:00-3:00	Ph. Anal-6 (HK)	Comm. Skills (*)	SAT	2:00-3:00
Second Semester (PCI)					
MON	09:00-10:00	Ph. Micro (S)	Ph. Micro (S)	MON	09:00-10:00
TUE	10:00-11:00	Ph. Micro (S)	Ph. Micro (S)	TUE	10:00-11:00
WED	11:00-12:00	Ph. Micro (S)	Ph. Micro (S)	WED	11:00-12:00
THU	12:00-1:00	Ph. Micro (S)	Ph. Micro (S)	THU	12:00-1:00
FRI	1:00-2:00	Ph. Micro (S)	Ph. Micro (S)	FRI	1:00-2:00
SAT	2:00-3:00	Ph. Micro (S)	Ph. Micro (S)	SAT	2:00-3:00
Third Semester (PCI)					
MON	09:00-10:00	Ph. Micro (S)	Ph. Micro (S)	MON	09:00-10:00
TUE	10:00-11:00	Ph. Micro (S)	Ph. Micro (S)	TUE	10:00-11:00
WED	11:00-12:00	Ph. Micro (S)	Ph. Micro (S)	WED	11:00-12:00
THU	12:00-1:00	Ph. Micro (S)	Ph. Micro (S)	THU	12:00-1:00
FRI	1:00-2:00	Ph. Micro (S)	Ph. Micro (S)	FRI	1:00-2:00
SAT	2:00-3:00	Ph. Micro (S)	Ph. Micro (S)	SAT	2:00-3:00
Fourth Semester (PCI)					
MON	09:00-10:00	Ph. Micro (S)	Ph. Micro (S)	MON	09:00-10:00
TUE	10:00-11:00	Ph. Micro (S)	Ph. Micro (S)	TUE	10:00-11:00
WED	11:00-12:00	Ph. Micro (S)	Ph. Micro (S)	WED	11:00-12:00
THU	12:00-1:00	Ph. Micro (S)	Ph. Micro (S)	THU	12:00-1:00
FRI	1:00-2:00	Ph. Micro (S)	Ph. Micro (S)	FRI	1:00-2:00
SAT	2:00-3:00	Ph. Micro (S)	Ph. Micro (S)	SAT	2:00-3:00
Fifth Semester (PCI)					
MON	09:00-10:00	Ph. Micro (S)	Ph. Micro (S)	MON	09:00-10:00
TUE	10:00-11:00	Ph. Micro (S)	Ph. Micro (S)	TUE	10:00-11:00
WED	11:00-12:00	Ph. Micro (S)	Ph. Micro (S)	WED	11:00-12:00
THU	12:00-1:00	Ph. Micro (S)	Ph. Micro (S)	THU	12:00-1:00
FRI	1:00-2:00	Ph. Micro (S)	Ph. Micro (S)	FRI	1:00-2:00
SAT	2:00-3:00	Ph. Micro (S)	Ph. Micro (S)	SAT	2:00-3:00
Sixth Semester (PCI)					
MON	09:00-10:00	Ph. Micro (S)	Ph. Micro (S)	MON	09:00-10:00
TUE	10:00-11:00	Ph. Micro (S)	Ph. Micro (S)	TUE	10:00-11:00
WED	11:00-12:00	Ph. Micro (S)	Ph. Micro (S)	WED	11:00-12:00
THU	12:00-1:00	Ph. Micro (S)	Ph. Micro (S)	THU	12:00-1:00
FRI	1:00-2:00	Ph. Micro (S)	Ph. Micro (S)	FRI	1:00-2:00
SAT	2:00-3:00	Ph. Micro (S)	Ph. Micro (S)	SAT	2:00-3:00
Seventh Semester (PCI)					
MON	09:00-10:00	Ph. Micro (S)	Ph. Micro (S)	MON	09:00-10:00
TUE	10:00-11:00	Ph. Micro (S)	Ph. Micro (S)	TUE	10:00-11:00
WED	11:00-12:00	Ph. Micro (S)	Ph. Micro (S)	WED	11:00-12:00
THU	12:00-1:00	Ph. Micro (S)	Ph. Micro (S)	THU	12:00-1:00
FRI	1:00-2:00	Ph. Micro (S)	Ph. Micro (S)	FRI	1:00-2:00
SAT	2:00-3:00	Ph. Micro (S)	Ph. Micro (S)	SAT	2:00-3:00

*(Signature)*  
Director/Principal  
Govt. College of Pharmacy Rohru  
Disstt. Shimla HP-171207



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Abbreviation:-	Subjects	Subject Code	Abbreviation:-	Subjects	Subject Code
HAP-I	Human Anatomy & Physiology-I	(BP 101T & BP 107P)	P. Eng	Pharmaceutical Engineering	(BP 304T & BP 308P)
Ph. Anal-I	Pharmaceutical Analysis	(BP 102T & BP 108P)	Med Chem-II	Medicinal Chemistry-II	(BP 501T)
PH-I	Pharmacology-I	(BP 103T & BP 110P)	Ind. Pharmacy-I	Industrial Pharmacy-I	(BP 502T & BP 506P)
Ph. C	Pharmaceutical Inorganic Chemistry	(BP 104T & BP 112P)	P. Coll-II	Pharmacology-II	(BP 503T & BP 507P)
Comm. Skills	Communication Skills	(BP 105T & BP 111P)	P. Coll-II	Pharmacognosy & Phytochemistry-II	(BP 504T & BP 508P)
R. Mathematics	Remedial Mathematics	(BP 106RMT)	PJ	Pharmaceutical Jurisprudence	(BP 505T)
R. Biology	Remedial Biology	(BP 106RBT & 112RBP)	IP-II	Instrumental method of Analysis	(BP 701T & BP 705P)
P. Org II	Pharmaceutical Organic Chemistry II	(BP 301T & BP 305P)	P. Practice	Industrial Pharmacy-II	(BP 702T)
P.P-I, N	Physical Pharmacy-I	(BP 302T & BP 306P)	NDDS	Pharmacy Practice	(BP 703T)
Ph. Micro	Pharmaceutical Microbiology	(BP 303T & BP 307P)	Practise School	Novel Drug Delivery System	(BP 704T)
NSS/NCC	National Service Scheme/ National Cadet Corps				(BP 706 PS)
			Work Load allocated		
	Faculty	Theory	Practical		Total Hrs
VS	Dr. Vivek Sharma	BP 703T	//	T	P
HK	Dr. Harish Kumar	BP 102T, BP 501T		4	0
RT	Sh. Rameer Thakur	BP 304T, BP 704T		8	16
PN	Ms. Priyanka Nagu	BP 103T, BP 502T		8	16
VM	Dr. Vinod Mehta	BP 101T, BP 505T		8	32
PS	Dr. Pankaj Sharma	BP 502T, BP 505T, BP 702T		8	24
SR	Ms. Shivani	BP 504T, BP 505T		13	8
PJ	Mr. Pankaj Jinta	BP 701T, BP 106RBT		8	24
CJ	Ms. Chetna Jhagra	BP 104T, BP 301T		6	26
MS	Sh. Mohinder Singh	BP 106 RMT		8	24
				2	0
					2

Faculty to be deputed  
Note: (\*) Faculty member to be deputed from other Institute  
Recharge Time table:  
Date: 5/8/2023

Director/Principal  
Govt. College of Pharmacy,  
Rohru, Shimla, H.P.  
Date: 5/8/2023  
Director/Principal  
Govt. College of Pharmacy,  
Rohru, Shimla, H.P.

(D) Circulation among Non-teaching Staff  
(IV) Guard File

End44:Not GPCR/Time Table/2023...1838

(D) Concerned Staff  
(III) Notice Board

Copy to:

Director/Principal  
Govt. College of Pharmacy Rohru  
Disstt. Shimla HP-171207



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Handwritten notes: 1/151, all teachers, HPTU

**GOVT. COLLEGE OF PHARMACY, ROHRU**  
**DISTT. SHIMLA, HIMACHAL PRADESH 171207**

**TIME TABLE FOR B. PHARMA. (To be effective from ...)**  
Date: 15/01/2024

No. 9

Sl. No.	Section		Semester		Class
	1	2	3	4	
1	9:30 - 10:30 AM	10:30 - 12:00 AM	2:00 - 4:00 PM	4:30 - 6:00 PM	
2	R.V.A (RD)	Maths (SB)	Pharmacology (V)	Pharmacology (V)	E
3	K.S.S (RD)	Maths (SB)	SSS/CC/Chemistry (Group A)	SSS/CC/Chemistry (Group A)	E
4	K.V.S (RD)	Maths (SB)	SSS/CC/Chemistry (Group B)	SSS/CC/Chemistry (Group B)	E
5	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
6	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
7	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
8	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
9	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
10	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
11	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
12	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
13	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
14	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
15	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
16	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
17	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
18	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
19	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
20	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
21	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
22	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
23	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
24	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
25	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
26	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
27	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
28	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
29	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
30	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
31	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
32	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
33	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
34	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
35	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
36	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
37	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
38	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
39	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
40	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
41	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
42	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
43	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
44	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
45	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
46	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
47	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
48	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
49	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
50	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
51	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
52	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
53	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
54	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
55	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
56	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
57	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
58	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
59	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
60	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
61	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
62	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
63	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
64	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
65	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
66	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
67	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
68	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
69	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
70	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
71	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
72	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
73	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
74	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
75	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
76	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
77	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
78	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
79	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
80	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
81	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
82	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
83	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
84	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
85	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
86	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
87	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
88	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
89	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
90	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
91	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
92	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
93	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
94	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
95	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
96	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
97	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
98	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
99	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E
100	Pharmacology (SB)	Maths (SB)	Ph. Org. Chem (I/II)	Ph. Org. Chem (I/II)	E

Handwritten note: 15 JAN 2024

Director/Principal  
Govt. College of Pharmacy Rohru  
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		Sixth Semester (PCI)			
		09:00 - 10:00 AM	10:00 - 11:00 AM	11:00 - 12:00 AM	12:00 - 01:00 AM
MON	Med. Chem.-III (HK)	P. Co-III (VM)	BioPharm. & Preclinical (PN)	BioPharm. & Preclinical (PN)	Herbal drug tech. (Group-A) (PR)
TUE	Med. Chem.-III (HK)	P. Co-III (VM)	BioPharm. & Preclinical (PN)	BioPharm. & Preclinical (PN)	Herbal drug tech. (Group-B) (SK)
WED	NSS/NCC/Revision Class	P. Co-III (VM)	BioPharm. & Preclinical (PN)	BioPharm. & Preclinical (PN)	NSS/NCC/Revision Class (Spoken material)
THU	HDT (SR)	NSS/NCC/Revision Class	BioPharm. & Preclinical (PN)	BioPharm. & Preclinical (PN)	NSS/NCC/Revision Class (Spoken material)
FRI	Med. Chem.-III (HK)	HDT (SR)	QA (RT)	QA (RT)	Med. Chem. III (Group-A) (HK) / Pharmacology - II (Group-B) (PN)
SAT	Med. Chem.-III (HK)	P. Co-III (VM)	QA (RT)	QA (RT)	Med. Chem. III (Group-B) (CA) / Pharmacology - II (Group-A) (PN)
		L			
		U			
		N			
		C			
		H			
		Eighth Semester (PCI)			
		09:00 - 10:00 AM	10:00 - 11:00 AM	11:00 - 12:00 AM	12:00 - 01:00 AM
MON	Advanced D.M.A. (PD)	Pharm. Rep. Sciences (PS)	Pharm. Rep. Sciences (PS)	Project work	Biostat. & Research Methodology (MS)
TUE	Social and Preventive Pharmacy (PN)	Pharm. Rep. Sciences (PS)	Pharm. Rep. Sciences (PS)	Project work	Biostat. & Research Methodology (MS)
WED	Social and Preventive Pharmacy (PN)	Pharm. Rep. Sciences (PS)	Pharm. Rep. Sciences (PS)	Project work	Biostat. & Research Methodology (MS)
THU	Social and Preventive Pharmacy (PN)	Pharm. Rep. Sciences (PS)	Pharm. Rep. Sciences (PS)	Project work	Biostat. & Research Methodology (MS)
FRI	Pharm. Rep. Sciences (PS)	Pharm. Rep. Sciences (PS)	Pharm. Rep. Sciences (PS)	Project work	Biostat. & Research Methodology (MS)
SAT	Pharm. Rep. Sciences (PS)	Pharm. Rep. Sciences (PS)	Pharm. Rep. Sciences (PS)	Project work	Biostat. & Research Methodology (MS)
		L			
		U			
		N			
		C			
		H			

Note: The practicals of evening session will be conducted from 02:00 pm to 06:00 pm

Page-2/3

15 Jan 2021

Director/Principal  
Govt. College of Pharmacy Rohru  
Disstt. Shimla HP-171207



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Abbreviation:-	Subjects	Subject Code	Abbreviation:-	Subjects	Subject Code
HAP-II	Human Anatomy and Physiology II	BP201T & BP207P	Med. Chem. III	Medical Chemistry III	BP-601Y/BP-607P
P.Organic Chem-I	Pharmaceutical Organic Chemistry I	BP202T & BP208P	P. Col-III	Pharmacology III	BP-602T/BP603P
Biotech.	Biochemistry	BP203T & BP209P	DDT	Herbal Drug Technology	BP-603T/BP-609P
Pathophysiology	Pathophysiology	BP204T	Biopharm. & Probiotics	Biopharmaceutics and Pharmacokinetics	BP-604T
Computer Applications	Computer Applications in Pharmacy	BP205T & BP210P	Biotech.	Pharmaceutical Biotechnology	BP-605T
E.V.S	Environmental sciences	BP206T	QA	Quality Assurance	BP-606T
P.Organic Chem-II	Pharmaceutical Organic-Chemistry III	BP401T	Biostat. & Research Methodology	Bio-statistics and Research Methodology	BP-607T
Med. Chem. I	Medicinal Chemistry-I	BP402T/BP-406P	Social and Preventive Pharmacy	Social and Preventive Pharmacy	BP-608T
Phy. Pharmacy	Physical Pharmaceutics II	BP403T/BP-407P	Exp-Pharmacology	Experimental Pharmacology	BP410ET
P. Col-I	Pharmacology I	BP404T/BP-408P	Advanced L.M.A.	Advanced instrumental analysis	BP411ET
P.Cog. & Physchem-I	Pharmacology and Physicochemistry I	BP405T/BP-409P	Pharm. Reg. Sciences	Pharmaceutical Regulatory Sciences	BP404 ET

Work Load allocated		Theory		Practical		Total Hrs	
Faculty		T	P	T	P	T	P
VS	Dr. Vivek Sharma	BP204T, BP404T, BP810ET	0	12	0	12	12
HK	Dr. Harish Kumar	BP402T, BP-601T	8	8	12	20	20
RT	Sh. Ranveer Thakur	BP-606T, BP206T	7 + 2	8	8	17	17
PN	Dr. Priyanka Nagu	BP-604T, BP-802T	8 + 2	8	8	18	18
VM	Dr. Vineet Mehta	BP201T, BP-602T	8	8	8	16	16
PS	Dr. Parbaj Sharma	BP403T, BP-605T, BP804 ET	12	8	8	20	20
SR	Ms. Shikha	BP-603T, BP203T	8	8	16	24	24
FJ	Mr. Pankaj Jaita	BP405T, BP411ET	8	8	8	16	16
CJ	Ms. Chetana Jhagta	BP202T, BP401T	8	8	12	20	20
MS	Sh. Mohinder Singh	BP-801T		8	12		

\* Faculty to be deputed  
Note: (\*) Faculty member to be deputed from other institute

Incharge Time table:

Endst.No: GPCR/Time Table 2024.../49

Copy to:

(I) Concerned Staff  
(II) Notice Board

(ID) Circulation among Non-teaching Staff  
(IV) Guard File

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15 JAN 2024

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## Semester wise Syllabus and Course Mapping (Session 2023-2024)

  
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## 1<sup>st</sup> Year (Semester I) PCI Syllabus

  
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**BP 101 T: Human Anatomy and Physiology-I (Theory)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours


**Scope:** This subject is designed to impart fundamental knowledge on the structure and functions of the various systems of the human body. It also helps in understanding both homeostatic mechanisms. The subject provides the basic knowledge required to understand the various disciplines of pharmacy.

**Objectives:** Upon completion of this course the student should be able to

- Explain the gross morphology, structure and functions of various organs of the human body.
- Describe the various homeostatic mechanisms and their imbalances.
- Identify the various tissues and organs of different systems of human body.
- Perform the various experiments related to special senses and nervous system.
- Appreciate coordinated working pattern of different organs of each system

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<p><b>Introduction to human body:</b> Definition and scope of anatomy and physiology, levels of structural organization and body systems, basic life processes, homeostasis, basic anatomical terminology.</p> <p><b>Cellular level of organization:</b> Structure and functions of cell, transport across cell membrane, cell division, cell junctions. General principles of cell communication, intracellular signaling pathway activation by extracellular signal molecule, Forms of intracellular signaling: a) Contact-dependent b) Paracrine c) Synaptic d) Endocrine.</p> <p><b>Tissue level of organization:</b> Classification of tissues, structure, location and</p>	10

  
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	functions of epithelial, muscular and nervous and connective tissues.	
II	<b>Integumentary system:</b> Structure and functions of skin.  <b>Skeletal system:</b> Divisions of skeletal system, types of bone, salient features and functions of bones of axial and appendicular skeletal system Organization of skeletal muscle, physiology of muscle contraction, neuromuscular junction.  <b>Joints:</b> Structural and functional classification, types of joints movements and its articulation	10
III	<b>Body fluids and blood:</b> Body fluids, composition and functions of blood, hemopoiesis, formation of hemoglobin, anemia, mechanisms of coagulation, blood grouping, Rh factors, transfusion, its significance and disorders of blood, Reticulo endothelial system.  <b>Lymphatic system:</b> Lymphatic organs and tissues, lymphatic vessels, lymph circulation and functions of lymphatic system	10
IV	<b>Peripheral nervous system:</b> Classification of peripheral nervous system: Structure and functions of sympathetic and parasympathetic nervous system. Origin and functions of spinal and cranial nerves.  <b>Special senses:</b> Structure and functions of eye, ear, nose and tongue and their disorders.	08
V	<b>Cardiovascular system:</b> Heart – anatomy of heart, blood circulation, blood vessels, structure and functions of artery, vein and capillaries, elements of conduction system of heart and heart beat, its regulation by autonomic nervous system, cardiac output, cardiac cycle. Regulation of blood pressure, pulse, electrocardiogram and disorders of heart.	07

## Recommended Books

1. Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypee brothers medical publishers, New Delhi.
2. Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York.

  
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3. Physiological basis of Medical Practice-Best and Taylor. Williams & Wilkins Co, Riverview, MI USA.
4. Textbook of Medical Physiology- Arthur C, Guyton and John.E. Hall. Miamisburg, OH, U.S.A.
5. Principles of Anatomy and Physiology by Tortora Grabowski. Palmetto, GA, U.S.A. 32
6. Textbook of Human Histology by Inderbir Singh, Jaypee brother's medical publishers, New Delhi.
7. Textbook of Practical Physiology by C.L. Ghai, Jaypee brother's medical publishers, New Delhi.
8. Physiological basis of Medical Practice-Best and Taylor. Williams & Wilkins Co Riverview, MI USA
9. Text book of Medical Physiology- Arthur C, Guyton and John. E. Hall. Miamisburg, OH, U.S.A.
10. Human Physiology (vol 1 and 2) by Dr. C.C. Chatterrje, Academic Publishers

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

### Course Outcomes:

On completion of this course, the students are expected to:

CO1	To memorize the anatomy, terminology, and gross morphology of various organs of the human body.
CO2	To understand the physiology of various organ systems of the body and their interlinked mechanisms
CO3	To analyze and apply the knowledge of anatomy and physiology in the body system to various disorders

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	2	-	-	-	1	2	-	3	2	3	-
CO2	3	-	3	2	-	-	-	1	2	-	3	2	3	-
CO3	3	2	3	1	-	-	-	-	2	-	3	1	3	-

  
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**BP102T: PHARMACEUTICAL ANALYSIS (Theory)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This course deals with the fundamentals of analytical chemistry and principles of electrochemical analysis of drugs

**Objectives:** Upon completion of the course student shall be able to

- understand the principles of volumetric and electro chemical analysis
- carryout various volumetric and electrochemical titrations
- develop analytical skills

#### COURSE CONTENT

UNIT	CONTENT	No. ofHrs.
I	(a) <b>Pharmaceutical analysis-</b> Definition and scope i) Different techniques of analysis ii) Methods of expressing concentration iii) Primary and secondary standards. iv) Preparation and standardization of various molar and normal solutions- Oxalic acid, sodium hydroxide, hydrochloric acid, sodium thiosulphate, sulphuric acid, potassium permanganate and ceric ammonium sulphate (b) <b>Errors:</b> Sources of errors, types of errors, methods of minimizing errors, accuracy, precision and significant figures (c) Pharmacopoeia, Sources of impurities in medicinal agents, limit tests.	10
II	<b>Acid base titration:</b> Theories of acid base indicators, classification of acid base titrations and theory involved in titrations of strong, weak, and very weak acids and bases, neutralization curves <b>Non aqueous titration:</b> Solvents, acidimetry and alkalimetry titration and estimation of Sodium benzoate and Ephedrine HCl	10

  
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III	<b>Precipitation titrations:</b> Mohr's method, Volhard's, Modified Volhard's, Fajans method, estimation of sodium chloride. <b>Complexometric titration:</b> Classification, metal ion indicators, masking and demasking reagents, estimation of Magnesium sulphate, and calcium gluconate. <b>Gravimetry:</b> Principle and steps involved in gravimetric analysis. Purity of the precipitate: co-precipitation and post precipitation, Estimation of barium sulphate. Basic Principles, methods and application of diazotisation titration.	10
IV	<b>Redox titrations</b> (a) Concepts of oxidation and reduction (b) Types of redox titrations (Principles and applications) Cerimetry, Iodimetry, Iodometry, Bromatometry, Dichrometry, Titration with potassium iodate	08
V	<b>Electrochemical methods of analysis</b> <b>Conductometry-</b> Introduction, Conductivity cell, Conductometric titrations, applications. <b>Potentiometry</b> - Electrochemical cell, construction and working of reference (Standard hydrogen, silver chloride electrode and calomel electrode) and indicator electrodes (metal electrodes and glass electrode), methods to determine end point of potentiometric titration and applications. <b>Polarography</b> - Principle, Ilkovic equation, construction and working of dropping mercury electrode and rotating platinum electrode, applications	07

**Recommended Books: (Latest Editions)**

1. A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London
2. A.I. Vogel, Text Book of Quantitative Inorganic analysis
3. P. Gundu Rao, Inorganic Pharmaceutical Chemistry
4. Bentley and Driver's Textbook of Pharmaceutical Chemistry
5. John H. Kennedy, Analytical chemistry principles
6. Indian Pharmacopoeia.

  
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### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To describe and understand the fundamentals of analytical chemistry.
CO2	To classify and execute various methods to perform various methods in drug analysis.
CO3	To attain basic learning and expertise in Pharmaceutical analysis for use in Pharmacy, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	1	2	-	3	2	-	3
CO2	3	-	2	2	-	-	-	1	2	-	3	2	-	3
CO3	3	2	3	3	-	-	1	-	2	1	3	1	-	3

  
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### BP 103 T: PHARMACEUTICS- I (Theory)

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This course is designed to impart a fundamental knowledge on the preparatory pharmacy with arts and science of preparing the different conventional dosage forms.

**Objectives:** Upon completion of this course the student should be able to:

- Know the history of profession of pharmacy
- Understand the basics of different dosage forms, pharmaceutical incompatibilities and pharmaceutical calculations
- Understand the professional way of handling the prescription
- Preparation of various conventional dosage forms

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Historical background and development of profession of pharmacy:</b> History of profession of Pharmacy in India in relation to pharmacy education, industry and organization, Pharmacy as a career, Pharmacopoeias: Introduction to IP, BP, USP and Extra Pharmacopoeia. <b>Dosage forms:</b> Introduction to dosage forms, classification and definitions <b>Prescription:</b> Definition, Parts of prescription, handling of Prescription and Errors in prescription. <b>Posology:</b> Definition, Factors affecting posology. Pediatric dose calculations based on age, body weight and body surface area.	10
II	<b>Pharmaceutical calculations:</b> Weights and measures – Imperial & Metric system, Calculations involving percentage solutions, alligation, proof spirit and isotonic solutions based on freezing point and molecular weight. <b>Powders:</b> Definition, classification, advantages and disadvantages, Simple & compound powders – official preparations, dusting powders, effervescent,	10

  
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	efflorescent and hygroscopic powders, eutectic mixtures, Geometric dilutions. <b>Liquid dosage forms:</b> Advantages and disadvantages of liquid dosage forms. Excipients used in formulation of liquid dosage forms. Solubility enhancement techniques	
III	<b>Monophasic liquids:</b> Definitions and preparations of Gargles, Mouthwashes, Throat Paint, Eardrops, Nasal drops, Enemas, Syrups, Elixirs, Liniments and Lotions. <b>Biphasic liquids: Suspensions:</b> Definition, advantages and disadvantages, classifications, Preparation of suspensions; Flocculated and Deflocculated suspension & stability problems and methods to overcome. <b>Emulsions:</b> Definition, classification, emulsifying agent, test for the identification of type of Emulsion, Methods of preparation & stability problems and methods to overcome.	08
IV	<b>Suppositories:</b> Definition, types, advantages and disadvantages, types of bases, methods of preparations. Displacement value & its calculations, evaluation of suppositories. <b>Pharmaceutical incompatibilities:</b> Definition, classification, physical, chemical and therapeutic incompatibilities with examples.	08
V	<b>Semisolid dosage forms:</b> Definitions, classification, mechanisms and factors influencing dermal penetration of drugs. Preparation of ointments, pastes, creams and gels. Excipients used in semi solid dosage forms. Evaluation of semi solid dosages forms	07

### Recommended Books:

1. H.C. Ansel et al., Pharmaceutical Dosage Form and Drug Delivery System, Lippincott Williams and Walkins, New Delhi.
2. Carter S.J., Cooper and Gunn's-Dispensing for Pharmaceutical Students, CBS publishers, New Delhi.
3. M.E. Aulton, Pharmaceutics, The Science & Dosage Form Design, Churchill Livingstone, Edinburgh.
4. Indian pharmacopoeia.
5. British pharmacopoeia.

  
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6. Lachmann. Theory and Practice of Industrial Pharmacy, Lea & Febiger Publisher, TheUniversity of Michigan.
7. Alfonso R. Gennaro Remington. The Science and Practice of Pharmacy, LippincottWilliams, New Delhi.
8. Carter S.J., Cooper and Gunn's. Tutorial Pharmacy, CBS Publications, New Delhi.
9. E.A. Rawlins, Bentley's Text Book of Pharmaceutics, English Language Book Society, Elsevier Health Sciences, USA.
10. Isaac Ghebre Sellassie: Pharmaceutical Pelletization Technology, Marcel Dekker, INC, New York.
11. Dilip M. Parikh: Handbook of Pharmaceutical Granulation Technology, Marcel Dekker, INC, New York.
12. Francoise Nieloud and Gilberte Marti-Mestres: Pharmaceutical Emulsions and Suspensions, Marcel Dekker, INC, New York.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand basic principles and calculations involved in the conduct of experiments to prepare different dosage forms and prepare, dispense and evaluate conventional dosage forms for the benefit of society.
CO2	To interpret prescription, drug interactions, and adverse effects associated with active pharmaceutical ingredients and excipients. Evaluate possible incompatibilities in dosage formulations and methods to overcome

  
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CO3	To apply knowledge of posology to calculate the dose of pediatrics and other population
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### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	1	2	-	3	2	3	3
CO2	3	-	2	2	-	-	-	1	2	-	3	2	3	3
CO3	3	2	3	3	-	1	1	-	2	-	3	1	3	3

  
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## BP 104 T: PHARMACEUTICAL INORGANIC CHEMISTRY (Theory)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject deals with the monographs of inorganic drugs and pharmaceuticals.


**Objectives:** Upon completion of course student shall be able to

- know the sources of impurities and methods to determine the impurities in inorganic drugs and pharmaceuticals
- understand the medicinal and pharmaceutical importance of inorganic compounds

### COURSE CONTENT

**General methods of preparation**, assay for the compounds superscripted with asterisk (\*), properties and medicinal uses of inorganic compounds belonging to the following classes

UNIT	CONTENT	No. of Hrs.
I	<b>Impurities in pharmaceutical substances:</b> History of Pharmacopoeia, Sources and types of impurities, principle involved in the limit test for Chloride, Sulphate, Iron, Arsenic, Lead and Heavy metals, modified limit test for Chloride and Sulphate	10
II	<b>Acids, Bases and Buffers:</b> Buffer equations and buffer capacity in general, buffers in pharmaceutical systems, preparation, stability, buffered isotonic solutions, measurements of tonicity, calculations and methods of adjusting isotonicity. <b>Major extra and intracellular electrolytes:</b> Functions of major physiological ions, Electrolytes used in the replacement therapy: Sodium chloride*, Potassium chloride, Calcium gluconate* and Oral Rehydration Salt (ORS), Physiological acid base balance. <b>Dental products:</b> Dentifrices, role of fluoride in the treatment of dental caries, Desensitizing agents, Calcium carbonate, Sodium fluoride, and Zinc eugenol cement.	10

  
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III	<b>Gastrointestinal agents</b> <b>Acidifiers:</b> Ammonium chloride* and Dil. HCl <b>Antacid:</b> Ideal properties of antacids, combinations of antacids, Sodium Bicarbonate*, Aluminum hydroxide gel, Magnesium hydroxide mixture <b>Cathartics:</b> Magnesium sulphate, Sodium orthophosphate, Kaolin and Bentonite <b>Antimicrobials:</b> Mechanism, classification, Potassium permanganate, Boric acid, Hydrogen peroxide*, Chlorinated lime*, Iodine and its preparations	10
IV	<b>General methods of preparation</b> , assay for the compounds superscripted with asterisk (*), properties and medicinal uses of inorganic compounds belonging to the following classes <b>Miscellaneous compounds</b> <b>Expectorants:</b> Potassium iodide, Ammonium chloride*. <b>Emetics:</b> Copper sulphate*, Sodium potassium tartarate <b>Haematinics:</b> Ferrous sulphate*, Ferrous gluconate <b>Poison and Antidote:</b> Sodium thiosulphate*, Activated charcoal, Sodium nitrite <b>Astringents:</b> Zinc Sulphate, Potash Alum	08
V	<b>Radiopharmaceuticals:</b> Radio activity, Measurement of radioactivity, Properties of $\alpha$ , $\beta$ , $\gamma$ radiations, Half life, radio isotopes and study of radio isotopes - Sodium iodide I131, Storage conditions, precautions & pharmaceutical application of radioactive substances.	07

### Recommended Books:

1. A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London, 4th edition.
2. A.I. Vogel, Text Book of Quantitative Inorganic analysis

  
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3. P. Gundu Rao, Inorganic Pharmaceutical Chemistry, 3rd Edition
4. M.L Schroff, Inorganic Pharmaceutical Chemistry
5. Bentley and Driver's Textbook of Pharmaceutical Chemistry
6. Anand & Chatwal, Inorganic Pharmaceutical Chemistry
7. Indian Pharmacopoeia

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To define impurities and their types in pharmaceutical preparations. To understand the methods for determination of impurities in inorganic drugs and pharmaceuticals
CO2	To classify inorganic pharmaceuticals on the basis of uses and properties. Understand the limit test and its importance in pharmaceuticals
CO3	To attain expertise in concentration calculations of a solution, its preparation, standardization, and its storage conditions

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	-	-	-	-	1	2	1	3	1	-	3
CO2	3	-	2	-	-	-	-	1	2	1	3	1	-	3
CO3	3	2	3	2	-	-	-	-	2	1	3	1	-	3

  
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**BP 105 T: COMMUNICATION SKILLS (Theory)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
2	0	0	2	15	35	50	1.5 hours

**Scope:** This course will prepare the young pharmacy student to interact effectively with doctors, nurses, dentists, physiotherapists and other health workers. At the end of this course the student will get the soft skills set to work cohesively with the team as a team player and will add value to the pharmaceutical business.

#### Objectives:

Upon completion of the course the student shall be able to

- Understand the behavioral needs for a Pharmacist to function effectively in the areas of pharmaceutical operation
- Communicate effectively (Verbal and Non Verbal)
- Effectively manage the team as a team player
- Develop interview skills
- Develop Leadership qualities and essentials

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Communication Skills:</b> Introduction, Definition, The Importance of Communication, The Communication Process – Source, Message, Encoding, Channel, Decoding, Receiver, Feedback, Context <b>Barriers to communication:</b> Physiological Barriers, Physical Barriers, Cultural Barriers, Language Barriers, Gender Barriers, Interpersonal Barriers, Psychological Barriers, Emotional barriers <b>Perspectives in Communication:</b> Introduction, Visual Perception, Language, Other factors affecting our perspective - Past Experiences, Prejudices, Feelings, Environment	07
II	<b>Elements of Communication:</b> Introduction, Face to Face Communication - Tone	07

  
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	of Voice, Body Language (Non-verbal communication), Verbal Communication, Physical Communication <b>Communication Styles:</b> Introduction, The Communication Styles Matrix with example for each -Direct Communication Style, Spirited Communication Style, Systematic Communication Style, Considerate Communication Style	
III	<b>Basic Listening Skills:</b> Introduction, Self-Awareness, Active Listening, Becoming an Active Listener, Listening in Difficult Situations <b>Effective Written Communication:</b> Introduction, When and When Not to Use Written Communication - Complexity of the Topic, Amount of Discussion Required, Shades of Meaning, Formal Communication <b>Writing Effectively:</b> Subject Lines, Put the Main Point First, Know Your Audience, Organization of the Message	07
IV	<b>Interview Skills:</b> Purpose of an interview, Do's and Dont's of an interview <b>Giving Presentations:</b> Dealing with Fears, Planning your Presentation, Structuring Your Presentation, Delivering Your Presentation, Techniques of Delivery	05
V	<b>Group Discussion:</b> Introduction, Communication skills in group discussion, Do's and Dont's of group discussion	04

### Recommended Books:

1. Basic communication skills for Technology, Andreja. J. Ruther Ford, 2nd Edition, Pearson Education, 2011
2. Communication skills, Sanjay Kumar, Pushpalata, 1st Edition, Oxford Press, 2011
3. Organizational Behaviour, Stephen .P. Robbins, 1st Edition, Pearson, 2013
4. Brilliant- Communication skills, Gill Hasson, 1st Edition, Pearson Life, 2011
5. The Ace of Soft Skills: Attitude, Communication and Etiquette for success, Gopala Swamy Ramesh, 5th Edition, Pearson, 2013
6. Developing your influencing skills, Deborah Dalley, Lois Burton, Margaret, Greenhall, 1<sup>st</sup> Edition Universe of Learning LTD, 2010
7. Communication skills for professionals, Konar nira, 2nd Edition, New arrivals –PHI, 2011
8. Personality development and soft skills, Barun K Mitra, 1st Edition, Oxford Press, 2011

  
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9. Soft skill for everyone, Butter Field, 1st Edition, Cengage Learning india pvt. ltd, 2011
10. Soft skills and professional communication, Francis Peters SJ, 1st Edition, Mc Graw Hill Education, 2011
11. Effective communication, John Adair, 4th Edition, Pan Mac Millan, 2009
12. Bringing out the best in people, Aubrey Daniels, 2nd Edition, Mc Graw Hill, 1999

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	05	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To become proficient in reading, writing, listening, and oral skills.
CO2	To learn basic concepts of debate, discussion, etc., and their implementation.
CO3	To attain good leadership quality with overall personality development.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	-	-	-	-	3	2	-	3	-	-	3	-	-	2
CO2	-	-	-	-	3	2	-	3	-	-	3	-	-	2
CO3	-	-	-	-	3	2	-	3	-	-	3	-	-	2

  
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**BP 106 RBT. REMEDIAL BIOLOGY (Theory)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
2	0	0	2	15	35	50	1.5 hours

**Scope:** To learn and understand the components of living world, structure and functional system of plant and animal kingdom.

**Objectives:** Upon completion of the course, the student shall be able to

- know the classification and salient features of five kingdoms of life
- understand the basic components of anatomy & physiology of plant
- know understand the basic components of anatomy & physiology animal with special reference to human

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Living world:</b> Definition and characters of living organisms; Diversity in the living world; Binomial nomenclature ; Five kingdoms of life and basis of classification. Salient features of Monera, Protista, Fungi, Animalia and Plantae, Virus. <b>Morphology of Flowering plants:</b> Morphology of different parts of flowering plants – Root, stem, inflorescence, flower, leaf, fruit, seed. General Anatomy of Root, stem, leaf of monocotyledons & Dicotyledones.	07
II	<b>Body fluids and circulation ;</b> Composition of blood, blood groups, coagulation of blood; Composition and functions of lymph; Human circulatory system; Structure of human heart and blood vessels; Cardiac cycle, cardiac output and ECG <b>Digestion and Absorption :</b> Human alimentary canal and digestive glands; Role of digestive enzymes; Digestion, absorption and assimilation of digested food <b>Breathing and respiration:</b> Human respiratory system; Mechanism of breathing and its regulation; Exchange of gases, transport of gases and regulation of	07

  
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	respiration ; Respiratory volumes	
III	<b>Excretory products and their elimination:</b> Modes of excretion; Human excretory system- structure and function; Urine formation; Rennin angiotensin system <b>Neural control and coordination:</b> Definition and classification of nervous system; Structure of a neuron; Generation and conduction of nerve impulse; Structure of brain and spinal cord; Functions of cerebrum, cerebellum, hypothalamus and medulla oblongata. <b>Chemical coordination and regulation:</b> Endocrine glands and their secretions; Functions of hormones secreted by endocrine glands <b>Human reproduction:</b> Parts of female reproductive system; Parts of male reproductive system; Spermatogenesis and Oogenesis; Menstrual cycle	07
IV	<b>Plants and mineral nutrition:</b> Essential mineral, macro and micronutrients; Nitrogen metabolism, Nitrogen cycle, biological nitrogen fixation <b>Photosynthesis:</b> Autotrophic nutrition, photosynthesis, Photosynthetic pigments, Factors affecting photosynthesis.	05
V	<b>Plant respiration:</b> Respiration, glycolysis, fermentation (anaerobic). <b>Plant growth and development:</b> Phases and rate of plant growth, Condition of growth,Introduction to plant growth regulators <b>Cell - The unit of life ;</b> Structure and functions of cell and cell organelles.Cell division. <b>Tissues :</b> Definition, types of tissues, location and functions.	04

### Recommended Books:

1. Text book of Biology by S. B. Gokhale
2. A Text book of Biology by Dr. Thulajappa and Dr. Seetaram
3. A Text book of Biology by B.V. Sreenivasa Naidu
4. A Text book of Biology by Naidu and Murthy
5. Botany for Degree students By A.C.Dutta.
6. Outlines of Zoology by M. Ekambaranatha Ayyer and T. N. Ananthkrishnan.
7. A manual for pharmaceutical biology practical by S.B. Gokhale and C. K. Kokate

  
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8. Practical human anatomy and physiology. by S.R. Kale and R.R. Kale.
9. A Manual of pharmaceutical biology practical by S.B. Gokhale, C.K. Kokate and S.P. Shriwastava.
10. Biology practical manual according to National core curriculum. Biology forum of Karnataka.  
Prof. M.J.H. Shafi

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	05	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To memorize and understand the physiology of the human body and plants.
CO2	To identify plants and animals based on morphological characters.
CO3	To apply the attained knowledge professionally in Pharmacy, Hospitals, and Research.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	2	-	-	-	-	-	-	-	1	-	3	2	-	2
CO2	2	-	-	-	-	-	-	-	1	-	3	2	-	2
CO3	2	-	-	-	-	-	-	-	1	-	3	2	-	2

  
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**BP 106 RMT. REMEDIAL MATHEMATICS (Theory)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
2	0	0	2	15	35	50	1.5 Hours

**Scope:** This is an introductory course in mathematics. This subject deals with the introduction to Partial fraction, Logarithm, matrices and Determinant, Analytical geometry, Calculus, differential equation and Laplace transform.

**Objectives:** Upon completion of the course the student shall be able to:-

- Know the theory and their application in Pharmacy
- Solve the different types of problems by applying theory
- Appreciate the important application of mathematics in Pharmacy

**COURSE CONTENT**

UNIT	CONTENT	No. of Hrs.
I	<p><b>Partial fraction</b> Introduction, Polynomial, Rational fractions, Proper and Improper fractions, Partial fraction, Resolving into Partial fraction, Application of Partial Fraction in Chemical Kinetics and Pharmacokinetics</p> <p><b>Logarithms</b> Introduction, Definition, Theorems/Properties of logarithms, Common logarithms, Characteristic and Mantissa, worked examples, application of logarithm to solve pharmaceutical problems.</p> <p><b>Function:</b> Real Valued function, Classification of real valued functions,</p> <p><b>Limits and continuity</b> Introduction, Limit of a function, Definition of limit of a function (<math>\epsilon - \delta</math> definition), <math>\lim_{x \rightarrow a} \frac{x^n - a^n}{x - a} = na^{n-1}</math>, <math>\lim_{\theta \rightarrow 0} \frac{\sin \theta}{\theta} = 1</math>.</p>	06
II	<p><b>Matrices and Determinant:</b> Introduction matrices, Types of matrices, Operation on matrices, Transpose of a matrix, Matrix Multiplication, Determinants, Properties of determinants, Product</p>	06

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	of determinants, Minors and co-Factors, Adjointor adjugate of a square matrix , Singular and non-singular matrices, Inverse of a matrix, Solution of system of linear of equations using matrix method, Cramer's rule, Characteristic equation and roots of a square matrix, Cayley–Hamilton theorem, Application of Matrices in solving Pharmacokinetic equations.	
III	<p><b>Calculus</b></p> <p><b>Differentiation</b> : Introductions, Derivative of a function, Derivative of a constant, Derivative of a product of a constant and a function , Derivative of the sum or difference of two functions, Derivative of the product of two functions (product formula), Derivative of the quotient of two functions (Quotient formula) – <b>Without Proof</b>, Derivative of <math>x^n</math> w.r.t. <math>x</math>, where <math>n</math> is any rational number, Derivative of <math>e^x</math>, Derivative of <math>\log_e x</math> , Derivative of <math>a^x</math>, Derivative of trigonometric functions from first principles (<b>without Proof</b>), Successive Differentiation, Conditions for a function to be a maximum or a minimum at a point. Application</p>	06
IV	<p><b>Analytical Geometry</b></p> <p><b>Introduction:</b> Signs of the Coordinates, Distance formula, <b>Straight Line</b> : Slope or gradient of a straight line, Conditions for parallelism and perpendicularity of two lines, Slope of a line joining two points, Slope – intercept form of a straight line. <b>Integration:</b> Introduction, Definition, Standard formulae, Rules of integration , Method of substitution, Method of Partial fractions, Integration by parts, definite integrals, application</p>	06
V	<p><b>Differential Equations</b> : Some basic definitions, Order and degree, Equations in separable form, Homogeneous equations, Linear Differential equations, Exact equations, Application in solving Pharmacokinetic equations</p> <p><b>Laplace Transform</b> : Introduction, Definition, Properties of Laplace transform, Laplace Transforms of elementary functions, Inverse Laplace transforms, Laplace transform of derivatives, Application to solve Linear differential equations, Application in solving Chemical kinetics and Pharmacokinetics equations</p>	06

### Recommended Books (Latest Edition)

1. Differential Calculus by Shanthinarayan
2. Pharmaceutical Mathematics with application to Pharmacy by Panchaksharappa Gowda D.H.

  
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3. Integral Calculus by Shanthinarayan
4. Higher Engineering Mathematics by Dr. B.S. Grewal

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:


Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	05	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To memorize the fundamentals of introductory mathematics.
CO2	To understand the concepts of analytical geometry and calculus
CO3	To calculate partial fraction, logarithm, matrices, and determinant.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	-	-	1	1	-	-	-	-	-	-	2	-	-	2
CO2	-	-	1	1	-	-	-	-	-	-	2	-	-	2
CO3	-	-	1	1	-	-	-	-	-	-	2	-	-	2

  
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**BP 107 P: HUMAN ANATOMY AND PHYSIOLOGY (Practical)**

#### Teaching and Examination Scheme:


Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

Practical physiology is complimentary to the theoretical discussions in physiology. Practicals allow the verification of physiological processes discussed in theory classes through experiments on living tissue, intact animals or normal human beings. This is helpful for developing an insight on the subject.

1. Study of compound microscope.
2. Microscopic study of epithelial and connective tissue
3. Microscopic study of muscular and nervous tissue
4. Identification of axial bones
5. Identification of appendicular bones
6. Introduction to hemocytometry.
7. Enumeration of white blood cell (WBC) count
8. Enumeration of total red blood corpuscles (RBC) count
9. Determination of bleeding time
10. Determination of clotting time
11. Estimation of hemoglobin content
12. Determination of blood group.
13. Determination of erythrocyte sedimentation rate (ESR).
14. Determination of heart rate and pulse rate.
15. Recording of blood pressure.

#### Recommended Books

1. Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypee brothers medical publishers, New Delhi.
2. Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York

  
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3. Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co, Riverview, MI USA
4. Textbook of Medical Physiology- Arthur C, Guyton and John.E. Hall. Miamisburg, OH, U.S.A.
5. Principles of Anatomy and Physiology by Tortora Grabowski. Palmetto, GA, U.S.A. 32
6. Textbook of Human Histology by Inderbir Singh, Jaypee brother's medical publishers, New Delhi.
7. Textbook of Practical Physiology by C.L. Ghai, Jaypee brother's medical publishers, New Delhi.
8. Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co Riverview, MI USA
9. Text book of Medical Physiology- Arthur C, Guyton and John. E. Hall. Miamisburg, OH, U.S.A.
10. Human Physiology (vol 1 and 2) by Dr. C.C. Chatterrje, Academic Publishers.


### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To memorize the anatomy, terminology, and gross morphology of various organs of the human body.
CO2	To understand the physiology of various organ systems of the body and their interlinked mechanisms
CO3	To analyze and apply the knowledge of anatomy and physiology in the body system to various disorders

  
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## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	2	-	-	-	1	2	-	3	2	3	-
CO2	3	-	3	2	-	-	-	1	2	-	3	2	3	-
CO3	3	2	3	1	-	-	-	-	2	-	3	1	3	-

  
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**BP 108 P: PHARMACEUTICAL ANALYSIS (Practical)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	4 hours

**I Limit Test of the following**

- (1) Chloride(2) Sulphate(3) Iron(4) Arsenic

**II Preparation and standardization of**

- (1) Sodium hydroxide(2) Sulphuric acid  
(3) Sodium thiosulfate(4) Potassium permanganate  
(5) Ceric ammonium sulphate

**III Assay of the following compounds along with Standardization of Titrant**

- (1) Ammonium chloride by acid base titration  
(2) Ferrous sulphate by Cerimetry  
(3) Copper sulphate by Iodometry  
(4) Calcium gluconate by complexometry  
(5) Hydrogen peroxide by Permanganometry  
(6) Sodium benzoate by non-aqueous titration  
(7) Sodium Chloride by precipitation titration

**IV Determination of Normality by electro-analytical methods**

- (1) Conductometric titration of strong acid against strong base  
(2) Conductometric titration of strong acid and weak acid against strong base  
(3) Potentiometric titration of strong acid against strong base

  
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### Recommended Books: (Latest Editions)

1. A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London
2. A.I. Vogel, Text Book of Quantitative Inorganic analysis
3. P. Gundu Rao, Inorganic Pharmaceutical Chemistry
4. Bentley and Driver's Textbook of Pharmaceutical Chemistry
5. John H. Kennedy, Analytical chemistry principles
6. Indian Pharmacopoeia.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To describe and understand the fundamentals of analytical chemistry.
CO2	To classify and execute various methods to perform various methods in drug analysis.
CO3	To analyze and apply the knowledge of pharmaceutical analysis.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	1	2	-	3	2	-	3
CO2	3	-	2	2	-	-	-	1	2	-	3	2	-	3
CO3	3	2	3	3	-	-	1	-	2	1	3	1	-	3

  
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**BP 109 P: PHARMACEUTICS-I (Practical)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	4 hours

- 1. Syrups:** a) Syrup IP'66  
b) Compound syrup of Ferrous Phosphate BPC'68
- 2. Elixirs:** a) Piperazine citrate elixir  
b) Paracetamol pediatric elixir
- 3. Linctus** a) Terpin Hydrate Linctus IP'66
- 4. Solutions** a) Iodine Throat Paint (Mandles Paint)  
b) Strong solution of ammonium acetate  
c) Cresol with soap solution  
d) Lugol's solution
- 5. Suspensions** a) Calamine lotion  
b) Magnesium Hydroxide mixture  
c) Aluminium Hydroxide gel
- 6. Emulsions** a) Turpentine Liniment  
b) Liquid paraffin emulsion
- 7. Powders and Granules** a) ORS powder (WHO)  
b) Effervescent granules  
c) Dusting powder  
d) Divided powders
- 8. Suppositories** a) Glycero gelatin suppository  
b) Cocoa butter suppository  
c) Zinc Oxide suppository
- 9. Semisolids** a) Sulphur ointment  
b) Non staining-iodine ointment with methyl salicylate  
c) Carbopal gel
- 10. Gargles and Mouthwashes** a) Iodine gargle  
b) Chlorhexidine mouthwash

  
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### Recommended Books:

1. H.C. Ansel et al., Pharmaceutical Dosage Form and Drug Delivery System, Lippincott Williams and Walkins, New Delhi.
2. Carter S.J., Cooper and Gunn's-Dispensing for Pharmaceutical Students, CBS publishers, New Delhi.
3. M.E. Aulton, Pharmaceutics, The Science & Dosage Form Design, Churchill Livingstone, Edinburgh.
4. Indian pharmacopoeia.
5. British pharmacopoeia.
6. Lachmann. Theory and Practice of Industrial Pharmacy, Lea & Febiger Publisher, The University of Michigan.
7. Alfonso R. Gennaro Remington. The Science and Practice of Pharmacy, Lippincott Williams, New Delhi.
8. Carter S.J., Cooper and Gunn's. Tutorial Pharmacy, CBS Publications, New Delhi.
9. E.A. Rawlins, Bentley's Text Book of Pharmaceutics, English Language Book Society, Elsevier Health Sciences, USA.
10. Isaac Ghebre Sellassie: Pharmaceutical Pelletization Technology, Marcel Dekker, INC, New York.
11. Dilip M. Parikh: Handbook of Pharmaceutical Granulation Technology, Marcel Dekker, INC, New York.
12. Françoise Nieloud and Gilberte Marti-Mestres: Pharmaceutical Emulsions and Suspensions, Marcel Dekker, INC, New York.

  
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### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand basic principles and calculations involved in the conduct of experiments to prepare different dosage forms and prepare, dispense and evaluate conventional dosage forms for the benefit of society.
CO2	To interpret prescription, drug interactions, and adverse effects associated with active pharmaceutical ingredients and excipients. Evaluate possible incompatibilities in dosage formulations and methods to overcome
CO3	To apply knowledge of posology to calculate the dose of pediatrics and other population

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	1	2	-	3	2	3	3
CO2	3	-	2	2	-	-	-	1	2	-	3	2	3	3
CO3	3	2	3	3	-	1	1	-	2	-	3	1	3	3

  
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## BP 110 P: PHARMACEUTICAL INORGANIC CHEMISTRY (Practical)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	4 hours

#### I. Limit tests for following ions

- Limit test for Chlorides and Sulphates
- Modified limit test for Chlorides and Sulphates
- Limit test for Iron
- Limit test for Heavymetals
- Limit test for Lead
- Limit test for Arsenic

#### II. Identification test for Magnesium hydroxide, Ferrous sulphate, Sodium bicarbonate, Calcium gluconate, Copper sulphate

#### III. Test for purity

- Swelling power of Bentonite
- Neutralizing capacity of aluminum hydroxide gel
- Determination of potassium iodate and iodine in potassium iodide

#### IV. Preparation of inorganic pharmaceuticals

- Boric acid
- Potash alum
- Ferrous sulphate

### Recommended Books:

- A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London, 4th edition.
- A.I. Vogel, Text Book of Quantitative Inorganic analysis
- P. Gundu Rao, Inorganic Pharmaceutical Chemistry, 3rd Edition
- M.L Schroff, Inorganic Pharmaceutical Chemistry
- Bentley and Driver's Textbook of Pharmaceutical Chemistry
- Anand & Chatwal, Inorganic Pharmaceutical Chemistry

  
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## 7. Indian Pharmacopoeia

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To define impurities and their types in pharmaceutical preparations. Understand the methods for determination of impurities in inorganic drugs and pharmaceuticals.
CO2	To classify inorganic pharmaceuticals based on uses and properties. Understand the limit test and its importance in pharmaceuticals.
CO3	To attain expertise in concentration calculations of a solution, its preparation, standardization, and its storage conditions.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	-	-	-	-	1	2	1	3	1	-	3
CO2	3	-	2	-	-	-	-	1	2	1	3	1	-	3
CO3	3	2	3	2	-	-	-	-	2	1	3	1	-	3

  
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**BP 111 P: COMMUNICATION SKILLS (Practical)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	2	1	10	15	25	2 hours

The following learning modules are to be conducted using wordsworth® English language lab software

**Basic communication covering the following topics**

Meeting People, Asking Questions, Making Friends, What did you do?, Do's and Don't's

**Pronunciations covering the following topics**

Pronunciation (Consonant Sounds), Pronunciation and Nouns, Pronunciation (Vowel Sounds)

**Advanced Learning**

Listening Comprehension / Direct and Indirect Speech, Figures of Speech, Effective Communication, Writing Skills, Effective Writing, Interview Handling Skills, E-Mail etiquette  
Presentation Skills.

**Recommended Books:**

1. Basic communication skills for Technology, Andreja. J. Ruther Ford, 2nd Edition, Pearson Education, 2011
2. Communication skills, Sanjay Kumar, Pushpalata, 1st Edition, Oxford Press, 2011
3. Organizational Behaviour, Stephen .P. Robbins, 1st Edition, Pearson, 2013
4. Brilliant- Communication skills, Gill Hasson, 1st Edition, Pearson Life, 2011
5. The Ace of Soft Skills: Attitude, Communication and Etiquette for success, Gopala Swamy Ramesh, 5th Edition, Pearson, 2013
6. Developing your influencing skills, Deborah Dalley, Lois Burton, Margaret, Greenhall, 1<sup>st</sup> Edition Universe of Learning LTD, 2010
7. Communication skills for professionals, Konar nira, 2nd Edition, New arrivals –PHI, 2011
13. Personality development and soft skills, Barun K Mitra, 1st Edition, Oxford Press, 2011
14. Soft skill for everyone, Butter Field, 1st Edition, Cengage Learning india pvt. ltd, 2011
15. Soft skills and professional communication, Francis Peters SJ, 1st Edition, Mc Graw Hill Education, 2011
16. Effective communication, John Adair, 4th Edition, Pan Mac Millan, 2009

  
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17. Bringing out the best in people, Aubrey Daniels, 2ndEdition, Mc Graw Hill, 1999

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	05	15
Total Marks	25		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To become proficient in reading, writing, listening, and oral skills.
CO2	To learn basic concepts of debate, discussion, etc., and their implementation.
CO3	To attain good leadership quality with overall personality development.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	-	-	-	-	3	2	-	3	-	-	3	-	-	2
CO2	-	-	-	-	3	2	-	3	-	-	3	-	-	2
CO3	-	-	-	-	3	2	-	3	-	-	3	-	-	2

  
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**BP 112 RBP. REMEDIAL BIOLOGY (Practical)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	2	1	10	15	25	2 hours

1. Introduction to experiments in biology
  - a) Study of Microscope
  - b) Section cutting techniques
  - c) Mounting and staining
  - d) Permanent slide preparation
2. Study of cell and its inclusions
3. Study of Stem, Root, Leaf, seed, fruit, flower and their modifications
4. Detailed study of frog by using computer models
5. Microscopic study and identification of tissues pertinent to Stem, Root, Leaf, seed, fruit and flower
6. Identification of bones
7. Determination of blood group
8. Determination of blood pressure
9. Determination of tidal volume

#### Recommended Books:

11. Text book of Biology by S. B. Gokhale
12. A Text book of Biology by Dr. Thulajappa and Dr. Seetaram
13. A Text book of Biology by B.V. Sreenivasa Naidu
14. A Text book of Biology by Naidu and Murthy
15. Botany for Degree students By A.C.Dutta.
16. Outlines of Zoology by M. Ekambaranatha Ayyer and T. N. Ananthkrishnan.
17. A manual for pharmaceutical biology practical by S.B. Gokhale and C. K. Kokate
18. Practical human anatomy and physiology. by S.R. Kale and R.R. Kale.
19. A Manual of pharmaceutical biology practical by S.B. Gokhale, C.K. Kokate and S.P. Shriwastava.
20. Biology practical manual according to National core curriculum. Biology forum of Karnataka.  
Prof. M.J.H. Shafi

  
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## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	05	15
Total Marks	25		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To memorize and understand the physiology of the human body and plants.
CO2	To identify plants and animals based on morphological characters.
CO3	To apply the attained knowledge professionally.

## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	2	-	-	-	-	-	-	-	1	-	3	2	-	2
CO2	2	-	-	-	-	-	-	-	1	-	3	2	-	2
CO3	2	-	-	-	-	-	-	-	1	-	3	2	-	2

  
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## 1<sup>st</sup> Year (Semester II)

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**BP 201 T: HUMAN ANATOMY AND PHYSIOLOGY-II (Theory)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject is designed to impart fundamental knowledge on the structure and functions of the various systems of the human body. It also helps in understanding both homeostatic mechanisms. The subject provides the basic knowledge required to understand the various disciplines of pharmacy.

**Objectives:** Upon completion of this course the student should be able to:

- Explain the gross morphology, structure and functions of various organs of the human body.
- Describe the various homeostatic mechanisms and their imbalances.
- Identify the various tissues and organs of different systems of human body.
- Perform the hematological tests like blood cell counts, haemoglobin estimation, bleeding/clotting time etc and also record blood pressure, heart rate, pulse and respiratory volume.
- Appreciate coordinated working pattern of different organs of each system
- Appreciate the interlinked mechanisms in the maintenance of normal functioning (homeostasis) of human body.

**COURSE CONTENT**

UNIT	CONTENT	No. of Hrs.
I	<b>Nervous system:</b> Organization of nervous system, neuron, neuroglia, classification and properties of nerve fibre, electrophysiology, action potential, nerve impulse, receptors, synapse, neurotransmitters. Central nervous system: Meninges, ventricles of brain and cerebrospinal fluid. structure and functions of brain (cerebrum, brainstem, cerebellum), spinal cord (gross structure, functions of afferent and efferent nerve tracts, reflex activity)	10

  
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II	<b>Digestive system:</b> Anatomy of GI Tract with special reference to anatomy and functions of stomach,(Acid production in the stomach, regulation of acid production throughparasympathetic nervous system, pepsin role in protein digestion) small intestineand large intestine, anatomy and functions of salivary glands, pancreas and liver,movements of GIT, digestion and absorption of nutrients and disorders of GIT. <b>Energetics:</b> Formation and role of ATP, Creatinine Phosphate and BMR.	10
III	<b>Respiratory system:</b> Anatomy of respiratory system with special reference to anatomy of lungs,mechanism of respiration, regulation of respirationLung Volumes and capacities transport of respiratory gases, artificial respiration,and resuscitation methods. <b>Urinary system:</b> Anatomy of urinary tract with special reference to anatomy of kidney andnephrons, functions of kidney and urinary tract, physiology of urine formation,micturition reflex and role of kidneys in acid base balance, role of RAS in kidneyand disorders of kidney.	10
IV	<b>Endocrine system</b> Classification of hormones, mechanism of hormone action, structureand functions of pituitary gland, thyroid gland, parathyroid gland,adrenalgland, pancreas, pineal gland, thymus and their disorders.	10
V	<b>Reproductive system</b> Anatomy of male and female reproductive system, Functions of male and femalereproductive system, sex hormones, physiology of menstruation, fertilization,spermatogenesis, oogenesis, pregnancy and parturition <b>Introduction to genetics</b> Chromosomes, genes and DNA, protein synthesis, genetic pattern of inheritance	09

### Recommended Books

1. Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam.Jaypee brothers medical publishers, New Delhi.
2. Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York

  
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3. Physiological basis of Medical Practice-Best and Taylor. Williams & Wilkins Co, Riverview, MI USA
4. Textbook of Medical Physiology- Arthur C, Guyton and John.E. Hall. Miamisburg, OH, U.S.A.
5. Principles of Anatomy and Physiology by Tortora Grabowski. Palmetto, GA, U.S.A. 32
6. Textbook of Human Histology by Inderbir Singh, Jaypee brother's medical publishers, New Delhi.
7. Textbook of Practical Physiology by C.L. Ghai, Jaypee brother's medical publishers, New Delhi.
8. Physiological basis of Medical Practice-Best and Taylor. Williams & Wilkins Co Riverview, MI USA
9. Text book of Medical Physiology- Arthur C, Guyton and John. E. Hall. Miamisburg, OH, U.S.A.
10. Human Physiology (vol 1 and 2) by Dr. C.C. Chatterrje, Academic Publishers.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To memorize the anatomy, terminology, and gross morphology of various organs of the human body.
CO2	To understand the physiology of various organ systems of the body and their interlinked mechanisms.
CO3	To analyze and apply the knowledge of anatomy and physiology in the body system to various disorders.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	2	-	-	-	1	2	-	3	2	3	-
CO2	3	-	3	2	-	-	-	1	2	-	3	2	3	-
CO3	3	2	3	1	-	-	-	-	2	-	3	1	3	-

  
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## BP 202 T: PHARMACEUTICAL ORGANIC CHEMISTRY –I (Theory)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject deals with classification and nomenclature of simple organic compounds, structural isomerism, intermediates forming in reactions, important physical properties, reactions and methods of preparation of these compounds. The syllabus also emphasizes on mechanisms and orientation of reactions.

**Objectives:** Upon completion of the course the student shall be able to

- write the structure, name and the type of isomerism of the organic compound
- write the reaction, name the reaction and orientation of reactions
- account for reactivity/stability of compounds,
- identify/confirm the identification of organic compound

### COURSE CONTENT

General methods of preparation and reactions of compounds superscripted with asterisk (\*) to be Explained To emphasize on definition, types, classification, principles/mechanisms, applications, examples and differences

UNIT	CONTENT	No. of Hrs.
I	<b>Classification, nomenclature and isomerism:</b> Classification of Organic Compounds, Common and IUPAC systems of nomenclature of organic compounds (up to 10 Carbons open chain and carbocyclic compounds). Structural isomerisms in organic compounds	07
II	<b>Alkanes*, Alkenes* and Conjugated dienes*:</b> SP <sup>3</sup> hybridization in alkanes, Halogenation of alkanes, uses of paraffins. Stabilities of alkenes, SP <sup>2</sup> hybridization in alkenes E1 and E2 reactions – kinetics, order of reactivity of alkyl halides, rearrangement of carbocations, Saytzeff's orientation and evidences. E1 versus E2	10

  
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	reactions, Factors affecting E1 and E2 reactions. Ozonolysis, electrophilic addition reactions of alkenes, Markownikoff's orientation, free radical addition reactions of alkenes, Anti Markownikoff's orientation. Stability of conjugated dienes, Diel-Alder, electrophilic addition, free radical addition reactions of conjugated dienes, allylic rearrangement	
III	<b>Alkyl halides*</b> : SN1 and SN2 reactions - kinetics, order of reactivity of alkyl halides, stereochemistry and rearrangement of carbocations. SN1 versus SN2 reactions, Factors affecting SN1 and SN2 reactions. Structure and uses of ethylchloride, Chloroform, trichloroethylene, tetrachloroethylene, dichloromethane, tetrachloromethane and iodoform. <b>Alcohols*</b> - Qualitative tests, Structure and uses of Ethyl alcohol, Methyl alcohol, chlorobutanol, Cetosteryl alcohol, Benzyl alcohol, Glycerol, Propylene glycol	10
IV	<b>Carbonyl compounds* (Aldehydes and ketones)</b> : Nucleophilic addition, Electromeric effect, aldol condensation, Crossed Aldol condensation, Cannizzaro reaction, Crossed Cannizzaro reaction, Benzoin condensation, Perkin condensation, qualitative tests, Structure and uses of Formaldehyde, Paraldehyde, Acetone, Chloral hydrate, Hexamine, Benzaldehyde, Vanilin, Cinnamaldehyde.	10
V	<b>Carboxylic acids*</b> : Acidity of carboxylic acids, effect of substituents on acidity, inductive effect and qualitative tests for carboxylic acids, amide and ester. Structure and Uses of Acetic acid, Lactic acid, Tartaric acid, Citric acid, Succinic acid. Oxalic acid, Salicylic acid, Benzoic acid, Benzyl benzoate, Dimethyl phthalate, Methyl salicylate and Acetyl salicylic acid <b>Aliphatic amines*</b> - Basicity, effect of substituent on Basicity. Qualitative test, Structure and uses of Ethanolamine, Ethylenediamine, Amphetamine	08

### Recommended Books

1. Organic Chemistry by Morrison and Boyd
2. Organic Chemistry by I.L. Finar, Volume-I
3. Textbook of Organic Chemistry by B.S. Bahl & Arun Bahl.

  
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4. Organic Chemistry by P.L. Soni
5. Practical Organic Chemistry by Mann and Saunders.
6. Vogel's text book of Practical Organic Chemistry
7. Advanced Practical organic chemistry by N.K. Vishnoi.
8. Introduction to Organic Laboratory techniques by Pavia, Lampman and Kriz.
9. Reaction and reaction mechanism by Ahluwalia/ Chatwal.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the classification and nomenclature of organic compounds and memorize the structure, and physical and chemical properties of organic compounds.
CO2	To synthesize organic compounds and understand mechanisms and orientation of reactions. Demonstrate various applications of organic compounds.
CO3	To attain expertise in basic learning and application of Organic chemistry in Pharmacy and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	1	2	-	3	1	1	3
CO2	3	-	2	2	-	-	-	1	2	-	3	1	1	3
CO3	3	2	3	2	-	-	2	1	2	1	3	1	1	3

  
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### BP 203 T. BIOCHEMISTRY (Theory)

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** Biochemistry deals with complete understanding of the molecular levels of the chemical process associated with living cells. The scope of the subject is providing biochemical facts and the principles to understand metabolism of nutrient molecules in physiological and pathological conditions. It is also emphasizing on genetic organization of mammalian genome and hetero & autocatalytic functions of DNA.

**Objectives:** Upon completion of course student shall be able to

- Understand the catalytic role of enzymes, importance of enzyme inhibitors in design of new drugs, therapeutic and diagnostic applications of enzymes.
- Understand the metabolism of nutrient molecules in physiological and pathological conditions.
- Understand the genetic organization of mammalian genome and functions of DNA in the synthesis of RNAs and proteins.

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Biomolecules:</b> Introduction, classification, chemical nature and biological role of carbohydrate, lipids, nucleic acids, amino acids and proteins. <b>Bioenergetics:</b> Concept of free energy, endergonic and exergonic reaction, Relationship between free energy, enthalpy and entropy; Redox potential. Energy rich compounds; classification; biological significances of ATP and cyclic AMP	08
II	<b>Carbohydrate metabolism:</b> Glycolysis – Pathway, energetics and significance Citric acid cycle- Pathway, energetics and significance HMP shunt and its significance; Glucose-6-Phosphate dehydrogenase (G6PD) deficiency ; Glycogen metabolism Pathways and glycogen storage diseases (GSD); Gluconeogenesis- Pathway and its significance	10

  
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	Hormonal regulation of blood glucose level and Diabetes mellitus <b>Biological oxidation:</b> Electron transport chain (ETC) and its mechanism. Oxidative phosphorylation & its mechanism and substrate phosphorylation. Inhibitors ETC and oxidative phosphorylation/Uncouplers level	
III	<b>Lipid metabolism:</b> $\beta$ -Oxidation of saturated fatty acid (Palmitic acid) Formation and utilization of ketone bodies; ketoacidosis <i>De novo</i> synthesis of fatty acids (Palmitic acid) Biological significance of cholesterol and conversion of cholesterol into bile acids, steroid hormone and vitamin D Disorders of lipid metabolism: Hypercholesterolemia, atherosclerosis, fatty liver and obesity. <b>Amino acid metabolism:</b> General reactions of amino acid metabolism: Transamination, deamination & decarboxylation, urea cycle and its disorders Catabolism of phenylalanine and tyrosine and their metabolic disorders (Phenylketonuria, Albinism, alcaptonuria, tyrosinemia) Synthesis and significance of biological substances; 5-HT, melatonin, dopamine, noradrenaline, adrenaline, Catabolism of heme; hyperbilirubinemia and jaundice	10
IV	<b>Nucleic acid metabolism and genetic information transfer</b> Biosynthesis of purine and pyrimidine nucleotides Catabolism of purine nucleotides and Hyperuricemia and Gout disease Organization of mammalian genome Structure of DNA and RNA and their functions DNA replication (semi conservative model) Transcription or RNA synthesis Genetic code, Translation or Protein synthesis and inhibitors	10
V	<b>Enzymes:</b> Introduction, properties, nomenclature and IUB classification of enzymes, Enzyme kinetics (Michaelis plot, Line Weaver Burke plot) Enzyme inhibitors with examples, Regulation of enzymes: enzyme induction and repression, allosteric enzymes regulation, Therapeutic and diagnostic applications of enzymes and isoenzymes, Coenzymes – Structure and biochemical functions	07

  
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### Recommended Books

1. Principles of Biochemistry by Lehninger.
2. Harper's Biochemistry by Robert K. Murry, Daryl K. Granner and Victor W. Rodwell.
3. Biochemistry by Stryer.
4. Biochemistry by D. Satyanarayan and U.Chakrapani
5. Textbook of Biochemistry by Rama Rao.
6. Textbook of Biochemistry by Deb.
7. Outlines of Biochemistry by Conn and Stumpf
8. Practical Biochemistry by R.C. Gupta and S. Bhargavan.
9. Introduction of Practical Biochemistry by David T. Plummer. (3rd Edition)
10. Practical Biochemistry for Medical students by Rajagopal and Ramakrishna.
11. Practical Biochemistry by Harold Varley.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the concept of biomolecules involved in various biochemical processes and energy generation in the body.
CO2	To demonstrate the biochemical pathways operating in our body and the various diseases associated with them.
CO3	To attain expertise in basic learning and application of Biochemistry in Pharmacy and Industry

  
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## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	-	2	-	-	-	1	2	-	3	1	-	1
CO2	3	-	-	2	-	-	-	1	2	-	3	1	-	1
CO3	3	-	-	2	-	-	2	1	2	-	3	2	-	1

  
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## BP 204 T. PATHOPHYSIOLOGY (THEORY)

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** Pathophysiology is the study of causes of diseases and reactions of the body to such disease producing causes. This course is designed to impart a thorough knowledge of the relevant aspects of pathology of various conditions with reference to its pharmacological applications, and understanding of basic pathophysiological mechanisms. Hence it will not only help to study the syllabus of pathology, but also to get baseline knowledge required to practice medicine safely, confidently, rationally and effectively.

**Objectives:** Upon completion of the subject student shall be able to –

- Describe the etiology and pathogenesis of the selected disease states;
- Name the signs and symptoms of the diseases; and
- Mention the complications of the diseases.

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Basic principles of Cell injury and Adaptation:</b> Introduction, definitions, Homeostasis, Components and Types of Feedback systems, Causes of cellular injury, Pathogenesis (Cell membrane damage, Mitochondrial damage, Ribosome damage, Nuclear damage), Morphology of cell injury – Adaptive changes (Atrophy, Hypertrophy, hyperplasia, Metaplasia, Dysplasia), Cell swelling, Intra cellular accumulation, Calcification, Enzyme leakage and Cell Death Acidosis & Alkalosis, Electrolyte imbalance <b>Basic mechanism involved in the process of inflammation and repair:</b> Introduction, Clinical signs of inflammation, Different types of Inflammation, Mechanism, of Inflammation – Alteration in vascular permeability and blood flow, migration of WBC's, Mediators of inflammation, Basic principles of wound healing in the skin, Pathophysiology of Atherosclerosis.	10
II	<b>Cardiovascular System:</b> Hypertension, congestive heart failure, ischemic heart	10

  
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	disease (angina, myocardial infarction, atherosclerosis and arteriosclerosis) <b>Respiratory system:</b> Asthma, Chronic obstructive airways diseases. <b>Renal system:</b> Acute and chronic renal failure .	
<b>III</b>	<b>Haematological Diseases:</b> Iron deficiency, megaloblastic anemia (Vit B12 and folic acid), sickle cell anemia, thalassemia, hereditary acquired anemia, hemophilia <b>Endocrine system:</b> Diabetes, thyroid diseases, disorders of sex hormones <b>Nervous system:</b> Epilepsy, Parkinson's disease, stroke, psychiatric disorders: depression, schizophrenia and Alzheimer's disease. <b>Gastrointestinal system:</b> Peptic Ulcer	<b>10</b>
<b>IV</b>	Inflammatory bowel diseases, jaundice, hepatitis (A,B,C,D,E,F) alcoholic liver disease. <b>Disease of bones and joints:</b> Rheumatoid arthritis, osteoporosis and gout <b>Principles of cancer:</b> classification, etiology and pathogenesis of cancer <b>Diseases of bones and joints:</b> Rheumatoid Arthritis, Osteoporosis, Gout <b>Principles of Cancer:</b> Classification, etiology and pathogenesis of Cancer	<b>08</b>
<b>V</b>	<b>Infectious diseases:</b> Meningitis, Typhoid, Leprosy, Tuberculosis, Urinary tract infections <b>Sexually transmitted diseases:</b> AIDS, Syphilis, Gonorrhoea	<b>07</b>

### Recommended Books

1. Vinay Kumar, Abul K. Abas, Jon C. Aster; Robbins & Cotran Pathologic Basis of Disease; South Asia edition; India; Elsevier; 2014.
2. Harsh Mohan; Text book of Pathology; 6th edition; India; Jaypee Publications; 2010.
3. Laurence B, Bruce C, Bjorn K.; Goodman Gilman's The Pharmacological Basis of Therapeutics; 12th edition; New York; McGraw-Hill; 2011.
4. Best, Charles Herbert 1899-1978; Taylor, Norman Burke 1885-1972; West, John B (John Burnard); Best and Taylor's Physiological basis of medical practice; 12th ed; United States;
5. William and Wilkins, Baltimore; 1991 [1990 printing].
6. Nicki R. Colledge, Brian R. Walker, Stuart H. Ralston; Davidson's Principles and Practice of Medicine; 21st edition; London; ELBS/Churchill Livingstone; 2010.

  
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7. Guyton A, John E Hall; Textbook of Medical Physiology; 12th edition; WB Saunders Company; 2010.
8. Joseph DiPiro, Robert L. Talbert, Gary Yee, Barbara Wells, L. Michael Posey; Pharmacotherapy: A Pathophysiological Approach; 9th edition; London; McGraw Hill Medical; 2014.
9. V. Kumar, R. S. Cotran and S. L. Robbins; Basic Pathology; 6th edition; Philadelphia; WB; Saunders Company; 1997.
10. Roger Walker, Clive Edwards; Clinical Pharmacy and Therapeutics; 3rd edition; London; Churchill Livingstone publication; 2003.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basic pathophysiology of various diseases and disorders of the living system.
CO2	To understand and correlate the therapeutics with various mechanisms for the development and progression of various diseases and disorders of the living system.
CO3	To attain expertise in basic learning and application of Pathophysiology in Pharmacy and Hospital.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	-	-	-	-	-	1	2	-	3	3	1	-
CO2	3	-	-	-	-	-	-	1	2	-	3	3	1	-
CO3	3	-	-	-	-	-	-	1	2	-	3	-	1	-

  
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## BP 205 T. COMPUTER APPLICATIONS IN PHARMACY (Theory)

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	0	0	3	25	75	100	3 hours

**Scope:** This subject deals with the introduction Database, Database Management system, computer application in clinical studies and use of databases.

**Objectives:** Upon completion of the course the student shall be able to

- know the various types of application of computers in pharmacy
- know the various types of databases
- know the various applications of databases in pharmacy

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Number system:</b> Binary number system, Decimal number system, Octal number system, Hexadecimal number systems, conversion decimal to binary, binary to decimal, octal to binary etc, binary addition, binary subtraction – One's complement, Two's complement method, binary multiplication, binary division <b>Concept of Information Systems and Software :</b> Information gathering, requirement and feasibility analysis, data flow diagrams, process specifications, input/output design, process life cycle, planning and managing the project	06
II	<b>Web technologies:</b> Introduction to HTML, XML, CSS and Programming languages, introduction to web servers and Server Products, Introduction to databases, MYSQL, MS ACCESS, Pharmacy Drug database	06
III	<b>Application of computers in Pharmacy</b> – Drug information storage and retrieval, Pharmacokinetics, Mathematical model in Drug design, Hospital and Clinical Pharmacy, Electronic Prescribing and discharge (EP) systems, barcode medicine identification and automated dispensing of drugs, mobile technology and adherence monitoring, Diagnostic System, Lab-diagnostic System,	06

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	Patient Monitoring System,Pharma Information System	
IV	<b>Bioinformatics:</b> Introduction, Objective of Bioinformatics, BioinformaticsDatabases, Concept of Bioinformatics, Impact of Bioinformatics in Vaccine Discovery	06
V	<b>Computers as data analysis in Preclinical development:</b> Chromatographic dada analysis(CDS), Laboratory Information management System (LIMS) and Text Information Management System(TIMs)	06

### Recommended books:

1. Computer Application in Pharmacy – William E. Fassett – Lea and Febiger, 600 South Washington Square, USA, (215) 922-1330.
2. Computer Application in Pharmaceutical Research and Development –Sean Ekins –Wiley Interscience, A John Willey and Sons, INC., Publication, USA.
3. Bioinformatics (Concept, Skills and Applications) – S.C. Rastogi- CBS Publishers and Distributors, 4596/1- A, 11 Darya Gani, New Delhi – 110 002 (INDIA).
4. Microsoft office Access - 2003, Application Development Using VBA, SQL Server, DAP and Infopath – Cary N. Prague – Wiley Dreamtech India (P) Ltd., 4435/7, Ansari Road, Daryagani, New Delhi – 110002.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

	Theory		
Components	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

  
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**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the utilization of computers in pharmacies, hospitals, Industry, clinical studies, and use of databases.
CO2	To learn the application of computers in Bioinformatics, drug discovery, and data analysis.
CO3	To attain proficiency in working with computers for various applications.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	1	-	-	1	-	-	-	-	1	-	3	-	-	-
CO2	1	-	1	3	-	-	-	-	-	-	3	2	-	-
CO3	1	1	3	2	-	-	-	3	1	-	3	1	-	1

  
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**BP 206 T. ENVIRONMENTAL SCIENCES (Theory)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	0	0	3	25	75	100	3 hours

**Scope:** Environmental Sciences is the scientific study of the environmental system and the status of its inherent or induced changes on organisms. It includes not only the study of physical and biological characters of the environment but also the social and cultural factors and the impact of man on environment.

**Objectives:** Upon completion of the course the student shall be able to:

- Create the awareness about environmental problems among learners.
- Impart basic knowledge about the environment and its allied problems.
- Develop an attitude of concern for the environment.
- Motivate learner to participate in environment protection and environment improvement.
- Acquire skills to help the concerned individuals in identifying and solving environmental problems.
- Strive to attain harmony with Nature.

**COURSE CONTENT**

UNIT	CONTENT	No. of Hrs.
I	The Multidisciplinary nature of environmental studies: Natural Resources Renewable and non-renewable resources: Natural resources and associated problems a) Forest resources; b) Water resources; c) Mineral resources; d) Food resources; e) Energy resources; f) Land resources: Role of an individual in conservation of natural resources.	10
II	Ecosystems : Concept of an ecosystem, Structure and function of an ecosystem, Introduction, types, characteristic features, structure and function of the ecosystems: Forest ecosystem; Grassland ecosystem; Desert ecosystem; Aquatic ecosystems (ponds, streams, lakes, rivers, oceans, estuaries)	10
III	Environmental Pollution: Air pollution; Water pollution; Soil pollution	10

**Recommended Books:**

1. Y.K. Sing, Environmental Science, New Age International Pvt, Publishers, Bangalore

  
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2. Agarwal, K.C. 2001 Environmental Biology, Nidi Publ. Ltd. Bikaner.
3. Bharucha Erach, The Biodiversity of India, Mapin Pu blishing Pvt. Ltd., Ahmedabad – 380013, India.
4. Brunner R.C., 1989, Hazardous Waste Incineration, McGraw Hill Inc. 480p
5. Clark R.S., Marine Pollution, Clarendon Press Oxford
1. Cunningham, W.P. Cooper, T.H. Gorhani, E & Hepworth, M.T. 2001, Environmental Encyclopedia, Jaico Publ. House, Mumbai, 1196p
6. De A.K., Environmental Chemistry, Wiley Eastern Ltd.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand different dimensions of environmental studies; problems related to environmental degradation & the remedial steps taken to address them.
CO2	To understand the diverse problems associated with solid waste, as well as waste segregation and effective management techniques.
CO3	To create awareness about environmental issues and carry out outreach activities.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	-	-	-	-	-	-	-	-	-	3	3	-	-	-
CO2	-	-	-	-	-	-	-	-	-	3	3	-	-	-
CO3	-	1	3	-	-	-	-	3	-	3	3	-	-	-

  
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**BP 207 P: HUMAN ANATOMY AND PHYSIOLOGY (Practical)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
0	0	2	1	10	15	25	4 hours

Practical physiology is complimentary to the theoretical discussions in physiology. Practicals allow the verification of physiological processes discussed in theory classes through experiments on living tissue, intact animals or normal human beings. This is helpful for developing an insight on the subject.

1. To study the integumentary and special senses using specimen, models, etc.,
2. To study the nervous system using specimen, models, etc.,
3. To study the endocrine system using specimen, models, etc
4. To demonstrate the general neurological examination
5. To demonstrate the function of olfactory nerve
6. To examine the different types of taste.
7. To demonstrate the visual acuity
8. To demonstrate the reflex activity
9. Recording of body temperature
10. To demonstrate positive and negative feedback mechanism.
11. Determination of tidal volume and vital capacity.
12. Study of digestive, respiratory, cardiovascular systems, urinary and reproductive systems with the help of models, charts and specimens.
13. Recording of basal mass index .
14. Study of family planning devices and pregnancy diagnosis test.
15. Demonstration of total blood count by cell analyser
16. Permanent slides of vital organs and gonads.

  
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### Recommended Books

1. Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypee brothers medical publishers, New Delhi.
2. Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York
3. Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co, Riverview, MI USA
4. Textbook of Medical Physiology- Arthur C, Guyton and John.E. Hall. Miamisburg, OH, U.S.A.
5. Principles of Anatomy and Physiology by Tortora Grabowski. Palmetto, GA, U.S.A. 32
6. Textbook of Human Histology by Inderbir Singh, Jaypee brother's medical publishers, New Delhi.
7. Textbook of Practical Physiology by C.L. Ghai, Jaypee brother's medical publishers, New Delhi.
8. Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co Riverview, MI USA
9. Text book of Medical Physiology- Arthur C, Guyton and John. E. Hall. Miamisburg, OH, U.S.A.
10. Human Physiology (vol 1 and 2) by Dr. C.C. Chatterrje, Academic Publishers.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

	Practical		
Components	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To memorize the anatomy, terminology, and gross morphology of various organs of the human body.
CO2	To understand the physiology of various organ systems of the body and their interlinked mechanisms.

  
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CO3	To analyze and apply the knowledge of anatomy and physiology in the body system to various disorders
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### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	2	-	-	-	1	2	-	3	2	3	-
CO2	3	-	3	2	-	-	-	1	2	-	3	2	3	-
CO3	3	2	3	1	-	-	-	-	2	-	3	1	3	-

  
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**BP208P. PHARMACEUTICAL ORGANIC CHEMISTRY -I (Practical)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	4 hours

#### Systematic qualitative analysis of unknown organic compounds

1. Preliminary test: Color, odour, aliphatic/aromatic compounds, saturation and unsaturation, etc.
2. Detection of elements like Nitrogen, Sulphur and Halogen by Lassaigne's test
3. Solubility test
4. Functional group test like Phenols, Amides/ Urea, Carbohydrates, Amines, Carboxylic acids, Aldehydes and Ketones, Alcohols, Esters, Aromatic and Halogenated Hydrocarbons, Nitro compounds and Anilides.
5. Melting point/Boiling point of organic compounds
6. Identification of the unknown compound from the literature using melting point/ boiling point.
7. Preparation of the derivatives and confirmation of the unknown compound by melting point/ boiling point.
8. Minimum 5 unknown organic compounds to be analyzed systematically.
9. Preparation of suitable solid derivatives from organic compounds
10. Construction of molecular models

#### Recommended Books

1. Organic Chemistry by Morrison and Boyd
2. Organic Chemistry by I.L. Finar, Volume-I
3. Textbook of Organic Chemistry by B.S. Bahl & Arun Bahl.
4. Organic Chemistry by P.L. Soni
5. Practical Organic Chemistry by Mann and Saunders.
6. Vogel's text book of Practical Organic Chemistry
7. Advanced Practical organic chemistry by N.K. Vishnoi.

  
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8. Introduction to Organic Laboratory techniques by Pavia, Lampman and Kriz.
9. Reaction and reaction mechanism by Ahluwalia/ Chatwal.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:


Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn the basics of organic chemistry. Identify organic compounds and elements via physic-chemical testing. Determine the physical properties of organic compounds.
CO2	To synthesize organic compounds and understand mechanisms and orientation of reactions. Demonstrate stereo modeling and its applications.
CO3	To analyze and apply the knowledge of Pharmaceutical Organic Chemistry in Pharmacy and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	-	-	-	-	1	2	1	3	1	-	3
CO2	3	-	2	-	-	-	-	1	2	1	3	1	-	3
CO3	3	2	3	2	-	-	-	-	2	1	3	1	-	3

  
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**BP 209 P. BIOCHEMISTRY (Practical)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	2 hours

1. Qualitative analysis of carbohydrates (Glucose, Fructose, Lactose, Maltose, Sucrose and starch)
2. Identification tests for Proteins (albumin and Casein)
3. Quantitative analysis of reducing sugars (DNSA method) and Proteins (Biuret method)
4. Qualitative analysis of urine for abnormal constituents
5. Determination of blood creatinine
6. Determination of blood sugar
7. Determination of serum total cholesterol
8. Preparation of buffer solution and measurement of pH
9. Study of enzymatic hydrolysis of starch
10. Determination of Salivary amylase activity
11. Study the effect of Temperature on Salivary amylase activity.
12. Study the effect of substrate concentration on salivary amylase activity.

### Recommended Books

1. Principles of Biochemistry by Lehninger.
2. Harper's Biochemistry by Robert K. Murry, Daryl K. Granner and Victor W. Rodwell.
3. Biochemistry by Stryer.
4. Biochemistry by D. Satyanarayan and U. Chakrapani
5. Textbook of Biochemistry by Rama Rao.
6. Textbook of Biochemistry by Deb.
7. Outlines of Biochemistry by Conn and Stumpf
8. Practical Biochemistry by R.C. Gupta and S. Bhargavan.
9. Introduction of Practical Biochemistry by David T. Plummer. (3rd Edition)
10. Practical Biochemistry for Medical students by Rajagopal and Ramakrishna.
11. Practical Biochemistry by Harold Varley.

  
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## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the concept of biomolecules involved in various biochemical processes and energy generation in the body.
CO2	To demonstrate the biochemical pathways operating in our body and the various diseases associated with them.
CO3	To attain expertise in basic learning and application of Biochemistry in Pharmacy and Industry.

## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	-	2	-	3	1	-	1
CO2	3	-	2	2	-	-	-	-	2	-	3	1	-	1
CO3	3	-	3	2	-	-	1	1	2	-	3	1	-	1

  
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## BP210P. COMPUTER APPLICATIONS IN PHARMACY (Practical)

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	2	1	10	15	25	2 hours

1. Design a questionnaire using a word processing package to gather information about a particular disease.
2. Create a HTML web page to show personal information.
- 3 Retrieve the information of a drug and its adverse effects using online tools
- 4 Creating mailing labels Using Label Wizard , generating label in MS WORD
- 5 Create a database in MS Access to store the patient information with the required fields Using access
6. Design a form in MS Access to view, add, delete and modify the patient record in the database
7. Generating report and printing the report from patient database
8. Creating invoice table using – MS Access
9. Drug information storage and retrieval using MS Access
10. Creating and working with queries in MS Access
11. Exporting Tables, Queries, Forms and Reports to web pages
12. Exporting Tables, Queries, Forms and Reports to XML pages

### Recommended books:

1. Computer Application in Pharmacy – William E. Fassett – Lea and Febiger, 600 South Washington Square, USA, (215) 922-1330.
2. Computer Application in Pharmaceutical Research and Development –Sean Ekins –Wiley Interscience, A John Willey and Sons, INC., Publication, USA.
3. Bioinformatics (Concept, Skills and Applications) – S.C. Rastogi- CBS Publishers and Distributors, 4596/1- A, 11 Darya Gani, New Delhi – 110 002 (INDIA).

  
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4. Microsoft office Access - 2003, Application Development Using VBA, SQL Server, DAP and Infopath – Cary N. Prague – Wiley Dreamtech India (P) Ltd., 4435/7, Ansari Road, Daryagani, New Delhi – 110002.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:


Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the utilization of computers in pharmacies, hospitals, Industry, clinical studies, and use of databases.
CO2	To learn the application of computers in Bioinformatics, drug discovery, and data analysis.
CO3	To attain proficiency in working with computers for various applications.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	1	-	-	1	-	-	-	-	1	-	3	-	-	-
CO2	1	-	1	3	-	-	-	-	-	-	3	2	-	-
CO3	1	1	3	2	-	-	-	3	1	-	3	1	-	1

  
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**2<sup>nd</sup> Year (Semester III)**

**PCI Syllabus**

  
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**BP301T. PHARMACEUTICAL ORGANIC CHEMISTRY –II (Theory)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject deals with general methods of preparation and reactions of someorganic compounds. Reactivity of organic compounds are also studied here. The syllabusemphasizes on mechanisms and orientation of reactions. Chemistry of fats and oils arealso included in the syllabus.

**Objectives:** Upon completion of the course the student shall be able to

- write the structure, name and the type of isomerism of the organic compound
- write the reaction, name the reaction and orientation of reactions
- account for reactivity/stability of compounds,
- prepare organic compounds

#### COURSE CONTENT

General methods of preparation and reactions of compounds superscripted with asterisk (\*) to be ExplainedTo emphasize on definition, types, classification, principles/mechanisms, applications, examplesand differences

UNIT	CONTENT	No. of Hrs.
<b>I</b>	<b>Benzene and its derivatives:</b> A. Analytical, synthetic and other evidences in the derivation of structureof benzene, Orbital picture, resonance in benzene, aromaticcharacters, Huckel's rule B. Reactions of benzene - nitration, sulphonation, halogenations reactivity,Friedelcrafts alkylation- reactivity, limitations,Friedelcrafts acylation. C. Substituents, effect of substituents on reactivity and orientation ofmono substituted benzene compounds towards electrophilicsubstitution reaction D. Structure and uses of DDT, Saccharin, BHC and Chloramine	<b>10</b>
<b>II</b>	<b>Phenols*</b> - Acidity of phenols, effect of substituents on acidity, qualitativetests, Structure and uses of phenol, cresols, resorcinol, naphthols	<b>10</b>

  
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	<b>Aromatic Amines*</b> - Basicity of amines, effect of substituents on basicity, and synthetic uses of aryl diazonium salts <b>Aromatic Acids*</b> -Acidity, effect of substituents on acidity and important reactions of benzoic acid.	
<b>III</b>	<b>Fats and Oils</b> a. Fatty acids – reactions. b. Hydrolysis, Hydrogenation, Saponification and Rancidity of oils, Drying oils. c. Analytical constants – Acid value, Saponification value, Ester value, Iodine value, Acetyl value, Reichert Meissl (RM) value – significance and principle involved in their determination.	<b>10</b>
<b>IV</b>	<b>Polynuclear hydrocarbons:</b> a. Synthesis, reactions, structure and medicinal uses of Naphthalene, Phenanthrene, Anthracene, Diphenylmethane, Triphenylmethane and their derivatives	<b>08</b>
<b>V</b>	<b>Cyclo alkanes*</b> Stabilities – Baeyer's strain theory, limitation of Baeyer's strain theory, Coulson and Moffitt's modification, Sachse Mohr's theory (Theory of strainless rings), reactions of cyclopropane and cyclobutane only	<b>07</b>

### Recommended Books

1. Organic Chemistry by Morrison and Boyd
2. Organic Chemistry by I.L. Finar, Volume-I
3. Textbook of Organic Chemistry by B.S. Bahl & Arun Bahl.
4. Organic Chemistry by P.L. Soni
5. Practical Organic Chemistry by Mann and Saunders.
6. Vogel's text book of Practical Organic Chemistry
7. Advanced Practical organic chemistry by N.K. Vishnoi.
8. Introduction to Organic Laboratory techniques by Pavia, Lampman and Kriz.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

  
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Theory			
Components	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	Understand the classification and nomenclature of organic compounds and memorize the structure and physical and chemical properties of organic compounds.
CO2	Synthesize organic compounds and understand mechanisms and orientation of reactions. Demonstrate various applications of organic compounds.
CO3	Attain expertise in basic learning and application of Organic chemistry in Pharmacy and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	1	2	-	3	1	1	3
CO2	3	-	2	2	-	-	-	1	2	-	3	1	1	3
CO3	3	2	3	2	-	-	2	1	2	1	3	1	1	3

  
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**BP 302 T. PHYSICAL PHARMACEUTICS-I (Theory)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** The course deals with the various physical and physicochemical properties, and principles involved in dosage forms/formulations. Theory and practical components of the subject help the student to get a better insight into various areas of formulation research and development, and stability studies of pharmaceutical dosage forms.

**Objectives:** Upon the completion of the course student shall be able to

- Understand various physicochemical properties of drug molecules in the designing the dosage forms
- Know the principles of chemical kinetics & to use them for stability testing and determination of expiry date of formulations
- Demonstrate use of physicochemical properties in the formulation development and evaluation of dosage forms.

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Solubility of drugs:</b> Solubility expressions, mechanisms of solute solvent interactions, ideal solubility parameters, solvation & association, quantitative approach to the factors influencing solubility of drugs, diffusion principles in biological systems. Solubility of gas in liquids, solubility of liquids in liquids, (Binary solutions, ideal solutions) Raoult's law, real solutions, Partially miscible liquids, Critical solution temperature and applications. Distribution law, its limitations and applications.	10
II	<b>States of Matter and properties of matter:</b> State of matter, changes in the state of matter, latent heats, vapour pressure, sublimation critical point, eutectic mixtures, gases, aerosols- inhalers, relative humidity, liquid complexes, liquid crystals, glassy states, solid crystalline, amorphous & polymorphism.	10

  
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	<b>Physicochemical properties of drug molecules:</b> Refractive index, optical rotation, dielectric constant, dipole moment, dissociation constant, determinations and applications	
<b>III</b>	<b>Surface and interfacial phenomenon:</b> Liquid interface, surface & interfacial tensions, surface free energy, measurement of surface & interfacial tensions, spreading coefficient, adsorption at liquid interfaces, surface active agents, HLB Scale, solubilisation, detergency, adsorption at solid interface.	<b>08</b>
<b>IV</b>	<b>Complexation and protein binding:</b> Introduction, Classification of Complexation, Applications, methods of analysis, protein binding, Complexation and drug action, crystalline structures of complexes and thermodynamic treatment of stability constants	<b>08</b>
<b>V</b>	<b>pH, buffers and Isotonic solutions:</b> Sorensen's pH scale, pH determination (electrometric and calorimetric), applications of buffers, buffer equation, buffer capacity, buffers in pharmaceutical and biological systems, buffered isotonic solutions.	<b>07</b>

### Recommended Books: (Latest Editions)

1. Physical Pharmacy by Alfred Martin
2. Experimental Pharmaceutics by Eugene, Parott.
3. Tutorial Pharmacy by Cooper and Gunn.
4. Stocklosam J. Pharmaceutical Calculations, Lea & Febiger, Philadelphia.
5. Liberman H.A, Lachman C., Pharmaceutical Dosage forms, Tablets, Volume-1 to3, Marcel Dekkar Inc.
6. Liberman H.A, Lachman C, Pharmaceutical Dosage forms. Dispersesystems, volume 1, 2, 3. Marcel Dekkar Inc.
7. Physical Pharmaceutics by Ramasamy C and ManavalanR.
8. Laboratory Manual of Physical Pharmaceutics, C.V.S. Subramanyam, J.Thimma settee
9. Physical Pharmaceutics by C.V.S. Subramanyam
10. Test book of Physical Pharmacy, by Gaurav Jain & Roop K. Khar

  
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### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:


Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the principle involved in the formulation of dosage forms.
CO2	To understand the basic principles involved in various aspects of Physical Pharmaceutics.
CO3	Attain expertise in basic learning and application of Physical pharmaceutics in Pharmacy and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	-	-	-	3	2	2	3
CO2	3	-	2	2	-	-	-	-	-	-	3	2	2	3
CO3	3	-	3	2	-	-	2	-	-	1	3	2	2	3

  
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**BP 303 T. PHARMACEUTICAL MICROBIOLOGY (Theory)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours


**Scope:** Study of all categories of microorganisms especially for the production of alcoholantibiotics, vaccines, vitamins enzymes etc..

**Objectives:** Upon completion of the subject student shall be able to;

- Understand methods of identification, cultivation and preservation of various microorganisms
- To understand the importance and implementation of sterilization in pharmaceutical processing and industry
- Learn sterility testing of pharmaceutical products.
- Carried out microbiological standardization of Pharmaceuticals.
- Understand the cell culture technology and its applications in pharmaceutical industries.

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	Introduction, history of microbiology, its branches, scope and its importance. Introduction to Prokaryotes and Eukaryotes, Study of ultra-structure and morphological classification of bacteria, nutritional requirements, raw materials used for culture media and physical parameters for growth, growth curve, isolation and preservation methods for pure cultures, cultivation of anaerobes, quantitative measurement of bacterial growth (total & viable count). Study of different types of phase contrast microscopy, dark field microscopy and electron microscopy.	10
II	Identification of bacteria using staining techniques (simple, Gram's & Acid fast staining) and biochemical tests (IMViC). Study of principle, procedure, merits, demerits and applications of physical, chemical gaseous, radiation and mechanical method of sterilization. Evaluation of the efficiency of sterilization methods. Equipments employed in large scale sterilization. Sterility indicators.	10

  
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III	Study of morphology, classification, reproduction/replication and cultivation of Fungi and Viruses. Classification and mode of action of disinfectants. Factors influencing disinfection, antiseptics and their evaluation. For bacteriostatic and bactericidal actions. Evaluation of bactericidal & Bacteriostatic. Sterility testing of products (solids, liquids, ophthalmic and other sterile products) according to IP, BP and USP.	10
IV	Designing of aseptic area, laminar flow equipments; study of different sources of contamination in an aseptic area and methods of prevention, clean area classification. Principles and methods of different microbiological assay. Methods for standardization of antibiotics, vitamins and amino acids. Assessment of a new antibiotic.	08
V	Types of spoilage, factors affecting the microbial spoilage of pharmaceutical products, sources and types of microbial contaminants, assessment of microbial contamination and spoilage. Preservation of pharmaceutical products using antimicrobial agents, evaluation of microbial stability of formulations. Growth of animal cells in culture, general procedure for cell culture, Primary, established and transformed cell cultures. Application of cell cultures in pharmaceutical industry and research.	07

### Recommended Books

1. W.B. Hugo and A.D. Russel: Pharmaceutical Microbiology, Blackwell Scientific Publications, Oxford London.
2. Prescott and Dunn., Industrial Microbiology, 4th edition, CBS Publishers & Distributors, Delhi.
3. Pelczar, Chan Kreig, Microbiology, Tata McGraw Hill edn.
4. Malcolm Harris, Balliere Tindall and Cox: Pharmaceutical Microbiology.
5. Rose: Industrial Microbiology.
6. Probisher, Hinsdill et al: Fundamentals of Microbiology, 9th ed. Japan
7. Cooper and Gunn's: Tutorial Pharmacy, CBS Publisher and Distribution.
8. Pepler: Microbial Technology.
9. Ananthnarayan : Text Book of Microbiology, Orient-Longman, Chennai
10. Edward: Fundamentals of Microbiology.

  
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11. N.K. Jain: Pharmaceutical Microbiology, Vallabh Prakashan, Delhi

12. Bergeys manual of systematic bacteriology, Williams and Wilkins- A Waverly company

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To enlist the morphological characteristics of microorganisms and understand the applications of cell culture in the pharmaceutical industry and research.
CO2	To perform sterilization techniques, sterility testing, and standardization of pharmaceutical products and to employ types of culturing techniques for the cultivation, isolation, and identification of microorganisms in the laboratory
CO3	To attain expertise in basic learning and application of Microbiology in Pharmacy, Research, Hospital, and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	2	-	-	2	-	-	-	-	2	-	2	2	-	1
CO2	2	-	-	2	-	-	-	-	2	-	2	2	-	1
CO3	2	-	-	2	-	-	-	-	2	2	2	2	-	1

  
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**BP 304 T. PHARMACEUTICAL ENGINEERING (Theory)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours


**Scope:** This course is designed to impart a fundamental knowledge on the art and science of various unit operations used in pharmaceutical industry.

**Objectives:** Upon completion of the course student shall be able:

- To know various unit operations used in Pharmaceutical industries.
- To understand the material handling techniques.
- To perform various processes involved in pharmaceutical manufacturing process.
- To carry out various test to prevent environmental pollution.
- To appreciate and comprehend significance of plant lay out design for optimum use of resources.
- To appreciate the various preventive methods used for corrosion control in Pharmaceutical industries.

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<p><b>Flow of fluids:</b> Types of manometers, Reynolds number and its significance, Bernoulli's theorem and its applications, Energy losses, Orifice meter, Venturimeter, Pitot tube and Rotometer.</p> <p><b>Size Reduction:</b> Objectives, Mechanisms &amp; Laws governing size reduction, factors affecting size reduction, principles, construction, working, uses, merits and demerits of Hammer mill, ball mill, fluid energy mill, Edge runner mill &amp; endrunner mill.</p> <p><b>Size Separation:</b> Objectives, applications &amp; mechanism of size separation, official standards of powders, sieves, size separation Principles, construction, working, uses, merits and demerits of Sieve shaker, cyclone separator, Airseparator, Bag filter &amp; elutriation tank.</p>	10

  
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II	<p><b>Heat Transfer:</b> Objectives, applications &amp; Heat transfer mechanisms. Fourier's law, Heat transfer by conduction, convection &amp; radiation. Heat interchangers &amp; heat exchangers.</p> <p><b>Evaporation:</b> Objectives, applications and factors influencing evaporation, differences between evaporation and other heat process. principles, construction, working, uses, merits and demerits of Steam jacketed kettle, horizontal tube evaporator, climbing film evaporator, forced circulation evaporator, multiple effect evaporator &amp; Economy of multiple effect evaporator.</p> <p><b>Distillation:</b> Basic Principles and methodology of simple distillation, flash distillation, fractional distillation, distillation under reduced pressure, steam distillation &amp; molecular distillation</p>	10
III	<p><b>Drying:</b> Objectives, applications &amp; mechanism of drying process, measurements &amp; applications of Equilibrium Moisture content, rate of drying curve. principles, construction, working, uses, merits and demerits of Tray dryer, drum dryer spray dryer, fluidized bed dryer, vacuum dryer, freeze dryer.</p> <p><b>Mixing:</b> Objectives, applications &amp; factors affecting mixing, Difference between solid and liquid mixing, mechanism of solid mixing, liquids mixing and semisolids mixing. Principles, Construction, Working, uses, Merits and Demerits of Double cone blender, twin shell blender, ribbon blender, Sigma blade mixer, planetary mixers, Propellers, Turbines, Paddles &amp; Silverson Emulsifier</p>	08
IV	<p><b>Filtration:</b> Objectives, applications, Theories &amp; Factors influencing filtration, filter aids, filter medias. Principle, Construction, Working, Uses, Merits and demerits of plate &amp; frame filter, filter leaf, rotary drum filter, Meta filter &amp; Cartridge filter, membrane filters and Seidtz filter.</p> <p><b>Centrifugation:</b> Objectives, principle &amp; applications of Centrifugation, principles, construction, working, uses, merits and demerits of Perforated basket centrifuge, Non-perforated basket centrifuge, semi continuous centrifuge &amp; supercentrifuge.</p>	08
V	<p><b>Materials of pharmaceutical plant construction, Corrosion and its prevention:</b> Factors affecting during materials selected for Pharmaceutical plant construction, Theories of corrosion, types of corrosion and their prevention. Ferrous and nonferrous metals, inorganic and organic non metals, basic of material handling systems.</p>	07

  
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### Recommended Books:

1. Introduction to chemical engineering – Walter L Badger & Julius Banchero, Latest edition.
2. Solid phase extraction, Principles, techniques and applications by Nigel J.K. Simpson-Latest edition.
3. Unit operation of chemical engineering – McCabe Smith, Latest edition.
4. Pharmaceutical engineering principles and practices – C.V.S Subrahmanyam et al., Latest edition.
5. Remington practice of pharmacy- Martin, Latest edition.
6. Theory and practice of industrial pharmacy by Lachmann., Latest edition

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To explain the material handling techniques and to perform various processes involved in the pharmaceutical manufacturing process.
CO2	To appreciate and comprehend the significance of plant layout design for optimum use of resources.
CO3	Attain expertise in basic learning and application of Pharmaceutical Engineering in Pharmacy, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	2	2	2	2	-	-	-	-	-	-	2	-	2	-
CO2	2	2	2	2	-	-	-	-	-	-	2	-	2	-

  
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CO3	2	2	2	2	-	-	-	-	-	-	2	-	2	1
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## BP305P. PHARMACEUTICAL ORGANIC CHEMISTRY -II (Practical)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

### I Experiments involving laboratory techniques

- a) Recrystallization                      b) Steam distillation

### II Determination of following oil values (including standardization of reagents)


- a) Acid value                                  b) Saponification value c) Iodine value

### III Preparation of compounds

- Benzanilide/Phenyl benzoate/Acetanilide from Aniline/ Phenol /Aniline by acylation reaction.
- 2,4,6-Tribromo aniline/Para bromo acetanilide from Aniline/ Acetanilide by halogenation (Bromination) reaction.
- 5-Nitro salicylic acid/Meta di nitro benzene from Salicylic acid / Nitro benzene by nitration reaction.
- Benzoic acid from Benzyl chloride by oxidation reaction.
- Benzoic acid/ Salicylic acid from alkyl benzoate/ alkyl salicylate by hydrolysis reaction.
- 1-Phenyl azo-2-naphthol from Aniline by diazotization and coupling reactions.
- Benzil from Benzoin by oxidation reaction.
- Dibenzal acetone from Benzaldehyde by Claisen Schmidt reaction
- Cinnamic acid from Benzaldehyde by Perkin reaction
- *P*-Iodo benzoic acid from *P*-amino benzoic acid

### Recommended Books

1. Organic Chemistry by Morrison and Boyd

  
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2. Organic Chemistry by I.L. Finar , Volume-I
3. Textbook of Organic Chemistry by B.S. Bahl & Arun Bahl.
4. Organic Chemistry by P.L. Soni
5. Practical Organic Chemistry by Mann and Saunders.
6. Vogel's text book of Practical Organic Chemistry
7. Advanced Practical organic chemistry by N.K. Vishnoi.
8. Introduction to Organic Laboratory techniques by Pavia, Lampman and Kriz.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn the basics of organic chemistry. Identify organic compounds and elements via physic-chemical testing. To determine the physical properties of organic compounds.
CO2	To synthesize organic compounds and understand mechanisms and orientation of reactions. Demonstrate stereo modeling and its applications.
CO3	To analyze and apply the knowledge of Pharmaceutical Organic Chemistry in Pharmacy and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	-	-	-	-	1	2	1	3	1	-	3
CO2	3	-	2	-	-	-	-	1	2	1	3	1	-	3
CO3	3	2	3	2	-	-	-	-	2	1	3	1	-	3

  
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## BP306P. PHYSICAL PHARMACEUTICS – I (Practical)

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

1. Determination the solubility of drug at room temperature
2. Determination of pKa value by Half Neutralization/ Henderson Hasselbalchequation.
3. Determination of Partition co- efficient of benzoic acid in benzene and water
4. Determination of Partition co- efficient of Iodine in CCl<sub>4</sub> and water
5. Determination of % composition of NaCl in a solution using phenol-water system byCST method
6. Determination of surface tension of given liquids by drop count and drop weightmethod
7. Determination of HLB number of a surfactant by saponification method
8. Determination of Freundlich and Langmuir constants using activated char coal
9. Determination of critical micellar concentration of surfactants
10. Determination of stability constant and donor acceptor ratio of PABA-Caffeinecomplex by solubilitymethod
11. Determination of stability constant and donor acceptor ratio of Cupric-Glycinecomplex by pH titration method

### Recommended Books: (Latest Editions)

1. Physical Pharmacy by Alfred Martin
2. Experimental Pharmaceutics by Eugene, Parott.
3. Tutorial Pharmacy by Cooper and Gunn.
4. Stocklosam J. Pharmaceutical Calculations, Lea & Febiger, Philadelphia.
5. Liberman H.A, Lachman C., Pharmaceutical Dosage forms, Tablets, Volume-1 to3, Marcel Dekkar Inc.

  
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6. Liberman H.A, Lachman C, Pharmaceutical Dosage forms. Dispersesystems, volume 1, 2, 3. Marcel Dekkar Inc.
7. Physical Pharmaceutics by Ramasamy C and ManavalanR.
8. Laboratory Manual of Physical Pharmaceutics, C.V.S. Subramanyam, J.Thimma settee
9. Physical Pharmaceutics by C.V.S. Subramanyam
10. Test book of Physical Pharmacy, by Gaurav Jain & Roop K. Khar

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To describe the concept of complexation and protein binding of drugs.
CO2	To explain physicochemical properties of drugs including solubility, distribution, adsorption, and stability, and to provide knowledge on the importance of particle size and its distribution.
CO3	To analyze and apply the knowledge of Pharmaceutics in Pharmacy and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	-	-	-	3	1	2	3
CO2	3	-	2	2	-	-	-	-	-	-	3	1	2	3
CO3	3	-	3	3	-	-	-	-	-	-	3	-	3	3

  
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**BP 307P. PHARMACEUTICAL MICROBIOLOGY (Practical)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

1. Introduction and study of different equipments and processing, e.g., B.O.D. incubator, laminar flow, aseptic hood, autoclave, hot air sterilizer, deep freezer, refrigerator, microscopes used in experimental microbiology.
2. Sterilization of glassware, preparation and sterilization of media.
3. Sub culturing of bacteria and fungus. Nutrient stabs and slants preparations.
4. Staining methods- Simple, Grams staining and acid fast staining (Demonstration with practical).
5. Isolation of pure culture of micro-organisms by multiple streak plate technique and other techniques.
6. Microbiological assay of antibiotics by cup plate method and other methods
7. Motility determination by Hanging drop method.
8. Sterility testing of pharmaceuticals.
9. Bacteriological analysis of water
10. Biochemical test.

#### Recommended Books

1. W.B. Hugo and A.D. Russel: Pharmaceutical Microbiology, Blackwell Scientific Publications, Oxford London.
2. Prescott and Dunn., Industrial Microbiology, 4th edition, CBS Publishers & Distributors, Delhi.
3. Pelczar, Chan Kreig, Microbiology, Tata McGraw Hill edn.
4. Malcolm Harris, Balliere Tindall and Cox: Pharmaceutical Microbiology.
5. Rose: Industrial Microbiology.
6. Probisher, Hinsdill et al: Fundamentals of Microbiology, 9th ed. Japan
7. Cooper and Gunn's: Tutorial Pharmacy, CBS Publisher and Distribution.
8. Pepler: Microbial Technology.
9. Ananthnarayan : Text Book of Microbiology, Orient-Longman, Chennai

  
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10. Edward: Fundamentals of Microbiology.
11. N.K. Jain: Pharmaceutical Microbiology, Vallabh Prakashan, Delhi
12. Bergeys manual of systematic bacteriology, Williams and Wilkins- A Waverly company

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:


Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To enlist the morphological characteristics of microorganisms and understand the applications of cell culture in the pharmaceutical industry and research.
CO2	To perform sterilization techniques, sterility testing, and standardization of pharmaceutical products and to employ types of culturing techniques for the cultivation, isolation, and identification of microorganisms in the laboratory.
CO3	Attain expertise in basic learning and application of Microbiology in Pharmacy, Research, Hospital, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	2	-	-	2	-	-	-	-	2	-	2	2	-	1
CO2	2	-	-	2	-	-	-	-	2	-	2	2	-	1
CO3	2	-	-	2	-	-	-	-	2	2	2	2	-	1

  
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**BP308P - PHARMACEUTICAL ENGINEERING (Practical)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

1. Determination of radiation constant of brass, iron, unpainted and painted glass.
2. Steam distillation – To calculate the efficiency of steam distillation.
3. To determine the overall heat transfer coefficient by heat exchanger.
4. Construction of drying curves (for calcium carbonate and starch).
5. Determination of moisture content and loss on drying.
6. Determination of humidity of air – i) From wet and dry bulb temperatures – use of Dew point method.
7. Description of Construction working and application of Pharmaceutical Machinery such as rotary tablet machine, fluidized bed coater, fluid energy mill, de humidifier.
8. Size analysis by sieving – To evaluate size distribution of tablet granulations – Construction of various size frequency curves including arithmetic and logarithmic probability plots.
9. Size reduction: To verify the laws of size reduction using ball mill and determining Kicks, Rittinger's, Bond's coefficients, power requirement and critical speed of Ball Mill.
10. Demonstration of colloid mill, planetary mixer, fluidized bed dryer, freeze dryer and such other major equipment.
11. Factors affecting Rate of Filtration and Evaporation (Surface area, Concentration and Thickness/ viscosity)
12. To study the effect of time on the Rate of Crystallization.
13. To calculate the uniformity Index for given sample by using Double Cone Blender.

**Recommended Books:**

1. Introduction to chemical engineering – Walter L Badger & Julius Banchero, Latest edition.
2. Solid phase extraction, Principles, techniques and applications by Nigel J.K. Simpson-Latest edition.
3. Unit operation of chemical engineering – McCabe Smith, Latest edition.

  
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4. Pharmaceutical engineering principles and practices – C.V.S Subrahmanyam et al., Latest edition.
5. Remington practice of pharmacy- Martin, Latest edition.
6. Theory and practice of industrial pharmacy by Lachmann., Latest edition.
7. Physical pharmaceutics- C.V.S Subrahmanyam et al., Latest edition.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:


Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To explain the material handling techniques and to perform various processes involved in the pharmaceutical manufacturing process.
CO2	To appreciate and comprehend the significance of plant layout design for optimum use of resources.
CO3	Attain expertise in basic learning and application of Pharmaceutical Engineering in Pharmacy, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	2	2	2	2	-	-	-	-	-	-	2	-	2	-
CO2	2	2	2	2	-	-	-	-	-	-	2	-	2	-
CO3	2	2	2	2	-	-	-	-	-	-	2	-	2	1

  
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**2<sup>nd</sup> Year (Semester IV)**

**PCI Syllabus**

  
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## BP 401 T. PHARMACEUTICAL ORGANIC CHEMISTRY –III (Theory)

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject imparts knowledge on stereo-chemical aspects of organic compounds and organic reactions, important named reactions, chemistry of important hetero cyclic compounds. It also emphasizes on medicinal and other uses of organic compounds.

**Objectives:** At the end of the course, the student shall be able to

- understand the methods of preparation and properties of organic compounds
- explain the stereo chemical aspects of organic compounds and stereo chemical reactions
- know the medicinal uses and other applications of organic compounds

### Course Content:

**Note:** To emphasize on definition, types, mechanisms, examples, uses/applications

UNIT	CONTENT	No. of Hrs.
I	<b>Stereo isomerism:</b> Optical isomerism –Optical activity, enantiomerism, diastereoisomerism, meso compounds. Elements of symmetry, chiral and achiral molecules, DL system of nomenclature of optical isomers, sequence rules, RS system of nomenclature of optical isomers, Reactions of chiral molecules, Racemic modification and resolution of racemic mixture. Asymmetric synthesis: partial and absolute	10
II	<b>Geometrical isomerism:</b> Nomenclature of geometrical isomers (Cis Trans, EZ, Syn Anti systems), Methods of determination of configuration of geometrical isomers. Conformational isomerism in Ethane, n-Butane and Cyclohexane. Stereo isomerism in biphenyl compounds (Atropisomerism) and conditions for optical activity. Stereospecific and stereoselective reactions	10
III	<b>Heterocyclic compounds:</b> Nomenclature and classification, Synthesis, reactions and medicinal uses of following compounds/derivatives, Pyrrole, Furan, and Thiophene, Relative aromaticity and reactivity of Pyrrole, Furan and Thiophene	10

  
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IV	Synthesis, reactions and medicinal uses of following compounds/derivatives Pyrazole, Imidazole, Oxazole and Thiazole. Pyridine, Quinoline, Isoquinoline, Acridine and Indole. Basicity of pyridine. Synthesis and medicinal uses of Pyrimidine, Purine, azepines and their derivatives	08
V	<b>Reactions of synthetic importance:</b> Metal hydride reduction (NaBH <sub>4</sub> and LiAlH <sub>4</sub> ), Clemmensen reduction, Birch reduction, Wolff Kishner reduction. Oppenauer-oxidation and Dakin reaction. Beckmanns rearrangement and Schmidt rearrangement. Claisen-Schmidt condensation	07

### Recommended Books

1. Organic chemistry by I.L. Finar, Volume-I & II.
2. A text book of organic chemistry – Arun Bahl, B.S. Bahl.
3. Heterocyclic Chemistry by Raj K. Bansal
4. Organic Chemistry by Morrison and Boyd
5. Heterocyclic Chemistry by T.L. Gilchrist

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the chemistry of heterocyclic compounds and reaction of synthetic importance.
CO2	To understand the basics and importance of stereo and geometric isomerism.
CO3	Attain expertise in basic learning and application of Organic chemistry in Pharmacy, Research, and Industry.

  
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## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	1	2	-	3	1	1	3
CO2	3	-	2	2	-	-	-	1	2	-	3	1	1	3
CO3	3	2	3	2	-	-	2	1	2	1	3	1	1	3

  
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### BP402T. MEDICINAL CHEMISTRY – I (Theory)

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject is designed to impart fundamental knowledge on the structure, chemistry and therapeutic value of drugs. The subject emphasizes on structure activity relationships of drugs, importance of physicochemical properties and metabolism of drugs. The syllabus also emphasizes on chemical synthesis of important drugs under each class.

**Objectives:** Upon completion of the course the student shall be able to

- understand the chemistry of drugs with respect to their pharmacological activity
- understand the drug metabolic pathways, adverse effect and therapeutic value of drugs
- know the Structural Activity Relationship (SAR) of different class of drugs
- write the chemical synthesis of some drugs

#### Course Content:

**Note:** Study of the development of the following classes of drugs, Classification, mechanism of action, uses of drugs mentioned in the course, Structure activity relationship of selective class of drugs as specified in the course and synthesis of drugs superscripted (\*)

UNIT	CONTENT	No. of Hrs.
I	<b>Introduction to Medicinal Chemistry</b> <b>History and development of medicinal chemistry</b> <b>Physicochemical properties in relation to biological action</b> Ionization, Solubility, Partition Coefficient, Hydrogen bonding, Protein binding, Chelation, Bioisosterism, Optical and Geometrical isomerism. <b>Drug metabolism</b> Drug metabolism principles- Phase I and Phase II. Factors affecting drug metabolism including stereo chemical aspects.	10

  
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<b>II</b>	<b>Drugs acting on Autonomic Nervous System</b> <b>Adrenergic Neurotransmitters:</b> Biosynthesis and catabolism of catecholamine. Adrenergic receptors (Alpha & Beta) and their distribution. <b>Sympathomimetic agents: SAR of Sympathomimetic agents</b> Direct acting: Nor-epinephrine, Epinephrine, Phenylephrine*, Dopamine, Methyldopa, Clonidine, Dobutamine, Isoproterenol, Terbutaline, Salbutamol*, Bitolterol, Naphazoline, Oxymetazoline and Xylometazoline. Indirect acting agents: Hydroxyamphetamine, Pseudoephedrine, Propylhexedrine. Agents with mixed mechanism: Ephedrine, Metaraminol. <b>Adrenergic Antagonists:</b> <b>Alpha adrenergic blockers:</b> Tolazoline*, Phentolamine, Phenoxybenzamine, Prazosin, Dihydroergotamine, Methysergide. <b>Beta adrenergic blockers:</b> SAR of beta blockers, Propranolol*, Metibranolol, Atenolol, Betazolol, Bisoprolol, Esmolol, Metoprolol, Labetolol, Carvedilol.	<b>10</b>
<b>III</b>	<b>Cholinergic neurotransmitters:</b> Biosynthesis and catabolism of acetylcholine. Cholinergic receptors (Muscarinic & Nicotinic) and their distribution. <b>Parasympathomimetic agents: SAR of Parasympathomimetic agents</b> <b>Direct acting agents:</b> Acetylcholine, Carbachol*, Bethanechol, Methacholine, Pilocarpine. <b>Indirect acting/ Cholinesterase inhibitors (Reversible &amp; Irreversible):</b> Physostigmine, Neostigmine*, Pyridostigmine, Edrophonium chloride, Tacrine hydrochloride, Ambenonium chloride, Isoflurophate, Echothiophate iodide, Parathion, Malathion. <b>Cholinesterase reactivator:</b> Pralidoxime chloride. <b>Cholinergic Blocking agents: SAR of cholinolytic agents</b> <b>Solanaceous alkaloids and analogues:</b> Atropine sulphate, Hyoscyamine sulphate, Scopolamine hydrobromide, Homatropine hydrobromide, Ipratropium bromide*. <b>Synthetic cholinergic blocking agents:</b> Tropicamide,	<b>10</b>

  
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	Cyclopentolate hydrochloride, Clidinium bromide, Dicyclomine hydrochloride*, Glycopyrrolate, Methantheline bromide, Propantheline bromide, Benztropine mesylate, Orphenadrine citrate, Biperidine hydrochloride, Procyclidine hydrochloride*, Tridihexethyl chloride, Isopropamide iodide, Ethopropazine hydrochloride.	
IV	<b>Drugs acting on Central Nervous System</b> <b>A. Sedatives and Hypnotics:</b> <b>Benzodiazepines:</b> SAR of Benzodiazepines, Chlordiazepoxide, Diazepam*, Oxazepam, Chlorazepate, Lorazepam, Alprazolam, Zolpidem <b>Barbiturates:</b> SAR of barbiturates, Barbitol*, Phenobarbital, Mephobarbital, Amobarbital, Butobarbital, Pentobarbital, Secobarbital <b>Miscellaneous:</b> Amides & imides: Glutethimide. Alcohol & their carbamate derivatives: Meprobamate, Ethchlorvynol. Aldehyde & their derivatives: Triclofos sodium, Paraldehyde. <b>B. Antipsychotics</b> <b>Phenothiazines:</b> SAR of Phenothiazines - Promazine hydrochloride, Chlorpromazine hydrochloride*, Triflupromazine, Thioridazine hydrochloride, Piperacetazine hydrochloride, Prochlorperazine maleate, Trifluoperazine hydrochloride. <b>Ring Analogues of Phenothiazines:</b> Chlorprothixene, Thiothixene, Loxapine succinate, Clozapine. <b>Fluro buterophenones:</b> Haloperidol, Droperidol, Risperidone. <b>Beta amino ketones:</b> Molindone hydrochloride. <b>Benzamides:</b> Sulpieride. <b>C. Anticonvulsants:</b> SAR of Anticonvulsants, mechanism of anticonvulsant Action. <b>Barbiturates:</b> Phenobarbitone, Methabarbitol. <b>Hydantoins:</b> Phenytoin*, Mephenytoin, Ethotoin <b>Oxazolidine diones:</b> Trimethadione, Paramethadione <b>Succinimides:</b> Phensuximide, Methsuximide, Ethosuximide* <b>Urea and monoacylureas:</b> Phenacemide, Carbamazepine* <b>Benzodiazepines:</b> Clonazepam <b>Miscellaneous:</b> Primidone, Valproic acid, Gabapentin, Felbamate	08
V	<b>Drugs acting on Central Nervous System</b>	07

  
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<p><b>General anesthetics:</b></p> <p><b>Inhalation anesthetics:</b> Halothane*, Methoxyflurane, Enflurane, Sevoflurane, Isoflurane, Desflurane.</p> <p><b>Ultra short acting barbiturates:</b> Methohexital sodium*, Thiomyalsodium, Thiopental sodium. <b>Dissociative anesthetics:</b> Ketamine hydrochloride.*</p> <p><b>Narcotic and non-narcotic analgesics</b></p> <p><b>Morphine and related drugs:</b> SAR of Morphine analogues, Morphinesulphate, Codeine, Meperidine hydrochloride, Anilerdine hydrochloride, Diphenoxylate hydrochloride, Loperamide hydrochloride, Fentanyl citrate*, Methadone hydrochloride*, Propoxyphene hydrochloride, Pentazocine, Levorphanol tartarate.</p> <p><b>Narcotic antagonists:</b> Nalorphine hydrochloride, Levallorphan tartarate, Naloxone hydrochloride.</p> <p><b>Anti-inflammatory agents:</b> Sodium salicylate, Aspirin, Mefenamic acid*, Meclofenamate, Indomethacin, Sulindac, Tolmetin, Zomepirac, Diclofenac, Ketorolac, Ibuprofen*, Naproxen, Piroxicam, Phenacetin, Acetaminophen, Antipyrine, Phenylbutazone.</p>	
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### Recommended Books

1. Wilson and Giswold's Organic medicinal and Pharmaceutical Chemistry.
2. Foye's Principles of Medicinal Chemistry.
3. Burger's Medicinal Chemistry, Vol I to IV.
4. Introduction to principles of drug design- Smith and Williams.
5. Remington's Pharmaceutical Sciences.
6. Martindale's extra pharmacopoeia.
7. Organic Chemistry by I.L. Finar, Vol. II.
8. The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1-5.
9. Indian Pharmacopoeia.
10. Text book of practical organic chemistry- A.I. Vogel.

  
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### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basics of drug discovery, medicinal chemistry, drug nomenclature, etc.
CO2	To understand the basics and importance of synthesis of various drugs used for drugs describe in syllabus.
CO3	Attain expertise in basic learning and application of Medicinal Chemistry in Pharmacy, Research, and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	1	2	-	3	3	3	3

  
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**BP 403 T. PHYSICAL PHARMACEUTICS-II (Theory)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** The course deals with the various physical and physicochemical properties, and principles involved in dosage forms/formulations. Theory and practical components of the subject help the student to get a better insight into various areas of formulation research and development, and stability studies of pharmaceutical dosage forms.

**Objectives:** Upon the completion of the course student shall be able to

- Understand various physicochemical properties of drug molecules in the designing the dosage forms
- Know the principles of chemical kinetics & to use them for stability testing and determination of expiry date of formulations
- Demonstrate use of physicochemical properties in the formulation development and evaluation of dosage forms.

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Colloidal dispersions:</b> Classification of dispersed systems & their general characteristics, size & shapes of colloidal particles, classification of colloids & comparative account of their general properties. Optical, kinetic & electrical properties. Effect of electrolytes, coacervation, peptization & protective action	07
II	<b>Rheology:</b> Newtonian systems, law of flow, kinematic viscosity, effect of temperature, non-Newtonian systems, pseudoplastic, dilatant, plastic, thixotropy, thixotropy in formulation, determination of viscosity, capillary, falling Sphere, rotational viscometers. <b>Deformation of solids:</b> Plastic and elastic deformation, Heckel equation, Stress, Strain, Elastic Modulus	10

  
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III	<b>Coarse dispersion:</b> Suspension, interfacial properties of suspended particles, settling insuspensions, formulation of flocculated and deflocculated suspensions. Emulsions and theories of emulsification, microemulsion and multiple emulsions; Stability of emulsions, preservation of emulsions, rheological properties of emulsions and emulsion formulation by HLB method.	10
IV	<b>Micromeritics:</b> Particle size and distribution, mean particle size, number and weight distribution, particle number, methods for determining particle size by different methods, counting and separation method, particle shape, specific surface, methods for determining surface area, permeability, adsorption, derived properties of powders, porosity, packing arrangement, densities, bulkiness & flow properties.	10
V	<b>Drug stability:</b> Reaction kinetics: zero, pseudo-zero, first & second order, units of basic rate constants, determination of reaction order. Physical and chemical factors influencing the chemical degradation of pharmaceutical product: temperature, solvent, ionic strength, dielectric constant, specific & general acid base catalysis, Simple numerical problems. Stabilization of medicinal agents against common reactions like hydrolysis & oxidation. Accelerated stability testing in expiration dating of pharmaceutical dosage forms. Photolytic degradation and its prevention	10

### Recommended Books

1. Physical Pharmacy by Alfred Martin, Sixth edition
2. Experimental pharmaceutics by Eugene, Parott.
3. Tutorial pharmacy by Cooper and Gunn.
4. Stocklosam J. Pharmaceutical calculations, Lea & Febiger, Philadelphia.
5. Liberman H.A, Lachman C., Pharmaceutical Dosage forms, Tablets, Volume-1 to 3, Marcel Dekkar Inc.
6. Liberman H.A, Lachman C, Pharmaceutical dosage forms. Disperse systems, volume 1,2, 3. Marcel Dekkar Inc.
7. Physical Pharmaceutics by Ramasamy C, and Manavalan R.

  
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## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the principle, application, and basics of drug stability, coarse and colloidal dispersions, and.
CO2	To understand the principle, application, and basics of rheology and micromeritics.
CO3	Attain expertise in basic learning and application of Physical pharmaceutics in Pharmacy, Research, and Industry

## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	-	-	-	3	2	2	3
CO2	3	-	2	2	-	-	-	-	-	-	3	2	2	3
CO3	3	-	3	2	-	-	2	-	2	-	3	2	2	3

  
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### BP 404 T. PHARMACOLOGY-I (Theory)

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** The main purpose of the subject is to understand what drugs do to the living organisms and how their effects can be applied to therapeutics. The subject covers the information about the drugs like, mechanism of action, physiological and biochemical effects (pharmacodynamics) as well as absorption, distribution, metabolism and excretion (pharmacokinetics) along with the adverse effects, clinical uses, interactions, doses, contraindications and routes of administration of different classes of drugs.

**Objectives:** Upon completion of this course the student should be able to

- Understand the pharmacological actions of different categories of drugs
- Explain the mechanism of drug action at organ system/sub cellular/macromolecular levels.
- Apply the basic pharmacological knowledge in the prevention and treatment of various diseases.
- Observe the effect of drugs on animals by simulated experiments
- Appreciate correlation of pharmacology with other bio medical sciences

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>General Pharmacology</b> <b>Introduction to Pharmacology-</b> Definition, historical landmarks and scope of pharmacology, nature and source of drugs, essential drugs concept and routes of drug administration, Agonists, antagonists( competitive and non competitive), spare receptors, addiction, tolerance, dependence, tachyphylaxis, idiosyncrasy, allergy. <b>Pharmacokinetics-</b> Membrane transport, absorption, distribution, metabolism and excretion of drugs .Enzyme induction, enzyme inhibition, kinetics of elimination	08

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## Recommended Books

1. Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and Dale's Pharmacology, Churchill Livingstone Elsevier
2. Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical pharmacology, Tata McGrawHill
3. Goodman and Gilman's, The Pharmacological Basis of Therapeutics
4. Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin L.C., Joseph G. B., Wayne A.K., Bradley R.W., Applied Therapeutics, The Clinical use of Drugs, The Point Lippincott Williams & Wilkins
5. Mycek M.J, Gelnet S.B and Perper M.M. Lippincott's Illustrated Reviews-Pharmacology
6. K.D. Tripathi. Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi.
7. Sharma H. L., Sharma K. K., Principles of Pharmacology, Paras Medical Publisher

## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

	Theory		
Components	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basic principle and applications of Pharmacology in therapeutics and drug discovery.
CO2	To understand the adverse effects, therapeutic effects, and mechanisms of drug actions in various diseases and disorders mentioned in the syllabus.
CO3	To attain expertise in basic learning and application of Pharmacology in Pharmacy, Research, and Industry

## CO-PO Mapping

  
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Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
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CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	1	2	-	3	3	3	3

  
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## BP 405 T.PHARMACOGNOSY AND PHYTOCHEMISTRY I (Theory)

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** The subject involves the fundamentals of Pharmacognosy like scope, classification of crude drugs, their identification and evaluation, phytochemicals present in them and their medicinal properties.

**Objectives:** Upon completion of the course, the student shall be able to

- know the techniques in the cultivation and production of crude drugs
- know the crude drugs, their uses and chemical nature
- know the evaluation techniques for the herbal drugs
- carry out the microscopic and morphological evaluation of crude drugs

### Course Content

UNIT	CONTENT	No. of Hrs.
I	<p><b>Introduction to Pharmacognosy:</b></p> <p>(a) Definition, history, scope and development of Pharmacognosy</p> <p>(b) Sources of Drugs – Plants, Animals, Marine &amp; Tissue culture</p> <p>(c) Organized drugs, unorganized drugs (dried latex, dried juices, dried extracts, gums and mucilages, oleoresins and oleo-gum-resins).</p> <p><b>Classification of drugs:</b></p> <p>Alphabetical, morphological, taxonomical, chemical, pharmacological, chemo and serotaxonomical classification of drugs</p> <p><b>Quality control of Drugs of Natural Origin:</b></p> <p>Adulteration of drugs of natural origin. Evaluation by organoleptic, microscopic, physical, chemical and biological methods and properties.</p> <p>Quantitative microscopy of crude drugs including lycopodium spore method, leaf constants, camera lucida and diagrams of microscopic objects to scale with camera lucida.</p>	08


  
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<b>II</b>	<b>Cultivation, Collection, Processing and storage of drugs of natural origin:</b> Cultivation and Collection of drugs of natural origin, Factors influencing cultivation of medicinal plants. Plant hormones and their applications. Polyploidy, mutation and hybridization with reference to medicinal plants, Conservation of medicinal plants	<b>10</b>
<b>III</b>	<b>Plant tissue culture:</b> Historical development of plant tissue culture, types of cultures, Nutritional requirements, growth and their maintenance. Applications of plant tissue culture in pharmacognosy. Edible vaccines	<b>07</b>
<b>IV</b>	<b>Pharmacognosy in various systems of medicine:</b> Role of Pharmacognosy in allopathy and traditional systems of medicine namely, Ayurveda, Unani, Siddha, Homeopathy and Chinese systems of medicine. <b>Introduction to secondary metabolites:</b> Definition, classification, properties and test for identification of Alkaloids, Glycosides, Flavonoids, Tannins, Volatile oil and Resins	<b>10</b>
<b>V</b>	Study of biological source, chemical nature and uses of drugs of natural origin containing following drugs <b>Plant Products:</b> Fibers - Cotton, Jute, Hemp Hallucinogens, Teratogens, Natural allergens <b>Primary metabolites:</b> General introduction, detailed study with respect to chemistry, sources, preparation, evaluation, preservation, storage, therapeutic used and commercial utility as Pharmaceutical Aids and/or Medicines for the following Primary metabolites: <b>Carbohydrates:</b> Acacia, Agar, Tragacanth, Honey <b>Proteins and Enzymes :</b> Gelatin, casein, proteolytic enzymes (Papain, bromelain, serratiopeptidase, urokinase, streptokinase, pepsin). <b>Lipids (Waxes, fats, fixed oils) :</b> Castor oil, Chaulmoogra oil, Wool Fat, Bees Wax <b>Marine Drugs:</b> Novel medicinal agents from marine sources	<b>08</b>

  
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### Recommended Books

1. W.C.Evans, Trease and Evans Pharmacognosy, 16th edition, W.B. Saunders & Co., London, 2009.
2. Tyler, V.E., Brady, L.R. and Robbers, J.E., Pharmacognosy, 9th Edn., Lea and Febiger, Philadelphia, 1988.
3. Text Book of Pharmacognosy by T.E. Wallis
4. Mohammad Ali. Pharmacognosy and Phytochemistry, CBS Publishers & Distribution, New Delhi.
5. Text book of Pharmacognosy by C.K. Kokate, Purohit, Gokhlae (2007), 37th Edition, Nirali Prakashan, New Delhi.
6. Herbal drug industry by R.D. Choudhary (1996), Ist Edn, Eastern Publisher, New Delhi.
7. Essentials of Pharmacognosy, Dr. SH. Ansari, IInd edition, Birla publications, New Delhi, 2007.
8. Practical Pharmacognosy: C.K. Kokate, Purohit, Gokhlae
9. Anatomy of Crude Drugs by M.A. Iyengar.


### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

	Theory		
Components	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basic principle and applications of Pharmacognosy in therapeutics and drug discovery.
CO2	To learn the source, chemical constituents, and their utilization in the management of various disorders.
CO3	To attain expertise in basic learning and application of Pharmacognosy in Pharmacy, Research, and Industry.

  
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## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	1	2	-	3	3	3	3

  
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**BP406P. MEDICINAL CHEMISTRY – I (Practical)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

### I Preparation of drugs/ intermediates

- a) 1,3-pyrazole                      b) 1,3-oxazole                      c) Benzimidazole                      d) Benzotriazole  
e) 2,3- diphenyl quinoxaline                      f) Benzocaine                      g) Phenytoin                      h) Phenothiazine  
i) Barbiturate

II Assay of drugs: Chlorpromazine, Phenobarbitone, Atropine, Ibuprofen, Aspirin, Furosemide

III Determination of Partition coefficient for any two drugs

### Recommended Books

1. Wilson and Giswold's Organic medicinal and Pharmaceutical Chemistry.
2. Foye's Principles of Medicinal Chemistry.
3. Burger's Medicinal Chemistry, Vol I to IV.
4. Introduction to principles of drug design- Smith and Williams.
5. Remington's Pharmaceutical Sciences.
6. Martindale's extra pharmacopoeia.
7. Organic Chemistry by I.L. Finar, Vol. II.
8. The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1-5.
9. Indian Pharmacopoeia.
10. Text book of practical organic chemistry- A.I. Vogel.

  
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## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basics of drug discovery, medicinal chemistry, drug nomenclature, etc.
CO2	To understand the basics and synthesis of various drugs used for drugs describe in syllabus.
CO3	Attain expertise in basic learning and application of Medicinal Chemistry in Pharmacy, Research, and Industry.

## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	-	2	-	3	3	3	3

  
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**BP 407 P. PHYSICAL PHARMACEUTICS- II (Practical)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

1. Determination of particle size, particle size distribution using sieving method
2. Determination of particle size, particle size distribution using Microscopic method
3. Determination of bulk density, true density and porosity
4. Determine the angle of repose and influence of lubricant on angle of repose
5. Determination of viscosity of liquid using Ostwald's viscometer
6. Determination sedimentation volume with effect of different suspending agent
7. Determination sedimentation volume with effect of different concentration of single suspending agent
8. Determination of viscosity of semisolid by using Brookfield viscometer
9. Determination of reaction rate constant first order.
10. Determination of reaction rate constant second order
11. Accelerated stability studies

### Recommended Books: (Latest Editions)

1. Physical Pharmacy by Alfred Martin, Sixth edition
2. Experimental pharmaceuticals by Eugene, Parott.
3. Tutorial pharmacy by Cooper and Gunn.
4. Stocklosam J. Pharmaceutical calculations, Lea & Febiger, Philadelphia.
5. Liberman H.A, Lachman C., Pharmaceutical Dosage forms, Tablets, Volume-1 to 3, Marcel Dekkar Inc.
6. Liberman H.A, Lachman C, Pharmaceutical dosage forms. Disperse systems, volume 1,2, 3. Marcel Dekkar Inc.
7. Physical Pharmaceutics by Ramasamy C, and Manavalan R.

  
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## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To describe and demonstrate the concept of complexation and protein binding of drugs.
CO2	To demonstrate the physicochemical properties of drugs including solubility, distribution, adsorption, and stability, and to demonstrate knowledge of the importance of particle size and its distribution.
CO3	To analyze and apply the knowledge of Pharmaceutics in Pharmacy and Industry

## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	2	2	-	-	-	-	-	-	3	1	2	3
CO2	3	-	2	2	-	-	-	-	-	-	3	1	2	3
CO3	3	-	3	3	-	-	-	-	-	-	3	-	3	3

  
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**BP 408 P. PHARMACOLOGY-I (Practical)**

#### Teaching and Examination Scheme:


Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

*Note: All laboratory techniques and animal experiments are demonstrated by simulated experiments by softwares and videos*

1. Introduction to experimental pharmacology.
2. Commonly used instruments in experimental pharmacology.
3. Study of common laboratory animals.
4. Maintenance of laboratory animals as per CPCSEA guidelines.
5. Common laboratory techniques. Blood withdrawal, serum and plasma separation, anesthetics and euthanasia used for animal studies.
6. Study of different routes of drugs administration in mice/rats.
7. Study of effect of hepatic microsomal enzyme inducers on the phenobarbitone sleeping time in mice.
8. Effect of drugs on ciliary motility of frog oesophagus
9. Effect of drugs on rabbit eye.
10. Effects of skeletal muscle relaxants using rota-rod apparatus.
11. Effect of drugs on locomotor activity using actophotometer.
12. Anticonvulsant effect of drugs by MES and PTZ method.
13. Study of stereotype and anti-catatonic activity of drugs on rats/mice.
14. Study of anxiolytic activity of drugs using rats/mice.
15. Study of local anesthetics by different methods

#### Recommended Books

1. Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and Dale's Pharmacology, Churchill Livingstone Elsevier
2. Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical pharmacology, Tata McGrawHill

  
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3. Goodman and Gilman's, The Pharmacological Basis of Therapeutics
4. Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin L.C., Joseph G. B., Wayne A.K., Bradley R.W., Applied Therapeutics, The Clinical use of Drugs, The Point Lippincott Williams & Wilkins
5. Mycek M.J, Gelnet S.B and Perper M.M. Lippincott's Illustrated Reviews-Pharmacology
6. K.D. Tripathi. Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi.
7. Sharma H. L., Sharma K. K., Principles of Pharmacology, Paras Medical Publisher
8. Modern Pharmacology with clinical Applications, by Charles R.Craig & Robert,
9. Ghosh MN. Fundamentals of Experimental Pharmacology. Hilton & Company, Kolkata.
10. Kulkarni SK. Handbook of experimental pharmacology. Vallabh Prakashan.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn about the animals and instruments used in experimental pharmacology.
CO2	To demonstrate the effect of various drugs on the biological system by using computational techniques as described in the syllabus.
CO3	To analyze and apply the knowledge of Pharmacology in Pharmacy, Drug discovery, Research, and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	-	2	-	3	3	3	3

  
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## BP 408 P. PHARMACOGNOSY AND PHYTOCHEMISTRY I (Practical)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

1. Analysis of crude drugs by chemical tests: (i) Tragacanth (ii) Acacia (iii) Agar (iv) Gelatin (v) starch (vi) Honey (vii) Castor oil.
2. Determination of stomatal number and index
3. Determination of vein islet number, vein islet termination and palisade ratio.
4. Determination of size of starch grains, calcium oxalate crystals by eye piece micrometer
5. Determination of Fiber length and width
6. Determination of number of starch grains by Lycopodium spore method
7. Determination of Ash value
8. Determination of Extractive values of crude drugs
9. Determination of moisture content of crude drugs
10. Determination of swelling index and foaming

### Recommended Books

6. W.C. Evans, Trease and Evans Pharmacognosy, 16th edition, W.B. Saunders & Co., London, 2009.
7. Tyler, V.E., Brady, L.R. and Robbers, J.E., Pharmacognosy, 9th Edn., Lea and Febiger, Philadelphia, 1988.
8. Text Book of Pharmacognosy by T.E. Wallis
9. Mohammad Ali. Pharmacognosy and Phytochemistry, CBS Publishers & Distribution, New Delhi.
10. Text book of Pharmacognosy by C.K. Kokate, Purohit, Gokhale (2007), 37th Edition, Nirali Prakashan, New Delhi.

  
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10. Herbal drug industry by R.D. Choudhary (1996), Ist Edn, Eastern Publisher, New Delhi.
11. Essentials of Pharmacognosy, Dr. SH. Ansari, IInd edition, Birla publications, New Delhi, 2007.
12. Practical Pharmacognosy: C.K. Kokate, Purohit, Gokhlae
13. Anatomy of Crude Drugs by M.A. Iyengar.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn about the medicinal plants and instruments used in experimental Pharmacognosy.
CO2	To demonstrate the identification of various herbal drugs used in experimental Pharmacognosy.
CO3	To analyze and apply the knowledge of Pharmacognosy in Pharmacy, Drug discovery, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	1	2	-	3	3	3	3

  
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**3<sup>rd</sup> Year (Semester V)**

**PCI Syllabus**

  
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**BP 501 T. MEDICINAL CHEMISTRY – II (Theory)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject is designed to impart fundamental knowledge on the structure, chemistry and therapeutic value of drugs. The subject emphasizes on structure activity relationships of drugs, importance of physicochemical properties and metabolism of drugs. The syllabus also emphasizes on chemical synthesis of important drugs under each class.


**Objectives:** Upon completion of the course the student shall be able to

- Understand the chemistry of drugs with respect to their pharmacological activity
- Understand the drug metabolic pathways, adverse effect and therapeutic value of drugs
- Know the Structural Activity Relationship of different class of drugs
- Study the chemical synthesis of selected drugs

### COURSE CONTENT

**Note:** Study of the development of the following classes of drugs, Classification, mechanism of action, uses of drugs mentioned in the course, Structure activity relationship of selective class of drugs as specified in the course and synthesis of drugs superscripted (\*)

UNIT	CONTENT	No. of Hrs.
I	<b>Antihistaminic agents:</b> Histamine, receptors and their distribution in the human body. <b>H1-antagonists:</b> Diphenhydramine hydrochloride*, Dimenhydrinate, Doxylamines succinate, Clemastine fumarate, Diphenylpyraline hydrochloride, Tripelenamine hydrochloride, Chlorcyclizine hydrochloride, Meclizine hydrochloride, Buclizine hydrochloride, Chlorpheniramine maleate, Triprolidine hydrochloride*, Phenidamine tartarate, Promethazine hydrochloride*, Trimeprazine tartrate, Cyproheptadine hydrochloride, Azatidine maleate, Astemizole, Loratadine, Cetirizine, Levocetrazine Cromolyn sodium <b>H2-</b>	10

  
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	<p><b>antagonists:</b> Cimetidine*, Famotidine, Ranitidin.</p> <p><b>Gastric Proton pump inhibitors:</b> Omeprazole, Lansoprazole, Rabeprazole, Pantoprazole</p> <p><b>Anti-neoplastic agents:</b></p> <p><b>Alkylating agents:</b> Mecllorethamine*, Cyclophosphamide, Melphalan, Chlorambucil, Busulfan, Thiotepa</p> <p><b>Antimetabolites:</b> Mercaptopurine*, Thioguanine, Fluorouracil, Floxuridine, Cytarabine, Methotrexate*, Azathioprine</p> <p><b>Antibiotics:</b> Dactinomycin, Daunorubicin, Doxorubicin, Bleomycin</p> <p><b>Plant products:</b> Etoposide, Vinblastin sulphate, Vincristin sulphate</p> <p><b>Miscellaneous:</b> Cisplatin, Mitotane.</p>	
II	<p><b>Anti-anginal:</b></p> <p><b>Vasodilators:</b> Amyl nitrite, Nitroglycerin*, Pentaerythritol tetranitrate, Isosorbidedinitrite*, Dipyridamole.</p> <p><b>Calcium channel blockers:</b> Verapamil, Bepridil hydrochloride, Diltiazem hydrochloride, Nifedipine, Amlodipine, Felodipine, Nicardipine, Nimodipine.</p> <p><b>Diuretics:</b></p> <p>Carbonic anhydrase inhibitors: Acetazolamide*, Methazolamide, Dichlorphenamide. Thiazides: Chlorthiazide*, Hydrochlorothiazide, Hydroflumethiazide, Cyclothiazide. Loop diuretics: Furosemide*, Bumetanide, Ethacrynic acid. Potassium sparing Diuretics: Spironolactone, Triamterene, Amiloride. Osmotic Diuretics: Mannitol</p> <p><b>Anti-hypertensive Agents:</b> Timolol, Captopril, Lisinopril, Enalapril, Benazepril hydrochloride, Quinapril hydrochloride, Methyldopate hydrochloride, * Clonidine hydrochloride, Guanethidine monosulphate, Guanabenz acetate, Sodiumnitroprusside, Diazoxide, Minoxidil, Reserpine, Hydralazine hydrochloride.</p>	10
III	<p><b>Anti-arrhythmic Drugs:</b> Quinidine sulphate, Procainamide hydrochloride, Disopyramide phosphate*, Phenytoin sodium, Lidocaine hydrochloride, Tocainide hydrochloride, Mexiletine hydrochloride, Lorcaïnide hydrochloride,</p>	10

  
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	Amiodarone, Sotalol. <b>Anti-hyperlipidemic agents:</b> Clofibrate, Lovastatin, Cholesteramine and Cholestipol <b>Coagulant &amp; Anticoagulants:</b> Menadione, Acetomenadione, Warfarin*, Anisindione, clopidogrel <b>Drugs used in Congestive Heart Failure:</b> Digoxin, Digitoxin, Nesiritide, Bosentan, Tezosentan.	
<b>IV</b>	<b>Drugs acting on Endocrine system</b> Nomenclature, Stereochemistry and metabolism of steroids <b>Sex hormones:</b> Testosterone, Nandralone, Progestrones, Oestriol, Oestradiol, Oestrone, Diethyl stilbestrol. <b>Drugs for erectile dysfunction:</b> Sildenafil, Tadalafil. <b>Oral contraceptives:</b> Mifepristone, Norgestrel, Levonorgestrol <b>Corticosteroids:</b> Cortisone, Hydrocortisone, Prednisolone, Betamethasone, Dexamethasone <b>Thyroid and antithyroid drugs:</b> L-Thyroxine, L-Thyronine, Propylthiouracil, Methimazole.	<b>08</b>
<b>V</b>	<b>Antidiabetic agents:</b> Insulin and its preparations, Sulfonyl ureas: Tolbutamide*, Chlorpropamide, Glipizide, Glimepiride., Biguanides: Metformin., Thiazolidinediones: Pioglitazone, Rosiglitazone., Meglitinides: Repaglinide, Nateglinide. Glucosidase inhibitors: Acarbose, Voglibose. <b>Local Anesthetics:</b> SAR of Local anesthetics <b>Benzoic Acid derivatives:</b> Cocaine, Hexylcaine, Meprylcaine, Cyclomethycaine, Piperocaine. <b>Amino Benzoic acid derivatives:</b> Benzocaine*, Butamben, Procaine*, Butacaine, Propoxycaine, Tetracaine, Benoxinate. <b>Lidocaine/Anilide derivatives:</b> Lignocaine, Mepivacaine, Prilocaine, Etidocaine. <b>Miscellaneous:</b> Phenacaine, Dipiperodon, Dibucaine.*	<b>07</b>

## Recommended Books

1. Wilson and Giswold's Organic medicinal and Pharmaceutical Chemistry.
2. Foye's Principles of Medicinal Chemistry.

  
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3. Burger's Medicinal Chemistry, Vol I to IV.
4. Introduction to principles of drug design- Smith and Williams.
5. Remington's Pharmaceutical Sciences.
6. Martindale's extra pharmacopoeia.
7. Organic Chemistry by I.L. Finar, Vol. II.
8. The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1 to 5.
9. Indian Pharmacopoeia.
10. Text book of practical organic chemistry- A.I. Vogel.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:


Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the chemistry, SAR, synthesis, and adverse effects of drugs used for the management of CVS disorders.
CO2	To understand the chemistry, SAR, synthesis, and adverse effects of drugs used for the management of endocrine and respiratory disorders
CO3	Attain expertise in basic learning and application of Medicinal Chemistry in Pharmacy, Research, and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	1	2	-	3	3	3	3

  
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### BP 502 T. Industrial Pharmacy I (Theory)

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours


**Scope:** Course enables the student to understand and appreciate the influence of pharmaceutical additives and various pharmaceutical dosage forms on the performance of the drug product.

**Objectives:** Upon completion of the course the student shall be able to

- Know the various pharmaceutical dosage forms and their manufacturing techniques.
- Know various considerations in development of pharmaceutical dosage forms
- Formulate solid, liquid and semisolid dosage forms and evaluate them for their quality

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Preformulation Studies:</b> Introduction to preformulation, goals and objectives, study of physicochemical characteristics of drug substances. <b>a. Physical properties:</b> Physical form (crystal & amorphous), particle size, shape, flow properties, solubility profile (pKa, pH, partition coefficient), polymorphism <b>b. Chemical Properties:</b> Hydrolysis, oxidation, reduction, racemisation, polymerization, BCS classification of drugs & its significant, Application of preformulation considerations in the development of solid, liquid oral and parenteral dosage forms and its impact on stability of dosage forms.	07
II	<b>Tablets:</b> a. Introduction, ideal characteristics of tablets, classification of tablets. Excipients, Formulation of tablets, granulation methods, compression and processing problems. Equipments and tablet tooling. b. Tablet coating: Types of coating, coating materials, formulation of coating composition, methods of coating, equipment employed and defects in coating. c. Quality control tests: In process and finished product tests	10

  
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	<b>Liquid orals:</b> Formulation and manufacturing consideration of syrups and elixirs suspensions and emulsions; Filling and packaging; evaluation of liquid orals official in pharmacopoeia	
III	<b>Capsules:</b> a. <b>Hard gelatin capsules:</b> Introduction, Production of hard gelatin capsule shells. Size of capsules, Filling, finishing and special techniques of formulation of hard gelatin capsules, manufacturing defects. In process and final product quality control tests for capsules. b. <b>Soft gelatin capsules:</b> Nature of shell and capsule content, size of capsules, importance of base adsorption and minim/gram factors, production, in process and final product quality control tests. Packing, storage and stability testing of soft gelatin capsules and their applications. <b>Pellets:</b> Introduction, formulation requirements, pelletization process, equipments for manufacture of pellets	08
IV	<b>Parenteral Products:</b> a. Definition, types, advantages and limitations. Preformulation factors and essential requirements, vehicles, additives, importance of isotonicity b. Production procedure, production facilities and controls, aseptic processing c. Formulation of injections, sterile powders, large volume parenterals and lyophilized products. d. Containers and closures selection, filling and sealing of ampoules, vials and infusion fluids. Quality control tests of parenteral products. <b>Ophthalmic Preparations:</b> Introduction, formulation considerations; formulation of eye drops, eye ointments and eye lotions; methods of preparation; labeling, containers; evaluation of ophthalmic preparations	10
V	<b>Cosmetics:</b> Formulation and preparation of the following cosmetic preparations: lipsticks, shampoos, cold cream and vanishing cream, tooth pastes, hair dyes and sunscreens. <b>Pharmaceutical Aerosols:</b> Definition, propellants, containers, valves, types of aerosol systems; formulation and manufacture of aerosols; Evaluation of aerosols; Quality control and stability studies. <b>Packaging Materials Science:</b> Materials used for packaging of pharmaceutical	10

  
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	products, factors influencing choice of containers, legal and official requirements for containers, stability aspects of packaging materials, quality control tests.	
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## Recommended Books

1. Pharmaceutical dosage forms - Tablets, volume 1 -3 by H.A. Liberman, Leon Lachman & J.B. Schwartz
2. Pharmaceutical dosage form - Parenteral medication vol- 1&2 by Liberman & Lachman
3. Pharmaceutical dosage form disperse system VOL-1 by Liberman & Lachman
4. Modern Pharmaceutics by Gilbert S. Banker & C.T. Rhodes, 3rd Edition
5. Remington: The Science and Practice of Pharmacy, 20th edition Pharmaceutical Science (RPS)
6. Theory and Practice of Industrial Pharmacy by Liberman & Lachman
7. Pharmaceutics- The science of dosage form design by M.E. Aulton, Churchill livingstone, Latest edition
8. Introduction to Pharmaceutical Dosage Forms by H. C. Ansel, Lea & Febiger, Philadelphia, 5<sup>th</sup> edition, 2005
9. Drug stability - Principles and practice by Cartensen & C.J. Rhodes, 3rd Edition, Marcel Dekker Series, Vol 107.

## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

  
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**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basics and application of preformulation studies.
CO2	To understand the basics and application of various dosage forms such as tablets, capsules, cosmetics, and parenteral.
CO3	Attain expertise in basic learning and application of Industrial Pharmacy Research and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	-	1	2	-	3	3	3	3

  
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### BP503.T. PHARMACOLOGY-II (Theory)

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject is intended to impart the fundamental knowledge on various aspects(classification, mechanism of action, therapeutic effects, clinical uses, side effects andcontraindications) of drugs acting on different systems of body and in addition,emphasis on the basic concepts of bioassay.

**Objectives:** Upon completion of this course the student should be able to

- Understand the mechanism of drug action and its relevance in the treatment of different diseases
- Demonstrate isolation of different organs/tissues from the laboratory animals by simulated experiments
- Demonstrate the various receptor actions using isolated tissue preparation
- Appreciate correlation of pharmacology with related medical sciences

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
<b>I</b>	<b>Pharmacology of drugs acting on cardio vascular system</b> a. Introduction to hemodynamic and electrophysiology of heart. b. Drugs used in congestive heart failure c. Anti-hypertensive drugs. d. Anti-anginal drugs. e. Anti-arrhythmic drugs. f. Anti-hyperlipidemic drugs.	<b>10</b>
<b>II</b>	<b>Pharmacology of drugs acting on cardio vascular system</b> a. Drug used in the therapy of shock. b. Hematinics, coagulants and anticoagulants. c. Fibrinolytics and anti-platelet drugs	<b>10</b>

  
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	d. Plasma volume expanders <b>Pharmacology of drugs acting on urinary system</b> a. Diuretics b. Anti-diuretics.	
<b>III</b>	<b>Autocoids and related drugs</b> a. Introduction to autocoids and classification b. Histamine, 5-HT and their antagonists. c. Prostaglandins, Thromboxanes and Leukotrienes. d. Angiotensin, Bradykinin and Substance P. e. Non-steroidal anti-inflammatory agents f. Anti-gout drugs g. Antirheumatic drugs	<b>10</b>
<b>IV</b>	<b>Pharmacology of drugs acting on endocrine system</b> a. Basic concepts in endocrine pharmacology. b. Anterior Pituitary hormones- analogues and their inhibitors. c. Thyroid hormones- analogues and their inhibitors. d. Hormones regulating plasma calcium level- Parathormone, Calcitonin and Vitamin-D. d. Insulin, Oral Hypoglycemic agents and glucagon. e. ACTH and corticosteroids.	<b>08</b>
<b>V</b>	<b>Pharmacology of drugs acting on endocrine system</b> a. Androgens and Anabolic steroids. b. Estrogens, progesterone and oral contraceptives. c. Drugs acting on the uterus. <b>Bioassay</b> a. Principles and applications of bioassay. b. Types of bioassay c. Bioassay of insulin, oxytocin, vasopressin, ACTH, d-tubocurarine, digitalis, histamine and 5-HT	<b>07</b>

## Recommended Books

  
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1. Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and Dale's Pharmacology, Churchill Livingstone Elsevier
2. Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical pharmacology, Tata McGrawHill
3. Goodman and Gilman's, The Pharmacological Basis of Therapeutics
4. Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin L.C., Joseph G. B., Wayne A.K., Bradley R.W., Applied Therapeutics, The Clinical use of Drugs, The Point Lippincott Williams & Wilkins
5. Mycek M.J, Gelnet S.B and Perper M.M. Lippincott's Illustrated Reviews-Pharmacology
6. K.D. Tripathi. Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi.
7. Sharma H. L., Sharma K. K., Principles of Pharmacology, Paras Medical Publisher
8. Modern Pharmacology with clinical Applications, by Charles R.Craig & Robert,
9. Ghosh MN. Fundamentals of Experimental Pharmacology. Hilton & Company, Kolkata.
10. Kulkarni SK. Handbook of experimental pharmacology. Vallabh Prakashan.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the classification, mechanisms, adverse effects, etc. of the drugs acting on CVS.
CO2	To understand the classification, mechanisms, adverse effects, etc. of the autacoids and drugs acting on the endocrine system.
CO3	Attain expertise in basic learning and application of Pharmacology in Pharmacy, Research, Hospital, and Industry

  
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## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	1	2	-	3	3	3	3

  
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## BP 504 T. PHARMACOGNOSY AND PHYTOCHEMISTRY II (Theory)

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** The main purpose of subject is to impart the students the knowledge of how these secondary metabolites are produced in the crude drugs, how to isolate and identify and produce them industrially. Also this subject involves the study of producing the plants and phytochemicals through plant tissue culture, drug interactions and basic principles of traditional system of medicine

**Objectives:** Upon completion of the course, the student shall be able to:

- know the modern extraction techniques, characterization and identification of the herbal drugs and phytoconstituents
- understand the preparation and development of herbal formulation.
- understand the herbal drug interactions
- carry out isolation and identification of phytoconstituents

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Metabolic pathways in higher plants and their determination</b> a) Brief study of basic metabolic pathways and formation of different secondary metabolites through these pathways- Shikimic acid pathway, Acetate pathways and Amino acid pathway. b) Study of utilization of radioactive isotopes in the investigation of Biogenetic studies.	07
II	General introduction, composition, chemistry & chemical classes, biosources, therapeutic uses and commercial applications of following secondary metabolites: <b>Alkaloids:</b> Vinca, Rauwolfia, Belladonna, Opium, <b>Phenylpropanoids and Flavonoids:</b> Lignans, Tea, Ruta <b>Steroids, Cardiac Glycosides &amp; Triterpenoids:</b> Liquorice, Dioscorea, Digitalis	14

  
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	<b>Volatile oils:</b> Mentha, Clove, Cinnamon, Fennel, Coriander, <b>Tannins:</b> Catechu, Pterocarpus <b>Resins:</b> Benzoin, Guggul, Ginger, Asafoetida, Myrrh, Colophony <b>Glycosides:</b> Senna, Aloes, Bitter Almond <b>Iridoids, Other terpenoids &amp; Naphthaquinones:</b> Gentian, Artemisia, taxus, carotenoids	
<b>III</b>	Isolation, Identification and Analysis of Phytoconstituents a) Terpenoids: Menthol, Citral, Artemisin b) Glycosides: Glycyrrhetic acid & Rutin c) Alkaloids: Atropine, Quinine, Reserpine, Caffeine d) Resins: Podophyllotoxin, Curcumin	<b>06</b>
<b>IV</b>	Industrial production, estimation and utilization of the following phytoconstituents: Forskolin, Sennoside, Artemisinin, Diosgenin, Digoxin, Atropine, Podophyllotoxin, Caffeine, Taxol, Vincristine and Vinblastine	<b>10</b>
<b>V</b>	<b>Basics of Phytochemistry</b> Modern methods of extraction, application of latest techniques like Spectroscopy, chromatography and electrophoresis in the isolation, purification and identification of crude drugs.	<b>08</b>

### Recommended Books

1. W.C. Evans, Trease and Evans Pharmacognosy, 16th edition, W.B. Saunders & Co., London, 2009.
2. Mohammad Ali. Pharmacognosy and Phytochemistry, CBS Publishers & Distribution, New Delhi.
3. Text book of Pharmacognosy by C.K. Kokate, Purohit, Gokhale (2007), 37th Edition, Nirali Prakashan, New Delhi.
4. Herbal drug industry by R.D. Choudhary (1996), 1st Edn, Eastern Publisher, New Delhi.
5. Essentials of Pharmacognosy, Dr. SH. Ansari, 2nd edition, Birla publications, New Delhi, 2007
6. Herbal Cosmetics by H. Pande, Asia Pacific Business press, Inc, New Delhi.
7. A.N. Kalia, Textbook of Industrial Pharmacognosy, CBS Publishers, New Delhi, 2005.
8. R Endress, Plant cell Biotechnology, Springer-Verlag, Berlin, 1994.

  
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9. Pharmacognosy & Pharmacobiotechnology. James Bobbers, Marilyn KS, VE Tylor.
10. The formulation and preparation of cosmetic, fragrances and flavours.
11. Remington's Pharmaceutical sciences.
12. Text Book of Biotechnology by Vyas and Dixit.
13. Text Book of Biotechnology by R.C. Dubey.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basic principle and applications of various metabolic pathways and their determination.
CO2	To learn the source, isolation, identification, and medicinal importance of alkaloids, glycosides, etc. in the management of various disorders.
CO3	Attain expertise in basic learning and application of Pharmacognosy in Pharmacy, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	1	2	-	3	3	3	3

  
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**BP 505 T. PHARMACEUTICAL JURISPRUDENCE (Theory)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This course is designed to impart basic knowledge on important legislations related to the profession of pharmacy in India.

**Objectives:** Upon completion of the course, the student shall be able to understand:

- The Pharmaceutical legislations and their implications in the development and marketing of pharmaceuticals.
- Various Indian pharmaceutical Acts and Laws
- The regulatory authorities and agencies governing the manufacture and sale of pharmaceuticals
- The code of ethics during the pharmaceutical practice

**COURSE CONTENT**

UNIT	CONTENT	No. of Hrs.
I	<b>Drugs and Cosmetics Act, 1940 and its rules 1945:</b> Objectives, Definitions, Legal definitions of schedules to the Act and Rules Import of drugs – Classes of drugs and cosmetics prohibited from import, Import under license or permit. Offences and penalties. Manufacture of drugs – Prohibition of manufacture and sale of certain drugs, Conditions for grant of license and conditions of license for manufacture of drugs, Manufacture of drugs for test, examination and analysis, manufacture of new drug, loan license and repacking license.	10
II	<b>Drugs and Cosmetics Act, 1940 and its rules 1945.</b> Detailed study of Schedule G, H, M, N, P, T, U, V, X, Y, Part XII B, Sch F & DMR (OA) Sale of Drugs – Wholesale, Retail sale and Restricted license. Offences and penalties, Labeling & Packing of drugs- General labeling	10

  
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	requirements and specimen labels for drugs and cosmetics, List of permitted colors. Offences and penalties. Administration of the Act and Rules – Drugs Technical Advisory Board, Central drugs Laboratory, Drugs Consultative Committee, Government drug analysts, Licensing authorities, controlling authorities, Drugs Inspectors	
III	<b>Pharmacy Act –1948:</b> Objectives, Definitions, Pharmacy Council of India; its constitution and functions, Education Regulations, State and Joint state pharmacy councils; constitution and functions, Registration of Pharmacists, Offences and Penalties <b>Medicinal and Toilet Preparation Act –1955:</b> Objectives, Definitions, Licensing, Manufacture In bond and Outside bond, Export of alcoholic preparations, Manufacture of Ayurvedic, Homeopathic, Patent & Proprietary Preparations. Offences and Penalties. <b>Narcotic Drugs and Psychotropic substances Act-1985 and Rules:</b> Objectives, Definitions, Authorities and Officers, Constitution and Functions of narcotic & Psychotropic Consultative Committee, National Fund for Controlling the Drug Abuse, Prohibition, Control and Regulation, opium poppy cultivation and production of poppy straw, manufacture, sale and export of opium, Offences and Penalties	10
IV	<b>Study of Salient Features of Drugs and Magic Remedies Act and its rules:</b> Objectives, Definitions, Prohibition of certain advertisements, Classes of Exempted advertisements, Offences and Penalties <b>Prevention of Cruelty to animals Act-1960:</b> Objectives, Definitions, Institutional Animal Ethics Committee, CPCSEA guidelines for Breeding and Stocking of Animals, Performance of Experiments, Transfer and acquisition of animals for experiment, Records, Power to suspend or revoke registration, Offences and Penalties <b>National Pharmaceutical Pricing Authority:</b> Drugs Price Control Order (DPCO)-2013. Objectives, Definitions, Sale prices of bulk drugs, Retail price of formulations, Retail price and ceiling price of scheduled formulations, National List of Essential Medicines (NLEM)	08

  
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V	<b>Pharmaceutical Legislations – A brief review, Introduction, Study of drugs enquiry committee, Health survey and development committee, Hathi committee and Mudaliar committee</b> <b>Code of Pharmaceutical ethics</b> Definition, Pharmacist in relation to his job, trade, medical profession and his profession, Pharmacist's oath <b>Medical Termination of Pregnancy Act</b> <b>Right to Information Act</b> <b>Introduction to Intellectual Property Rights (IPR)</b>	07
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### Recommended books

1. Forensic Pharmacy by B. Suresh
2. Text book of Forensic Pharmacy by B.M. Mithal
3. Hand book of drug law-by M.L. Mehra
4. A text book of Forensic Pharmacy by N.K. Jain
5. Drugs and Cosmetics Act/Rules by Govt. of India publications.
6. Medicinal and Toilet preparations act 1955 by Govt. of India publications.
7. Narcotic drugs and psychotropic substances act by Govt. of India publications
8. Drugs and Magic Remedies act by Govt. of India publication
9. Bare Acts of the said laws published by Government. Reference books (Theory).

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To recognize the regulatory authorities and agencies governing the manufacture and sale of Pharmaceuticals and to understand the significance and relevance of pharmaceutical laws in India and the code of ethics in the pharmacy profession.
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CO2	To implement the applications of schedules in manufacturing, distribution, and sale of drugs and to impart basic knowledge on important legislations related to the profession of pharmacy in India.
CO3	Apply the knowledge of various drug licenses, registration certificates, and cancellation of licenses for the manufacturing, import, and export of pharmaceutical drugs

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	1	-	-	-	-	-	3	1	1	-	1	-	2	3
CO2	1	-	-	-	-	-	3	1	1	-	1	-	2	3
CO3	1	-	-	-	1	1	3	1	1	-	1	-	2	3

  
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**BP 506 P. INDUSTRIAL PHARMACY-I (PRACTICAL)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

1. Preformulation studies on paracetamol/asparin/or any other drug
2. Preparation and evaluation of Paracetamol tablets
3. Preparation and evaluation of Aspirin tablets
4. Coating of tablets- film coating of tables/granules
5. Preparation and evaluation of Tetracycline capsules
6. Preparation of Calcium Gluconate injection
7. Preparation of Ascorbic Acid injection
8. Quality control test of (as per IP) marketed tablets and capsules
9. Preparation of Eye drops/ and Eye ointments
10. Preparation of Creams (cold / vanishing cream)
11. Evaluation of Glass containers (as per IP)

### Recommended Books

1. Pharmaceutical dosage forms - Tablets, volume 1 -3 by H.A. Liberman, Leon Lachman & J.B. Schwartz
2. Pharmaceutical dosage form - Parenteral medication vol- 1&2 by Liberman &Lachman
3. Pharmaceutical dosage form disperse system VOL-1 by Liberman & Lachman
4. Modern Pharmaceutics by Gilbert S. Banker & C.T. Rhodes, 3rd Edition
5. Remington: The Science and Practice of Pharmacy, 20th edition Pharmaceutical Science (RPS)
6. Theory and Practice of Industrial Pharmacy by Liberman & Lachman
7. Pharmaceutics- The science of dosage form design by M.E.Aulton, Churchill livingstone, Latest edition
8. Introduction to Pharmaceutical Dosage Forms by H. C. Ansel, Lea & Febiger, Philadelphia, 5<sup>th</sup> edition, 2005

  
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9. Drug stability - Principles and practice by Cartensen & C.J. Rhodes, 3rd Edition, Marcel Dekker Series, Vol 107.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand and demonstrate preformulation studies.
CO2	To understand and demonstrate formulation of dosage forms such as tablets, capsules, cosmetics, and parenteral.
CO3	Attain expertise in basic learning and application of Industrial Pharmacy Research and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	-	1	2	-	3	3	3	3

  
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**BP 507 P. PHARMACOLOGY-II (Practical)**

### Teaching and Examination Scheme:


Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

*Note: All laboratory techniques and animal experiments are demonstrated by simulated experiments by softwares and videos*

1. Introduction to *in-vitro* pharmacology and physiological salt solutions.
2. Effect of drugs on isolated frog heart.
3. Effect of drugs on blood pressure and heart rate of dog.
4. Study of diuretic activity of drugs using rats/mice.
5. DRC of acetylcholine using frog rectus abdominis muscle.
6. Effect of physostigmine and atropine on DRC of acetylcholine using frog rectus abdominis muscle and rat ileum respectively.
7. Bioassay of histamine using guinea pig ileum by matching method.
8. Bioassay of oxytocin using rat uterine horn by interpolation method.
9. Bioassay of serotonin using rat fundus strip by three point bioassay.
10. Bioassay of acetylcholine using rat ileum/colon by four point bioassay.
11. Determination of PA<sub>2</sub> value of prazosin using rat anococcygeus muscle (by Schild's plot method).
12. Determination of PD<sub>2</sub> value using guinea pig ileum.
13. Effect of spasmogens and spasmolytics using rabbit jejunum.
14. Anti-inflammatory activity of drugs using carrageenan induced paw-edema model.
15. Analgesic activity of drug using central and peripheral methods

### Recommended Books

1. Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and Dale's Pharmacology, Churchill Livingstone Elsevier
2. Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical pharmacology, Tata McGrawHill
3. Goodman and Gilman's, The Pharmacological Basis of Therapeutics

  
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4. Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin L.C., Joseph G. B., Wayne A.K., Bradley R.W., Applied Therapeutics, The Clinical use of Drugs, The Point Lippincott Williams & Wilkins
5. Mycek M.J, Gelnet S.B and Perper M.M. Lippincott's Illustrated Reviews-Pharmacology
6. K.D. Tripathi. Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi.
7. Sharma H. L., Sharma K. K., Principles of Pharmacology, Paras Medical Publisher
8. Modern Pharmacology with clinical Applications, by Charles R.Craig & Robert,
9. Ghosh MN. Fundamentals of Experimental Pharmacology. Hilton & Company, Kolkata.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn about the animals and instruments used in <i>in-vitro</i> and <i>ex-vivo</i> experimental pharmacology.
CO2	To demonstrate the effect of various drugs on the biological system by using computational techniques as described in the syllabus.
CO3	To analyze and apply the knowledge of Pharmacology in Pharmacy, Drug discovery, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO3	3	-	3	3	-	-	-	-	2	-	3	3	3	3

  
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## BP 508 P. PHARMACOGNOSY AND PHYTOCHEMISTRY II (Practical)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

1. Morphology, histology and powder characteristics & extraction & detection of:  
Cinchona, Cinnamon, Senna, Clove, Ephedra, Fennel and Coriander
2. Exercise involving isolation & detection of active principles
  - a. Caffeine - from tea dust.
  - b. Diosgenin from Dioscorea
  - c. Atropine from Belladonnad. Sennosides from Senna
3. Separation of sugars by Paper chromatography
4. TLC of herbal extract
5. Distillation of volatile oils and detection of phytoconstitents by TLC
6. Analysis of crude drugs by chemical tests: (i) Asafoetida (ii) Benzoin (iii) Colophony (iv) Aloes (v) Myrrh

### Recommended Books

1. W.C.Evans, Trease and Evans Pharmacognosy, 16th edition, W.B. Saunders & Co., London, 2009.
2. Mohammad Ali. Pharmacognosy and Phytochemistry, CBS Publishers & Distribution, New Delhi.
3. Text book of Pharmacognosy by C.K. Kokate, Purohit, Gokhlae (2007), 37th Edition, Nirali Prakashan, New Delhi.
4. Herbal drug industry by R.D. Choudhary (1996), 1st Edn, Eastern Publisher, New Delhi.
5. Essentials of Pharmacognosy, Dr. SH. Ansari, 2nd edition, Birla publications, New Delhi, 2007
6. Herbal Cosmetics by H. Pande, Asia Pacific Business press, Inc, New Delhi.
7. A.N. Kalia, Textbook of Industrial Pharmacognosy, CBS Publishers, New Delhi, 2005.

  
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8. R Endress, Plant cell Biotechnology, Springer-Verlag, Berlin, 1994.
9. Pharmacognosy & Pharmacobiotechnology. James Bobbers, Marilyn KS, VE Tylor.
10. The formulation and preparation of cosmetic, fragrances and flavours.
11. Remington's Pharmaceutical sciences.
12. Text Book of Biotechnology by Vyas and Dixit.
13. Text Book of Biotechnology by R.C. Dubey.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basic principle and applications of various metabolic pathways and their determination.
CO2	To isolate, identify and learn the medicinal importance of alkaloids, glycosides, etc. in the management of various disorders.
CO3	Attain expertise in basic learning and application of Pharmacognosy in Pharmacy, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO3	3	-	3	3	-	-	-	-	2	-	3	3	3	3

  
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**3<sup>rd</sup> Year (Semester VI)**

**PCI Syllabus**

  
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**BP 601 T. MEDICINAL CHEMISTRY – III (Theory)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject is designed to impart fundamental knowledge on the structure, chemistry and therapeutic value of drugs. The subject emphasis on modern techniques of rational drug design like quantitative structure activity relationship (QSAR), Prodrug concept, combinatorial chemistry and Computer aided drug design (CADD). The subject also emphasizes on the chemistry, mechanism of action, metabolism, adverse effects, Structure Activity Relationships (SAR), therapeutic uses and synthesis of important drugs.

**Objectives:** Upon completion of the course student shall be able to

- Understand the importance of drug design and different techniques of drug design.
- Understand the chemistry of drugs with respect to their biological activity.
- Know the metabolism, adverse effects and therapeutic value of drugs.
- Know the importance of SAR of drugs.

### COURSE CONTENT

**Note:** Study of the development of the following classes of drugs, Classification, mechanism of action, uses of drugs mentioned in the course, Structure activity relationship of selective class of drugs as specified in the course and synthesis of drugs superscripted by (\*)

UNIT	CONTENT	No. of Hrs.
I	<b>Antibiotics</b> Historical background, Nomenclature, Stereochemistry, Structure activity relationship, Chemical degradation classification and important products of the following classes: <b>β-Lactam antibiotics:</b> Penicillin, Cephalosporins, β-Lactamase inhibitors, Monobactams <b>Aminoglycosides:</b> Streptomycin, Neomycin, Kanamycin <b>Tetracyclines:</b> Tetracycline, Oxytetracycline, Chlortetracycline, Minocycline,	10

  
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	Doxycycline	
II	<p><b>Antibiotics</b></p> <p>Historical background, Nomenclature, Stereochemistry, Structure activityrelationship, Chemical degradation classification and important products of the following classes.</p> <p><b>Macrolide:</b> Erythromycin Clarithromycin, Azithromycin.</p> <p><b>Miscellaneous:</b> Chloramphenicol*, Clindamycin.</p> <p><b>Prodrugs:</b> Basic concepts and application of prodrugs design.</p> <p><b>Antimalarials:</b> Etiology of malaria.</p> <p><b>Quinolines:</b> SAR, Quinine sulphate, Chloroquine*, Amodiaquine, Primaquine phosphate, Pamaquine*, Quinacrine hydrochloride, Mefloquine.</p> <p><b>Biguanides and dihydro triazines:</b> Cycloguanil pamoate, Proguanil.</p> <p><b>Miscellaneous:</b> Pyrimethamine, Artesunate, Artemether, Atovoquone.</p>	10
III	<p><b>Anti-tubercular Agents</b></p> <p><b>Synthetic anti tubercular agents:</b> Isoniazid*, Ethionamide, Ethambutol, Pyrazinamide, Para amino salicylic acid.*</p> <p><b>Anti tubercular antibiotics:</b> Rifampicin, Rifabutin, Cycloserine, Streptomycin, Capreomycin sulphate.</p> <p><b>Urinary tract anti-infective agents</b></p> <p><b>Quinolones:</b> SAR of quinolones, Nalidixic Acid, Norfloxacin, Enoxacin, Ciprofloxacin*, Ofloxacin, Lomefloxacin, Sparfloxacin, Gatifloxacin, Moxifloxacin</p> <p><b>Miscellaneous:</b> Furazolidine, Nitrofurantoin*, Methanamine.</p> <p><b>Antiviral agents:</b></p> <p>Amantadine hydrochloride, Rimantadine hydrochloride, Idoxuridine, trifluoride, Acyclovir*, Gancyclovir, Zidovudine, Didanosine, Zalcitabine, Lamivudine, Loviride, Delavirding, Ribavirin, Saquinavir, Indinavir, Ritonavir.</p>	10
IV	<p><b>Antifungal agents:</b></p> <p><b>Antifungal antibiotics:</b> Amphotericin-B, Nystatin, Natamycin, Griseofulvin.</p> <p><b>Synthetic Antifungal agents:</b> Clotrimazole, Econazole, Butoconazole, Oxiconazole, Tioconazole, Miconazole*, Ketoconazole,</p>	08

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	<p>Terconazole, Itraconazole, Fluconazole, Naftifine hydrochloride, Tolnaftate*.</p> <p><b>Anti-protozoal Agents:</b> Metronidazole*, Tinidazole, Ornidazole, Diloxanide, Iodoquinol, Pentamidine Isethionate, Atovaquone, Eflornithine.</p> <p><b>Anthelmintics:</b> Diethylcarbamazine citrate*, Thiabendazole, Mebendazole*, Albendazole, Niclosamide, Oxamniquine, Praziquantal, Ivermectin.</p> <p><b>Sulphonamides and Sulfones</b></p> <p>Historical development, chemistry, classification and SAR of Sulfonamides:</p> <p>Sulphamethizole, Sulfisoxazole, Sulphamethizine, Sulfacetamide*, Sulphapyridine, Sulfamethoxazole*, Sulphadiazine, Mefenide acetate, Sulfasalazine.</p> <p><b>Folate reductase inhibitors:</b> Trimethoprim*, Cotrimoxazole.</p> <p><b>Sulfones:</b> Dapsone*.</p>	
V	<p><b>Introduction to Drug Design</b></p> <p>Various approaches used in drug design. Physicochemical parameters used in quantitative structure activity relationship (QSAR) such as partition coefficient, Hammett's electronic parameter, Taft's steric parameter and Hansch analysis. Pharmacophore modeling and docking techniques.</p> <p><b>Combinatorial Chemistry:</b> Concept and applications chemistry: solid phase and solution phase synthesis of combinatorial</p>	07

### Recommended Books

1. Wilson and Giswold's Organic medicinal and Pharmaceutical Chemistry.
2. Foye's Principles of Medicinal Chemistry.
3. Burger's Medicinal Chemistry, Vol I to IV.
4. Introduction to principles of drug design- Smith and Williams.
5. Remington's Pharmaceutical Sciences.
6. Martindale's extra pharmacopoeia.
7. Organic Chemistry by I.L. Finar, Vol. II.
8. The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1 to 5.

  
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9. Indian Pharmacopoeia.
10. Text book of practical organic chemistry- A.I. Vogel.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the chemistry, SAR, synthesis, and adverse effects of Antibiotics
CO2	To understand the basics and learn the utilization of drug design, QSAR, and combinatorial chemistry
CO3	To attain expertise in basic learning and application of Medicinal Chemistry in Pharmacy, Research, and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	1	2	-	3	3	3	3

  
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### BP 602 T. PHARMACOLOGY-III (Theory)

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject is intended to impart the fundamental knowledge on various aspects(classification, mechanism of action, therapeutic effects, clinical uses, side effects andcontraindications) of drugs acting on respiratory and gastrointestinal system, infectiousdiseases, immuno-pharmacology and in addition,emphasis on the principles oftoxicology and chronopharmacology.

**Objectives:** Upon completion of this course the student should be able to:

- understand the mechanism of drug action and its relevance in the treatment ofdifferent infectious diseases
- comprehend the principles of toxicology and treatment of various poisoningsand
- appreciate correlation of pharmacology with related medical sciences.

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Pharmacology of drugs acting on Respiratory system</b> a. Anti -asthmatic drugs b. Drugs used in the management of COPD c. Expectorants and antitussives d. Nasal decongestants e. Respiratory stimulants <b>Pharmacology of drugs acting on the Gastrointestinal Tract</b> a. Antiulcer agents. b. Drugs for constipation and diarrhoea. c. Appetite stimulants and suppressants. d. Digestants and carminatives.	10

  
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	e. Emetics and anti-emetics.	
<b>II</b>	<b>Chemotherapy</b> a. General principles of chemotherapy. b. Sulfonamides and cotrimoxazole. c. Antibiotics- Penicillins, cephalosporins, chloramphenicol, macrolides, quinolones and fluoroquinolones, tetracycline and aminoglycosides	<b>10</b>
<b>III</b>	<b>Chemotherapy</b> a. Antitubercular agents b. Antileprotic agents c. Antifungal agents d. Antiviral drugs e. Anthelmintics f. Antimalarial drugs g. Antiamoebic agents	<b>10</b>
<b>IV</b>	<b>Chemotherapy:</b> Urinary tract infections and sexually transmitted diseases. Chemotherapy of malignancy <b>Immunopharmacology</b> a. Immunostimulants b. Immunosuppressant Protein drugs, monoclonal antibodies, target drugs to antigen, biosimilars	<b>08</b>
<b>V</b>	<b>Principles of toxicology</b> a. Definition and basic knowledge of acute, subacute and chronic toxicity. b. Definition and basic knowledge of genotoxicity, carcinogenicity, teratogenicity and mutagenicity c. General principles of treatment of poisoning d. Clinical symptoms and management of barbiturates, morphine, organophosphorus compound and lead, mercury and arsenic poisoning. <b>Chronopharmacology</b> a. Definition of rhythm and cycles. b. Biological clock and their significance leading to chronotherapy.	<b>07</b>

  
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## Recommended Books

1. Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and Dale's Pharmacology, Churchill Livingstone Elsevier
2. Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical pharmacology, Tata McGrawHill
3. Goodman and Gilman's, The Pharmacological Basis of Therapeutics
4. Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin L.C., Joseph G. B., Wayne A.K., Bradley R.W., Applied Therapeutics, The Clinical use of Drugs, The Point Lippincott Williams & Wilkins
5. Mycek M.J, Gelnet S.B and Perper M.M. Lippincott's Illustrated Reviews-Pharmacology
6. K.D. Tripathi. Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi.
7. Sharma H. L., Sharma K. K., Principles of Pharmacology, Paras Medical Publisher
8. Modern Pharmacology with clinical Applications, by Charles R. Craig & Robert,
9. Ghosh MN. Fundamentals of Experimental Pharmacology. Hilton & Company, Kolkata.
10. Kulkarni SK. Handbook of experimental pharmacology. Vallabh Prakashan.

## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the classification, mechanisms, adverse effects, etc. of the drugs acting on GIT and the respiratory system.
CO2	To understand the classification, mechanisms, adverse effects, etc. of chemotherapy.
CO3	Attain expertise in basic learning and application of Pharmacology in Pharmacy, Research, Hospital, and Industry

  
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## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	1	2	-	3	3	3	3

  
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**BP 603 T. HERBAL DRUG TECHNOLOGY (Theory)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours


**Scope:** This subject gives the student the knowledge of basic understanding of herbal drug industry, the quality of raw material, guidelines for quality of herbal drugs, herbal cosmetics, natural sweeteners, nutraceutical etc. The subject also emphasizes on Good Manufacturing Practices (GMP), patenting and regulatory issues of herbal drugs.

**Objectives:** Upon completion of this course the student should be able to:

- understand raw material as source of herbal drugs from cultivation to herbal drug product
- know the WHO and ICH guidelines for evaluation of herbal drugs
- know the herbal cosmetics, natural sweeteners, nutraceuticals
- appreciate patenting of herbal drugs, GMP.

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Herbs as raw materials</b> Definition of herb, herbal medicine, herbal medicinal product, herbal drug preparation, Source of Herbs, Selection, identification and authentication of herbal materials, Processing of herbal raw material <b>Biodynamic Agriculture</b> Good agricultural practices in cultivation of medicinal plants including Organic farming. Pest and Pest management in medicinal plants: Biopesticides/Bioinsecticides. <b>Indian Systems of Medicine</b> a) Basic principles involved in Ayurveda, Siddha, Unani and Homeopathy b) Preparation and standardization of Ayurvedic formulations viz Aristas and Asawas, Ghutika, Churna, Lehya and Bhasma.	11

  
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II	<b>Nutraceuticals</b> General aspects, Market, growth, scope and types of products available in the market. Healthbenefits and role of Nutraceuticals in ailments like Diabetes, CVS diseases, Cancer, Irritablebowel syndrome and various Gastro intestinal diseases. Study of following herbs as health food: Alfaalfa, Chicory, Ginger, Fenugreek, Garlic,Honey, Amla, Ginseng, Ashwagandha, Spirulina <b>Herbal-Drug and Herb-Food Interactions:</b> General introduction to interaction andclassification. Study of following drugs and their possible side effects and interactions:Hypercium, kava-kava, Ginkobiloba, Ginseng, Garlic, Pepper & Ephedra.	07
III	<b>Herbal Cosmetics</b> Sources and description of raw materials of herbal origin used via, fixed oils, waxes, gumscolors, perfumes, protective agents, bleaching agents, antioxidants in products such as skincare, hair care and oral hygiene products. <b>Herbal excipients:</b> Herbal Excipients – Significance of substances of natural origin as excipients – colorants,sweeteners, binders, diluents, viscosity builders, disintegrants, flavors & perfumes. <b>Herbal formulations :</b> Conventional herbal formulations like syrups, mixtures and tablets and Novel dosage formslike phytosomes	10
IV	<b>Evaluation of Drugs</b> WHO & ICH guidelines for the assessment of herbal drugs Stability testing of herbal drugs. <b>Patenting and Regulatory requirements of natural products:</b> a) Definition of the terms: Patent, IPR, Farmers right, Breeder's right, Bioprospecting andBiopiracy b) Patenting aspects of Traditional Knowledge and Natural Products. Case study of Curcuma& Neem. <b>Regulatory Issues</b> - Regulations in India (ASU DTAB, ASU DCC), Regulation ofmanufacture of ASU drugs - Schedule Z of Drugs & Cosmetics Act for ASU drugs.	10
V	<b>General Introduction to Herbal Industry</b>	07

  
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	<p>Herbal drugs industry: Present scope and future prospects.</p> <p>A brief account of plant based industries and institutions involved in work on medicinal and aromatic plants in India.</p> <p><b>Schedule T – Good Manufacturing Practice of Indian systems of medicine</b></p> <p>Components of GMP (Schedule – T) and its objectives. Infrastructural requirements, working space, storage area, machinery and equipments, standard operating procedures, health and hygiene, documentation and records.</p>	
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### Recommended Books

1. Textbook of Pharmacognosy by Trease & Evans.
2. Textbook of Pharmacognosy by Tyler, Brady & Robber.
3. Pharmacognosy by Kokate, Purohit and Gokhale
4. Essential of Pharmacognosy by Dr.S.H.Ansari
5. Pharmacognosy & Phytochemistry by V.D.Rangari
6. Pharmacopoeal standards for Ayurvedic Formulation (Council of Research in Indian Medicine & Homeopathy)
7. Mukherjee, P.W. Quality Control of Herbal Drugs: An Approach to Evaluation of Botanicals. Business Horizons Publishers, New Delhi, India, 2002.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To describe the history, scope, and future prospects of the herbal industry and to understand guidelines for GMP and evaluation of herbal drugs, herbal cosmetics, natural sweeteners, and nutraceuticals.
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CO2	To understand the health benefits and role of herbals and nutraceuticals in the treatment of ailments and to apply the knowledge of herbs, and raw materials in the formulation of herbal cosmetics and herbal drugs.
CO3	Attain expertise in basic learning and application of Herbal Drug Technology in Pharmacy, Research, and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	-	3	3	-	-	-	1	2	-	3	3	3	3

  
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## BP 604 T. BIOPHARMACEUTICS AND PHARMACOKINETICS (Theory)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject is designed to impart knowledge and skills of Biopharmaceutics and pharmacokinetics and their applications in pharmaceutical development, design of dose and dosage regimen and in solving the problems arising therein.

**Objectives:** Upon completion of the course student shall be able to:

- Understand the basic concepts in biopharmaceutics and pharmacokinetics and their significance.
- Use of plasma drug concentration-time data to calculate the pharmacokinetic parameters to describe the kinetics of drug absorption, distribution, metabolism, excretion, elimination.
- To understand the concepts of bioavailability and bioequivalence of drug products and their significance.
- Understand various pharmacokinetic parameters, their significance & applications.

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Introduction to Biopharmaceutics</b> <b>Absorption:</b> Mechanisms of drug absorption through GIT, factors influencing drug absorption through GIT, absorption of drug from Non per oral extra-vascular routes, <b>Distribution</b> Tissue permeability of drugs, binding of drugs, apparent, volume of drug distribution, plasma and tissue protein binding of drugs, factors affecting protein-drug binding. Kinetics of protein binding, Clinical significance of protein binding of drugs	10
II	<b>Elimination:</b> Drug metabolism and basic understanding metabolic pathways renal excretion of drugs, factors affecting renal excretion of drugs, renal clearance, Non renal routes of drug excretion of drugs	10

  
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	<b>Bioavailability and Bioequivalence:</b> Definition and Objectives of bioavailability, absolute and relative bioavailability, measurement of bioavailability, <i>in-vitro</i> drug dissolution models, <i>in-vitro-in-vivo</i> correlations, bioequivalence studies, methods to enhance the dissolution rates and bioavailability of poorly soluble drugs.	
<b>III</b>	<b>Pharmacokinetics:</b> Definition and introduction to Pharmacokinetics, Compartment models, Non compartment models, physiological models, One compartment open model. (a). Intravenous Injection (Bolus) (b). Intravenous infusion and (c) Extravascular administrations. Pharmacokinetics parameters - $KE$ , $t_{1/2}$ , $V_d$ , $AUC$ , $K_a$ , $Cl_t$ and $CLR$ - definitions methods of eliminations, understanding of their significance and application	<b>10</b>
<b>IV</b>	<b>Multicompartment models:</b> Two compartment open model. IV bolus Kinetics of multiple dosing, steady state drug levels, calculation of loading and maintenance doses and their significance in clinical settings.	<b>08</b>
<b>V</b>	<b>Nonlinear Pharmacokinetics:</b> a. Introduction, b. Factors causing Non-linearity. c. Michaelis-menton method of estimating parameters, Explanation with example of drugs.	<b>07</b>

### Recommended Books

1. Biopharmaceutics and Clinical Pharmacokinetics by, Milo Gibaldi.
2. Biopharmaceutics and Pharmacokinetics; By Robert F Notari
3. Applied biopharmaceutics and pharmacokinetics, Leon Shargel and Andrew B.C. YU 4<sup>th</sup> edition, Prentice-Hall International edition. USA
4. Bio pharmaceutics and Pharmacokinetics-A Treatise, By D. M. Brahmankar and Sunil B. Jaiswal, Vallabh Prakashan Pitampura, Delhi
5. Pharmacokinetics: By Milo Gibaldi Donald, R. Mercel Dekker Inc.
6. Hand Book of Clinical Pharmacokinetics, By Milo Gibaldi and Laurie Prescott by ADIS Health Science Press.
7. Biopharmaceutics; By Swarbrick
8. Clinical Pharmacokinetics, Concepts and Applications: By Malcolm Rowland and Thomas, N. Tozen, Lea and Febiger, Philadelphia, 1995.

  
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9. Dissolution, Bioavailability and Bioequivalence, By Abdou H.M, Mack, Publishing Company, Pennsylvania 1989.
10. Biopharmaceutics and Clinical Pharmacokinetics-An introduction 4th edition Revised and expanded by Rebert F Notari Marcel Dekker Inn, New York and Basel, 1987.
11. Remington's Pharmaceutical Sciences, ByMack Publishing Company, Pennsylvania

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basic principle and applications of biopharmaceutics and compartment modeling.
CO2	To understand the physiology of drug absorption, distribution, metabolism, and excretion.
CO3	Attain expertise in basic learning and application of biopharmaceutics in Pharmacy, Hospital, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	-	3	3	-	-	-	1	2	-	3	3	3	3

  
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**BP 605 T. PHARMACEUTICAL BIOTECHNOLOGY (Theory)**

#### Teaching and Examination Scheme:


Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** Biotechnology has a long promise to revolutionize the biological sciences and technology. Scientific application of biotechnology in the field of genetic engineering, medicine and fermentation technology makes the subject interesting. Biotechnology is leading to new biological revolutions in diagnosis, prevention and cure of diseases, new and cheaper pharmaceutical drugs. Biotechnology has already produced transgenic crops and animals and the future promises lot more. It is basically a research-based subject.

**Objectives:** Upon completion of the subject student shall be able to;

- Understanding the importance of Immobilized enzymes in Pharmaceutical Industries
- Genetic engineering applications in relation to production of pharmaceuticals
- Importance of Monoclonal antibodies in Industries
- Appreciate the use of microorganisms in fermentation technology

UNIT	CONTENT	No. of Hrs.
I	a) Brief introduction to Biotechnology with reference to Pharmaceutical Sciences. b) Enzyme Biotechnology- Methods of enzyme immobilization and applications. c) Biosensors- Working and applications of biosensors in Pharmaceutical Industries. d) Brief introduction to Protein Engineering. e) Use of microbes in industry. Production of Enzymes- General consideration - Amylase, Catalase, Peroxidase, Lipase, Protease, Penicillinase. f) Basic principles of genetic engineering.	10
II	a) Study of cloning vectors, restriction endonucleases and DNA ligase. b) Recombinant DNA technology. Application of genetic engineering in medicine. c) Application of r DNA technology and genetic engineering in the production of: i) Interferon ii) Vaccines- hepatitis- B iii) Hormones- Insulin. d) Brief introduction to PCR	10

  
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<b>III</b>	Types of immunity- humoral immunity, cellular immunity a) Structure of Immunoglobulins b) Structure and Function of MHC c) Hypersensitivity reactions, Immune stimulation and Immune suppressions. d) General method of the preparation of bacterial vaccines, toxoids, viral vaccine, antitoxins, serum-immune blood derivatives and other products relative to immunity. e) Storage conditions and stability of official vaccines f) Hybridoma technology- Production, Purification and Applications g) Blood products and Plasma Substitutes.	<b>10</b>
<b>IV</b>	a) Immuno blotting techniques- ELISA, Western blotting, Southern blotting. b) Genetic organization of Eukaryotes and Prokaryotes c) Microbial genetics including transformation, transduction, conjugation, plasmids and transposons. d) Introduction to Microbial biotransformation and applications. e) Mutation: Types of mutation/mutants.	<b>08</b>
<b>V</b>	a) Fermentation methods and general requirements, study of media, equipments, sterilization methods, aeration process, stirring. b) Large scale production fermenter design and its various controls. c) Study of the production of - penicillins, citric acid, Vitamin B12, Glutamic acid, Griseofulvin, d) Blood Products: Collection, Processing and Storage of whole human blood, dried human plasma, plasma Substitutes.	<b>07</b>

### Recommended Books

1. B.R. Glick and J.J. Pasternak: Molecular Biotechnology: Principles and Applications of Recombinant DNA: ASM Press Washington D.C.
2. RA Goldshy et. al.,: Kuby Immunology.
3. J.W. Goding: Monoclonal Antibodies.
4. J.M. Walker and E.B. Gingold: Molecular Biology and Biotechnology by Royal Society of Chemistry.
5. Zaborsky: Immobilized Enzymes, CRC Press, Degrand, Ohio.
6. S.B. Primrose: Molecular Biotechnology (Second Edition) Blackwell Scientific Publication.

  
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7. Stanbury F., P., Whitakar A., and Hall J., S., Principles of fermentation technology, 2<sup>nd</sup> edition, Aditya books Ltd., New Delhi

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:


Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the concept of immunity, its comparative characteristics, and various antigen-antibody reactions along with diagnosis, prevention, and treatment of disease with pharmaceutical drug products by recombinant technology.
CO2	To apply scientific applications of biotechnology in the field of genetic engineering and production of enzymes and to illustrate various diagnostic techniques for the identification of disease.
CO3	To attain expertise in basic learning and application of Pharmaceutical Biotechnology in Pharmacy, Hospital, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	1	-	1	3	-	-	-	-	-	-	2	1	-	-
CO2	1	-	1	3	-	-	-	-	-	-	2	1	-	-
CO3	1	-	1	3	-	-	-	-	2	-	2	1	-	-

  
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**BP 606 T. PHARMACEUTICAL QUALITY ASSURANCE (Theory)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This course deals with the various aspects of quality control and quality assurance aspects of pharmaceutical industries. It deals with the important aspects like cGMP, QC tests, documentation, quality certifications and regulatory affairs.

**Objectives:** Upon completion of the course student shall be able to:

- understand the cGMP aspects in a pharmaceutical industry
- appreciate the importance of documentation
- understand the scope of quality certifications applicable to pharmaceutical industries
- understand the responsibilities of QA & QC departments

**COURSE CONTENT**

UNIT	CONTENT	No. of Hrs.
I	<b>Quality Assurance and Quality Management concepts:</b> Definition and concept of Quality control, Quality assurance and GMP <b>Total Quality Management (TQM):</b> Definition, elements, philosophies <b>ICH Guidelines:</b> purpose, participants, process of harmonization, Brief overview of QSEM, with special emphasis on Q-series guidelines, ICH stability testing guidelines <b>Quality by design (QbD):</b> Definition, overview, elements of QbD program, tools <b>ISO 9000 &amp; ISO14000:</b> Overview, Benefits, Elements, steps for registration <b>NABL accreditation :</b> Principles and procedures	10
II	<b>Organization and personnel:</b> Personnel responsibilities, training, hygiene and personal records. <b>Premises:</b> Design, construction and plant layout, maintenance, sanitation, environmental control, utilities and maintenance of sterile areas, control of contamination.	10

  
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	<b>Equipments and raw materials:</b> Equipment selection, purchase specifications, maintenance, purchase specifications and maintenance of stores for raw materials.	
<b>III</b>	<b>Quality Control:</b> Quality control test for containers, rubber closures and secondary packing materials. <b>Good Laboratory Practices:</b> General Provisions, Organization and Personnel, Facilities, Equipment, Testing Facilities Operation, Test and Control Articles, Protocol for Conduct of a Nonclinical Laboratory Study, Records and Reports, Disqualification of Testing Facilities	<b>10</b>
<b>IV</b>	<b>Complaints:</b> Complaints and evaluation of complaints, Handling of return good, recalling and waste disposal. <b>Document maintenance in pharmaceutical industry:</b> Batch Formula Record, Master Formula Record, SOP, Quality audit, Quality Review and Quality documentation, Reports and documents, distribution records.	<b>08</b>
<b>V</b>	<b>Calibration and Validation:</b> Introduction, definition and general principles of calibration, qualification and validation, importance and scope of validation, types of validation, validation master plan. Calibration of pH meter, Qualification of UV-Visible spectrophotometer, General principles of Analytical method Validation. <b>Warehousing:</b> Good warehousing practice, materials management	<b>07</b>

#### Recommended Books

1. Quality Assurance Guide by organization of Pharmaceutical Products of India.
2. Good Laboratory Practice Regulations, 2nd Edition, Sandy Weinberg Vol. 69.
3. Quality Assurance of Pharmaceuticals- A compendium of Guide lines and Related materials Vol I WHO Publications.
4. A guide to Total Quality Management- Kushik Maitra and Sedhan K Ghosh
5. How to Practice GMP's – P P Sharma.
6. ISO 9000 and Total Quality Management – Sadhank G Ghosh
7. The International Pharmacopoeia – Vol I, II, III, IV- General Methods of Analysis and Quality specification for Pharmaceutical Substances, Excipients and Dosage forms
8. Good laboratory Practices – Marcel Deckker Series
9. ICH guidelines, ISO 9000 and 14000 guidelines

  
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### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn the concept of quality, quality control, quality assurance, TQM, QbD, QMS, and documentation.
CO2	To Understand the basic practices and concepts of cGMP, GLP, QC, QA, documentation, and validations in context to regulatory aspects.
CO3	Attain expertise in basic learning and application of Pharmaceutical Quality Assurance in Pharmacy, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	2	-	-	-	-	-	-	-	-	-	2	-	-	3
CO2	2	-	-	-	-	-	-	-	-	-	2	-	-	3
CO3	2	-	1	3	-	-	-	-	-	-	2	-	-	3

  
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**BP 607 P. MEDICINAL CHEMISTRY- III (Practical)**

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

#### I Preparation of drugs and intermediates

Sulphanilamide, 7-Hydroxy, 4-methyl coumarin, Chlorobutanol, Triphenyl imidazole,  
Tolbutamide, Hexamine

#### II Assay of drugs

Isonicotinic acid hydrazide, Chloroquine, Metronidazole, Dapsone, Chlorpheniramine maleate  
Benzyl penicillin

III Preparation of medicinally important compounds or intermediates by Microwave irradiation  
technique

IV Drawing structures and reactions using chem draw®

V Determination of physicochemical properties such as logP, clogP, MR, Molecular weight,  
Hydrogen bond donors and acceptors for class of drugs course content using drug design  
software Drug likeness screening (Lipinski's RO5)

### Recommended Books

1. Wilson and Giswold's Organic medicinal and Pharmaceutical Chemistry.
2. Foye's Principles of Medicinal Chemistry.
3. Burger's Medicinal Chemistry, Vol I to IV.
4. Introduction to principles of drug design- Smith and Williams.
5. Remington's Pharmaceutical Sciences.
6. Martindale's extra pharmacopoeia.
7. Organic Chemistry by I.L. Finar, Vol. II.
8. The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1 to 5.
9. Indian Pharmacopoeia.

  
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10. Text book of practical organic chemistry- A.I. Vogel.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basics of drug discovery, medicinal chemistry, drug nomenclature, etc.
CO2	To understand the basics and synthesis of various drugs used for drugs describe in syllabus.
CO3	Attain expertise in basic learning and application of Medicinal Chemistry in Pharmacy, Research, and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO3	3	2	3	3	-	-	2	-	2	-	3	3	3	3

  
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### BP 608 P. PHARMACOLOGY-III (Practical)

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

\*Experiments are demonstrated by simulated experiments/videos

1. Dose calculation in pharmacological experiments
2. Antiallergic activity by mast cell stabilization assay
3. Study of anti-ulcer activity of a drug using pylorus ligand (SHAY) rat model and NSAIDS induced ulcer model.
4. Study of effect of drugs on gastrointestinal motility
5. Effect of agonist and antagonists on guinea pig ileum
6. Estimation of serum biochemical parameters by using semi- autoanalyser
7. Effect of saline purgative on frog intestine
8. Insulin hypoglycemic effect in rabbit
9. Test for pyrogens ( rabbit method)
10. Determination of acute oral toxicity (LD50) of a drug from a given data
11. Determination of acute skin irritation / corrosion of a test substance
12. Determination of acute eye irritation / corrosion of a test substance
13. Calculation of pharmacokinetic parameters from a given data
14. Biostatistics methods in experimental pharmacology( student's t test, ANOVA)
15. Biostatistics methods in experimental pharmacology (Chi square test, Wilcoxon Signed Rank test)

#### Recommended Books

1. Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and Dale's Pharmacology, Churchill Livingstone Elsevier
2. Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical pharmacology, Tata McGrawHill
3. Goodman and Gilman's, The Pharmacological Basis of Therapeutics

  
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4. Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin L.C., Joseph G. B., Wayne A.K., Bradley R.W., Applied Therapeutics, The Clinical use of Drugs, The Point Lippincott Williams & Wilkins
5. Mycek M.J, Gelnet S.B and Perper M.M. Lippincott's Illustrated Reviews-Pharmacology
6. K.D. Tripathi. Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi.
7. Sharma H. L., Sharma K. K., Principles of Pharmacology, Paras Medical Publisher
8. Modern Pharmacology with clinical Applications, by Charles R.Craig & Robert,

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn about the animals and instruments used in <i>in-vitro</i> and <i>ex-vivo</i> experimental pharmacology.
CO2	To demonstrate the effect of various drugs on the biological system by using computational techniques as described in the syllabus.
CO3	To analyze and apply the knowledge of Pharmacology in Pharmacy, Drug discovery, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO2	3	-	3	3	-	-	-	-	2	-	3	3	3	3
CO3	3	-	3	3	-	-	-	-	2	-	3	3	3	3

  
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## BP 609 P. HERBAL DRUG TECHNOLOGY (Practical)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

1. To perform preliminary phytochemical screening of crude drugs.
2. Determination of the alcohol content of Asava and Arista
3. Evaluation of excipients of natural origin
4. Incorporation of prepared and standardized extract in cosmetic formulations like creams, lotions and shampoos and their evaluation.
5. Incorporation of prepared and standardized extract in formulations like syrups, mixtures and tablets and their evaluation as per Pharmacopoeial requirements.
6. Monograph analysis of herbal drugs from recent Pharmacopoeias
7. Determination of Aldehyde content
8. Determination of Phenol content
9. Determination of total alkaloids

### Recommended Books

1. Textbook of Pharmacognosy by Trease & Evans.
2. Textbook of Pharmacognosy by Tyler, Brady & Robber.
3. Pharmacognosy by Kokate, Purohit and Gokhale
4. Essential of Pharmacognosy by Dr.S.H.Ansari
5. Pharmacognosy & Phytochemistry by V.D.Rangari
6. Pharmacopoeal standards for Ayurvedic Formulation (Council of Research in Indian Medicine & Homeopathy)
7. Mukherjee, P.W. Quality Control of Herbal Drugs: An Approach to Evaluation of Botanicals. Business Horizons Publishers, New Delhi, India, 2002.

  
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### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To demonstrate the preparation, identification, and characteristics of various herbal products.
CO2	To prepare and standardize herbal products.
CO3	Attain expertise in basic learning and application of Herbal Drug Technology in Pharmacy, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	-	3	-	-	-	-	-	-	2	-	-	3
CO2	3	-	-	3	-	-	-	-	-	-	2	-	-	3
CO3	3	-	-	3	-	-	-	-	-	-	2	-	-	3

  
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**4<sup>th</sup> Year (Semester VII)**

**PCI Syllabus**

  
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**BP701T. INSTRUMENTAL METHODS OF ANALYSIS (Theory)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject deals with the application of instrumental methods in qualitative and quantitative analysis of drugs. This subject is designed to impart a fundamental knowledge on the principles and instrumentation of spectroscopic and chromatographic technique. This also emphasizes on theoretical and practical knowledge on modern analytical instruments that are used for drug testing.

**Objectives:** Upon completion of the course the student shall be able to

- Understand the interaction of matter with electromagnetic radiations and its applications in drug analysis
- Understand the chromatographic separation and analysis of drugs.
- Perform quantitative & qualitative analysis of drugs using various analytical instruments.

**COURSE CONTENT**

UNIT	CONTENT	No. of Hrs.
I	<b>UV Visible spectroscopy</b> Electronic transitions, chromophores, auxochromes, spectral shifts, solvent effect on absorption spectra, Beer and Lambert's law, Derivation and deviations. Instrumentation - Sources of radiation, wavelength selectors, sample cells, detectors-Photo tube, Photomultiplier tube, Photo voltaic cell, Silicon Photodiode. Applications - Spectrophotometric titrations, Single component and multi component analysis <b>Fluorimetry</b> Theory, Concepts of singlet, doublet and triplet electronic states, internal and external conversions, factors affecting fluorescence, quenching, instrumentation and applications	10
II	<b>IR spectroscopy:</b> Introduction, fundamental modes of vibrations in poly atomic	10

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	<p>molecules, sample handling, factors affecting vibrations, Instrumentation - Sources of radiation, wavelength selectors, detectors - Golay cell, Bolometer, Thermocouple, Thermister, Pyroelectric detector and applications</p> <p><b>Flame Photometry</b>-Principle, interferences, instrumentation and applications</p> <p><b>Atomic absorption spectroscopy</b>- Principle, interferences, instrumentation and applications</p> <p><b>Nepheloturbidometry</b>- Principle, instrumentation and applications</p>	
<b>III</b>	<p><b>Introduction to chromatography</b></p> <p><b>Adsorption and partition column chromatography</b>-Methodology, advantages, disadvantages and applications.</p> <p><b>Thin layer chromatography</b>- Introduction, Principle, Methodology, Rf values, advantages, disadvantages and applications.</p> <p><b>Paper chromatography</b>-Introduction, methodology, development techniques, advantages, disadvantages and applications</p> <p><b>Electrophoresis</b>- Introduction, factors affecting electrophoretic mobility, Techniques of paper, gel, capillary electrophoresis, applications</p>	<b>10</b>
<b>IV</b>	<p><b>Gas chromatography</b> - Introduction, theory, instrumentation, derivatization, temperature programming, advantages, disadvantages and applications</p> <p><b>High performance liquid chromatography (HPLC)</b>-Introduction, theory, instrumentation, advantages and applications.</p>	<b>08</b>
<b>V</b>	<p><b>Ion exchange chromatography</b>- Introduction, classification, ion exchange resins, properties, mechanism of ion exchange process, factors affecting ion exchange, methodology and applications</p> <p><b>Gel chromatography</b>- Introduction, theory, instrumentation and applications</p> <p><b>Affinity chromatography</b>- Introduction, theory, instrumentation and applications</p>	<b>07</b>

### Recommended Books

1. Instrumental Methods of Chemical Analysis by B.K Sharma
2. Organic spectroscopy by Y.R Sharma

  
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- Text book of Pharmaceutical Analysis by Kenneth A. Connors
- Vogel's Text book of Quantitative Chemical Analysis by A.I. Vogel
- Practical Pharmaceutical Chemistry by A.H. Beckett and J.B. Stenlake
- Organic Chemistry by I. L. Finar
- Organic spectroscopy by William Kemp
- Quantitative Analysis of Drugs by D. C. Garrett
- Quantitative Analysis of Drugs in Pharmaceutical Formulations by P. D. Sethi
- Spectrophotometric identification of Organic Compounds by Silverstein

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn the operation of various analytical instruments used in the analysis of drugs and pharmaceuticals and to perform estimation of pharmaceuticals by analytical techniques.
CO2	To understand the basics and applications of chromatographic techniques, NMR, IR, and MS Spectroscopies
CO3	Attain expertise in basic learning and application of Instrumental Methods of Analysis in Pharmacy, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	1	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	1	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	1	3	3	-	-	-	1	2	-	3	3	3	3

  
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**BP 702 T. INDUSTRIAL PHARMACY II (Theory)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This course is designed to impart fundamental knowledge on pharmaceutical product development and translation from laboratory to market

**Objectives:** Upon completion of the course, the student shall be able to:

- Know the process of pilot plant and scale up of pharmaceutical dosage forms
- Understand the process of technology transfer from lab scale to commercial batch
- Know different Laws and Acts that regulate pharmaceutical industry
- Understand the approval process and regulatory requirements for drug products

**Course Content:**

UNIT	CONTENT	No. of Hrs.
I	<b>Pilot plant scale up techniques:</b> General considerations - including significance of personnel requirements, space requirements, raw materials, Pilot plant scale up considerations for solids, liquid orals, semi solids and relevant documentation, SUPAC guidelines, Introduction to platform technology	10
II	<b>Technology development and transfer:</b> WHO guidelines for Technology Transfer (TT): Terminology, Technology transfer protocol, Quality risk management, Transfer from R& D to production (Process, packaging and cleaning), Granularity of TT Process (API, excipients, finished products, packaging materials) Documentation, Premises and equipments, qualification and validation, quality control, analytical method transfer, Approved regulatory bodies and agencies, Commercialization - practical aspects and problems (case studies), TT agencies in India - APCTD, NRDC, TIFAC, BCIL, TBSE /SIDBI; TT related documentation - confidentiality agreement, licensing, MoUs, legal issues	10
III	<b>Regulatory affairs:</b> Introduction, Historical overview of Regulatory Affairs,	10

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	Regulatory authorities, Role of Regulatory affairs department, Responsibility of Regulatory Affairs Professionals <b>Regulatory requirements for drug approval:</b> Drug Development Teams, Non-Clinical Drug Development, Pharmacology, Drug Metabolism and Toxicology, General considerations of Investigational New Drug (IND) Application, Investigator's Brochure (IB) and New Drug Application (NDA), Clinical research / BE studies, Clinical Research Protocols, Biostatistics in Pharmaceutical Product Development, Data Presentation for FDA Submissions, Management of Clinical Studies.	
IV	<b>Quality management systems:</b> Quality management & Certifications: Concept of Quality, Total Quality Management, Quality by Design (QbD), Six Sigma concept, Out of Specifications (OOS), Change control, Introduction to ISO 9000 series of quality systems standards, ISO 14000, NABL, GLP	08
V	<b>Indian Regulatory Requirements:</b> Central Drug Standard Control Organization (CDSCO) and State Licensing Authority: Organization, Responsibilities, Certificate of Pharmaceutical Product (COPP), Regulatory requirements and approval procedures for New Drugs.	07

### Recommended Books:

1. Regulatory Affairs from Wikipedia, the free encyclopedia modified on 7th April available at [http://en.wikipedia.org/wiki/Regulatory\\_Affairs](http://en.wikipedia.org/wiki/Regulatory_Affairs).
2. International Regulatory Affairs Updates, 2005. available at <http://www.iraup.com/about.php>
3. Douglas J Pisano and David S. Mantus. Text book of FDA Regulatory Affairs A Guide for Prescription Drugs, Medical Devices, and Biologics' Second Edition.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

  
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**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basics and application of pilot plant scale-up techniques with special mention to technology development and transfer.
CO2	To understand the drug regularity affairs, with special reference to Indian regulatory requirements, requirements for drug approvals, and quality management system.
CO3	Attain expertise in basic learning and application of Industrial Pharmacy Research and Industry

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	3	-	-	-	2	1	-	3	3	-	3
CO2	3	-	3	3	-	-	-	2	1	-	3	3	-	3
CO3	3	2	3	3	-	-	-	2	1	-	3	3	-	3

  
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### BP 703T. PHARMACY PRACTICE (Theory)

#### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** In the changing scenario of pharmacy practice in India, for successful practice of Hospital Pharmacy, the students are required to learn various skills like drug distribution, drug information, and therapeutic drug monitoring for improved patient care. In community pharmacy, students will be learning various skills such as dispensing of drugs, responding to minor ailments by providing suitable safe medication, patient counselling for improved patient care in the community set up.

**Objectives:** Upon completion of the course, the student shall be able to

- know various drug distribution methods in a hospital
- appreciate the pharmacy stores management and inventory control
- monitor drug therapy of patient through medication chart review and clinical review
- obtain medication history interview and counsel the patients
- identify drug related problems
- detect and assess adverse drug reactions
- interpret selected laboratory results (as monitoring parameters in therapeutics) of specific disease states
- know pharmaceutical care services
- do patient counseling in community pharmacy;
- appreciate the concept of Rational drug therapy.

UNIT	CONTENT	No. of Hrs.
1	<p><b>a) Hospital and its organization:</b> Definition, Classification of hospital- Primary, Secondary and Tertiary hospitals, Classification based on clinical and non- clinical basis, Organization Structure of a Hospital, and Medical staffs involved in the hospital and their functions.</p> <p><b>b) Hospital pharmacy and its organization:</b> Definition, functions of hospital pharmacy, Organization structure, Location, Layout and staff requirements, and</p>	10

  
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	<p>Responsibilities and functions of hospital pharmacists.</p> <p><b>c) Adverse drug reaction:</b> Classifications - Excessive pharmacological effects, secondary pharmacological effects, idiosyncrasy, allergic drug reactions, genetically determined toxicity, toxicity following sudden withdrawal of drugs, Drug interaction- beneficial interactions, adverse interactions, and pharmacokinetic drug interactions, Methods for detecting drug interactions, spontaneous case reports and record linkage studies, and Adverse drug reaction reporting and management.</p> <p><b>d) Community Pharmacy:</b> Organization and structure of retail and wholesale drug store, types and design, Legal requirements for establishment and maintenance of a drug store, Dispensing of proprietary products, maintenance of records of retail and wholesale drug store.</p>	
II	<p><b>a) Drug distribution system in a hospital:</b> Dispensing of drugs to inpatients, types of drug distribution systems, charging policy and labelling, Dispensing of drugs to ambulatory patients, and Dispensing of controlled drugs.</p> <p><b>b) Hospital formulary:</b> Definition, contents of hospital formulary, Differentiation of hospital formulary and Drug list, preparation and revision, and addition and deletion of drug from hospital formulary.</p> <p><b>c) Therapeutic drug monitoring:</b> Need for Therapeutic Drug Monitoring, Factors to be considered during the Therapeutic Drug Monitoring, and Indian scenario for Therapeutic Drug Monitoring.</p> <p><b>d) Medication adherence:</b> Causes of medication non-adherence, pharmacist role in the medication adherence, and monitoring of patient medication adherence.</p> <p><b>e) Patient medication history interview:</b> Need for the patient medication history interview, medication interview forms.</p> <p><b>f) Community pharmacy management:</b> Financial, materials, staff, and infrastructure requirements.</p>	10
III	<p><b>a) Pharmacy and therapeutic committee:</b> Organization, functions, Policies of the pharmacy and therapeutic committee in including drugs into formulary, inpatient and outpatient prescription, automatic stop order, and emergency drug list preparation.</p>	10

  
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	<p><b>b) Drug information services:</b> Drug and Poison information centre, Sources of drug information, Computerised services, and storage and retrieval of information.</p> <p><b>c) Patient counseling:</b> Definition of patient counseling; steps involved in patient counseling, and Special cases that require the pharmacist</p> <p><b>d) Education and training program in the hospital:</b> Role of pharmacist in the education and training program, Internal and external training program, Services to the nursing homes/clinics, Code of ethics for community pharmacy, and Role of pharmacist in the interdepartmental communication and community health education.</p> <p><b>e) Prescribed medication order and communication skills:</b> Prescribed medication order- interpretation and legal requirements, and Communication skills- communication with prescribers and patients.</p>	
IV	<p><b>a) Budget preparation and implementation:</b> Budget preparation and implementation</p> <p><b>b) Clinical Pharmacy:</b> Introduction to Clinical Pharmacy, Concept of clinical pharmacy, functions and responsibilities of clinical pharmacist, Drug therapy monitoring - medication chart review, clinical review, pharmacist intervention, Ward round participation, Medication history and Pharmaceutical care. Dosing pattern and drug therapy based on Pharmacokinetic &amp; disease pattern.</p> <p><b>c) Over the counter (OTC) sales:</b> Introduction and sale of over the counter, and Rational use of common over the counter medications.</p>	08
V	<p><b>a) Drug store management and inventory control:</b> Organisation of drug store, types of materials stocked and storage conditions, Purchase and inventory control: principles, purchase procedure, purchase order, procurement and stocking, Economic order quantity, Reorder quantity level, and Methods used for the analysis of the drug expenditure</p> <p><b>b) Investigational use of drugs:</b> Description, principles involved, classification, control, identification, role of hospital pharmacist, advisory committee.</p> <p><b>c) Interpretation of Clinical Laboratory Tests:</b> Blood chemistry, hematology,</p>	07

  
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	and urinalysis	
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## Recommended Books

1. Merchant S.H. and Dr. J.S.Quadry. A textbook of hospital pharmacy, 4th ed. Ahmadabad: B.S. Shah Prakakshan; 2001.
2. Parthasarathi G, Karin Nyfort-Hansen, Milap C Nahata. A textbook of Clinical Pharmacy, Practice- essential concepts and skills, 1st ed. Chennai: OrientLongman Private Limited; 2004.
3. William E. Hassan. Hospital pharmacy, 5th ed. Philadelphia: Lea & Febiger; 1986.
4. Tipnis Bajaj. Hospital Pharmacy, 1st ed. Maharashtra: Career Publications; 2008.
5. Scott LT. Basic skills in interpreting laboratory data, 4th ed. American Society of Health System Pharmacists Inc; 2009.
6. Parmar N.S. Health Education and Community Pharmacy, 18th ed. India: CBS Publishers & Distributers; 2008.

## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand and learn terminologies, techniques, and concepts in hospital and community pharmacy
CO2	To classify types of adverse drug reactions and drug interactions and to understand drug distribution, drug information, and therapeutic drug monitoring.
CO3	Attain expertise in basic learning and application of Pharmacy practice in Research and Hospitals.

  
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## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	1	3	1	-	2	1	1	3	-	3	-	2	1
CO2	3	1	3	1	-	2	1	1	3	-	3	-	2	1
CO3	3	1	3	1	-	2	1	1	3	-	3	-	2	1

  
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**BP 704 T: NOVEL DRUG DELIVERY SYSTEMS (Theory)**

**Teaching and Examination Scheme:**

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject is designed to impart basic knowledge on the area of novel drug delivery systems.

**Objectives:** Upon completion of the course student shall be able

- To understand various approaches for development of novel drug delivery systems.
- To understand the criteria for selection of drugs and polymers for the development of novel drug delivery systems, their formulation and evaluation

**COURSE CONTENT**

UNIT	CONTENT	No. of Hrs.
I	<b>Controlled drug delivery systems:</b> Introduction, terminology/definitions and rationale, advantages, disadvantages, selection of drug candidates. Approaches to design controlled release formulations based on diffusion, dissolution and ion exchange principles. Physicochemical and biological properties of drugs relevant to controlled release formulations. <b>Polymers:</b> Introduction, classification, properties, advantages and application of polymers in formulation of controlled release drug delivery systems.	10
II	<b>Microencapsulation:</b> Definition, advantages and disadvantages, microspheres/microcapsules, microparticles, methods of microencapsulation, applications <b>Mucosal Drug Delivery system:</b> Introduction, Principles of bioadhesion /mucoadhesion, concepts, advantages and disadvantages, transmucosal permeability and formulation considerations of buccal delivery systems <b>Implantable Drug Delivery Systems:</b> Introduction, advantages and disadvantages, concept of implants and osmotic pump	10
III	<b>Transdermal Drug Delivery Systems:</b> Introduction, Permeation through skin,	10

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	factors affecting permeation, permeation enhancers, basic components of TDDS, formulation approaches <b>Gastroretentive drug delivery systems:</b> Introduction, advantages, disadvantages, approaches for GRDDS – Floating, high density systems, inflatable and gastroadhesive systems and their applications <b>Nasopulmonary drug delivery system:</b> Introduction to Nasal and Pulmonary routes of drug delivery, Formulation of Inhalers (dry powder and metered dose), nasal sprays, nebulizers	
IV	<b>Targeted drug Delivery:</b> Concepts and approaches advantages and disadvantages, introduction to liposomes, niosomes, nanoparticles, monoclonal antibodies and their applications	08
V	<b>Ocular Drug Delivery Systems:</b> Introduction, intra ocular barriers and methods to overcome – Preliminary study, ocular formulations and ocuserts <b>Intrauterine Drug Delivery Systems:</b> Introduction, advantages and disadvantages, development of intra uterine devices (IUDs) and applications	07

### Recommended Books

1. Y W. Chien, Novel Drug Delivery Systems, 2nd edition, revised and expanded, Marcel Dekker, Inc., New York, 1992.
2. Robinson, J. R., Lee V. H. L, Controlled Drug Delivery Systems, Marcel Dekker, Inc., New York, 1992.
3. Encyclopedia of Controlled Delivery. Edith Mathiowitz, Published by Wiley Interscience Publication, John Wiley and Sons, Inc, New York. Chichester/Weinheim
4. N.K. Jain, Controlled and Novel Drug Delivery, CBS Publishers & Distributors, New Delhi, First edition 1997 (reprint in 2001).
5. S.P. Vyas and R.K. Khar, Controlled Drug Delivery -concepts and advances, Vallabh Prakashan, New Delhi, First edition 2002.

  
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### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the basics and application of novel drug delivery systems, rationalize the use and requirements of the polymer for the formulation of a controlled release system.
CO2	To attain knowledge of principles and fundamentals of drug targeting in the design of site-specific drug delivery systems and to understand the process of microencapsulation and other techniques.
CO3	Attain expertise in basic learning and application of Novel Drug Discovery systems in Research, Industry, and Hospitals.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	1	3	3	-	-	-	1	-	-	3	3	1	1
CO2	3	1	3	3	-	-	-	1	-	-	3	3	1	1
CO3	3	1	3	3	-	-	-	1	-	-	3	3	1	1

  
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## BP 705 P. INSTRUMENTAL METHODS OF ANALYSIS (Practical)

### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
I	T	P		Sessional	End Semester Exam	Total	
0	0	4	2	15	35	50	3 hours

- 1 Determination of absorption maxima and effect of solvents on absorption maxima of organic compounds
- 2 Estimation of dextrose by colorimetry
- 3 Estimation of sulfanilamide by colorimetry
- 4 Simultaneous estimation of ibuprofen and paracetamol by UV spectroscopy
- 5 Assay of paracetamol by UV- Spectrophotometry
- 6 Estimation of quinine sulfate by fluorimetry
- 7 Study of quenching of fluorescence
- 8 Determination of sodium by flame photometry
- 9 Determination of potassium by flame photometry
- 10 Determination of chlorides and sulphates by nephelo turbidometry
- 11 Separation of amino acids by paper chromatography
- 12 Separation of sugars by thin layer chromatography
- 13 Separation of plant pigments by column chromatography
- 14 Demonstration experiment on HPLC
- 15 Demonstration experiment on Gas Chromatography

### Recommended Books

1. Instrumental Methods of Chemical Analysis by B.K Sharma
2. Organic spectroscopy by Y.R Sharma
3. Text book of Pharmaceutical Analysis by Kenneth A. Connors
4. Vogel's Text book of Quantitative Chemical Analysis by A.I. Vogel
5. Practical Pharmaceutical Chemistry by A.H. Beckett and J.B. Stenlake
6. Organic Chemistry by I. L. Finar

  
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7. Organic spectroscopy by William Kemp
8. Quantitative Analysis of Drugs by D. C. Garrett
9. Quantitative Analysis of Drugs in Pharmaceutical Formulations by P. D. Sethi
10. Spectrophotometric identification of Organic Compounds by Silverstein

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	5	10	35
Total Marks	50		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn the operation of various analytical instruments used in the analysis of drugs and pharmaceuticals and to perform estimation of pharmaceuticals by analytical techniques.
CO2	To understand the basics and applications of chromatographic techniques, NMR, IR, and MS Spectroscopies
CO3	To attain expertise in basic learning and application of Instrumental Methods of Analysis in Pharmacy, Research, and Industry.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	1	3	3	-	-	-	1	2	-	3	3	3	3
CO2	3	1	3	3	-	-	-	1	2	-	3	3	3	3
CO3	3	1	3	3	-	-	-	1	2	-	3	3	3	3

  
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### BP 706 P: PRACTICE SCHOOL

#### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	-	25	125
Total Marks	150		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To learn the basics of literature review, publications, project designing, designing of experiments, etc.
CO2	To implement CO1 in designing the projects, performing experiments, and developing scientific writing skills.
CO3	Attain expertise in scientific writing, project designing, and implementation.

#### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	2	3	3	-	-	1	3	-	-	3	3	-	-
CO2	3	2	3	3	-	-	1	3	-	-	3	3	-	-
CO3	3	2	3	3	-	-	1	3	-	-	3	3	-	-

  
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**4<sup>th</sup> Year (Semester VIII)**

**PCI Syllabus**

  
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## BP 801 T. BIostatistics AND RESEARCH METHODOLOGY (Theory)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** To understand the applications of Biostatistics in Pharmacy. This subject deals with descriptive statistics, Graphics, Correlation, Regression, logistic regression Probability theory, Sampling technique, Parametric tests, Non Parametric tests, ANOVA, Introduction to Design of Experiments, Phases of Clinical trials and Observational and Experimental studies, SPSS, R and MINITAB statistical software's, analyzing the statistical data using Excel.

**Objectives:** Upon completion of the course the student shall be able to

- Know the operation of M.S. Excel, SPSS, R and MINITAB®, DoE (Design of Experiment)
- Know the various statistical techniques to solve statistical problems
- Appreciate statistical techniques in solving the problems.

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<p><b>Introduction:</b> Statistics, Biostatistics, Frequency distribution</p> <p><b>Measures of central tendency:</b> Mean, Median, Mode- Pharmaceutical examples</p> <p><b>Measures of dispersion:</b> Dispersion, Range, standard deviation, Pharmaceutical problems</p> <p><b>Correlation:</b> Definition, Karl Pearson's coefficient of correlation, Multiple correlation -Pharmaceuticals examples</p>	10
II	<p><b>Regression:</b> Curve fitting by the method of least squares, fitting the lines <math>y = a + bx</math> and <math>x = a + by</math>, Multiple regression, standard error of regression- Pharmaceutical Examples</p> <p><b>Probability:</b> Definition of probability, Binomial distribution, Normal distribution, Poisson's distribution, properties – problems</p>	10

  
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	Sample, Population, large sample, small sample, Null hypothesis, alternative hypothesis, sampling, essence of sampling, types of sampling, Error-I type, Error-II type, Standard error of mean (SEM) - Pharmaceutical examples <b>Parametric test:</b> t-test (Sample, Pooled or Unpaired and Paired), ANOVA, (One way and Two way), Least Significance difference	
III	<b>Non Parametric tests:</b> Wilcoxon Rank Sum Test, Mann-Whitney U test, Kruskal-Wallis test, Friedman Test <b>Introduction to Research:</b> Need for research, Need for design of Experiments, Experiential Design Technique, plagiarism <b>Graphs:</b> Histogram, Pie Chart, Cubic Graph, response surface plot, Counter Plot graph <b>Designing the methodology:</b> Sample size determination and Power of a study, Report writing and presentation of data, Protocol, Cohorts studies, Observational studies, Experimental studies, Designing clinical trial, various phases.	10
IV	Blocking and confounding system for Two-level factorials <b>Regression modeling:</b> Hypothesis testing in Simple and Multiple regression models <b>Introduction to Practical components of Industrial and Clinical Trials</b> <b>Problems:</b> Statistical Analysis Using Excel, SPSS, MINITAB®, DESIGN OF EXPERIMENTS, R -Online Statistical Software's to Industrial and Clinical trial approach	08
V	<b>Design and Analysis of experiments:</b> <b>Factorial Design:</b> Definition, 2 <sup>2</sup> , 2 <sup>3</sup> design. Advantage of factorial design <b>Response Surface methodology:</b> Central composite design, Historical design, Optimization Techniques	07

### Recommended Books

1. Pharmaceutical statistics- Practical and clinical applications, Sanford Bolton, Publisher Marcel Dekker Inc. New York.
2. Fundamental of Statistics – Himalaya Publishing House- S.C. Gupta
3. Design and Analysis of Experiments – PHI Learning Private Limited, R. Pannerselvam,

  
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4. Design and Analysis of Experiments – Wiley Students Edition, Douglas and C. Montgomery

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:


Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the application of biostatistics in Pharmacy and to implement knowledge of Design of Experiments, Phases of Clinical trials, and Observational and Experimental studies, SPSS, R, and MINITAB statistical software.
CO2	To recognize Descriptive statistics, Graphics, Correlation, Regression, logistic regression Probability theory, Sampling technique, Parametric tests, non-Parametric tests, ANOVA, etc.
CO3	Attain expertise in statistical analysis of the results and attain proficiency in the theoretical and practical implementation of Biostatistics.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	1	1	3	3	-	-	-	-	-	-	3	3	-	-
CO2	1	1	3	3	-	-	-	-	-	-	3	3	-	-
CO3	1	1	3	3	-	-	-	-	-	-	3	3	-	-

  
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**BP 802 T: SOCIAL AND PREVENTIVE PHARMACY**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

#### Scope:

The purpose of this course is to introduce to students a number of health issues and their challenges. This course also introduced a number of national health programmes. The roles of the pharmacist in these contexts are also discussed.

#### Objectives:

After the successful completion of this course, the student shall be able to:

- Acquire high consciousness/realization of current issues related to health and pharmaceutical problems within the country and worldwide.
- Have a critical way of thinking based on current healthcare development.
- Evaluate alternative ways of solving problems related to health and pharmaceutical issues

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Concept of health and disease:</b> Definition, concepts and evaluation of public health. Understanding the concept of prevention and control of disease, social causes of diseases and social problems of the sick. <b>Social and health education:</b> Food in relation to nutrition and health, Balanced diet, Nutritional deficiencies, Vitamin deficiencies, Malnutrition and its prevention. <b>Sociology and health:</b> Socio cultural factors related to health and disease, Impact of urbanization on health and disease, Poverty and health <b>Hygiene and health:</b> personal hygiene and health care; avoidable habits	10
II	<b>Preventive medicine:</b> General principles of prevention and control of diseases such as cholera, SARS, Ebola virus, influenza, acute respiratory infections, malaria, chickenpox, dengue, lymphatic filariasis, pneumonia, hypertension,	10

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	diabetes mellitus, cancer, drug addiction-drug substance abuse	
<b>III</b>	<b>National health programs, its objectives, functioning and outcome of the following:</b> HIV AND AIDS control programme, TB, Integrated disease surveillance program (IDSP), National leprosy control programme, National mental health program, National programme for prevention and control of deafness, Universal immunization programme, National programme for control of blindness, Pulse polio programme.	<b>10</b>
<b>IV</b>	National health intervention programme for mother and child, National family welfare programme, National tobacco control programme, National Malaria Prevention Program, National programme for the health care for the elderly, Social health programme; role of WHO in Indian national program	<b>08</b>
<b>V</b>	Community services in rural, urban and school health: Functions of PHC, Improvement in rural sanitation, national urban health mission, Health promotion and education in school.	<b>07</b>

### Recommended Books

1. Short Textbook of Preventive and Social Medicine, Prabhakara GN, 2nd Edition, 2010, ISBN: 9789380704104, JAYPEE Publications.
2. Textbook of Preventive and Social Medicine (Mahajan and Gupta), Edited by Roy Rabindra Nath, Saha Indranil, 4th Edition, 2013, ISBN: 9789350901878, JAYPEE Publications.
3. Review of Preventive and Social Medicine (Including Biostatistics), Jain Vivek, 6<sup>th</sup> Edition, 2014, ISBN: 9789351522331, JAYPEE Publications.
4. Essentials of Community Medicine—A Practical Approach, Hiremath Lalita D, Hiremath Dhananjaya A, 2nd Edition, 2012, ISBN: 9789350250440, JAYPEE Publications.
5. Park Textbook of Preventive and Social Medicine, K Park, 21st Edition, 2011, ISBN-14: 9788190128285, BANARSIDAS BHANOT PUBLISHERS.
6. Community Pharmacy Practice, Ramesh Adepu, BSP publishers, Hyderabad.

  
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## Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand various Healthcare issues and policies in India and around the globe and to develop awareness about the number of national health programmes.
CO2	To address current issues related to health and pharmaceutical problems within the country and worldwide.
CO3	To attain proficiency in theoretical and practical aspects of Social and Preventive Pharmacy.

## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	1	3	-	1	2	-	2	2	-	3	-	3	-
CO2	3	1	3	-	1	2	-	2	2	-	3	-	3	-
CO3	3	1	3	-	1	2	-	2	2	-	3	-	3	-

  
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**BP 803 ET. PHARMA MARKETING MANAGEMENT (Theory)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** The pharmaceutical industry not only needs highly qualified researchers, chemists and technical people, but also requires skilled managers who can take the industry forward by managing and taking the complex decisions which are imperative for the growth of the industry. The Knowledge and Know-how of marketing management groom the people for taking a challenging role in Sales and Product management.

**Course Objective:** The course aims to provide an understanding of marketing concepts and techniques and their applications in the pharmaceutical industry.

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>Marketing:</b> Definition, general concepts and scope of marketing; Distinction between marketing & selling; Marketing environment; Industry and competitive analysis; Analyzing consumer buying behavior; industrial buying behavior. <b>Pharmaceutical market:</b> Quantitative and qualitative aspects; size and composition of the market; demographic descriptions and socio-psychological characteristics of the consumer; market segmentation & targeting. Consumer profile; Motivation and prescribing habits of the physician; patients' choice of physician and retail pharmacist. Analyzing the Market; Role of market research.	10
II	<b>Product decision:</b> Classification, product line and product mix decisions, product lifecycle, product portfolio analysis; product positioning; New product decisions; Product branding, packaging and labeling decisions, Product management in pharmaceutical industry.	10
III	<b>Promotion:</b> Methods, determinants of promotional mix, promotional budget; An	10

  
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	overview of personal selling, advertising, direct mail, journals, sampling, retailing, medicalexhibition, public relations, online promotional techniques for OTC Products.	
IV	<b>Pharmaceutical marketing channels:</b> Designing channel, channel members, selecting the appropriate channel, conflict in channels, physical distribution management: Strategic importance, tasks in physical distribution management. <b>Professional sales representative (PSR):</b> Duties of PSR, purpose of detailing, selection and training, supervising, norms for customer calls, motivating, evaluating, compensation and future prospects of the PSR.	10
V	<b>Pricing:</b> Meaning, importance, objectives, determinants of price; pricing methods and strategies, issues in price management in pharmaceutical industry. An overview of DPCO (Drug Price Control Order) and NPPA (National Pharmaceutical Pricing Authority). <b>Emerging concepts in marketing:</b> Vertical & Horizontal Marketing; Rural Marketing; Consumerism; Industrial Marketing; Global Marketing.	10

### Recommended Books

1. Philip Kotler and Kevin Lane Keller: Marketing Management, Prentice Hall of India, New Delhi.
2. Walker, Boyd and Larreche : Marketing Strategy- Planning and Implementation, Tata MC Graw Hill, New Delhi.
3. Dhruv Grewal and Michael Levy: Marketing, Tata MC Graw Hill
4. Arun Kumar and N Menakshi: Marketing Management, Vikas Publishing, India
5. Rajan Saxena: Marketing Management; Tata MC Graw-Hill (India Edition)
6. Ramaswamy, U.S & Nanakamari, S: Marketing Management: Global Perspective, Indian Context, Macmillan India, New Delhi.
7. Shanker, Ravi: Service Marketing, Excell Books, New Delhi.
7. Subba Rao Changanti, Pharmaceutical Marketing in India (GIFT – Excel series) Excel Publications.

  
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### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand and learn marketing concepts, principles, and techniques and their applications in the pharmaceutical industry.
CO2	To attain expertise in administration management, its sections, and functions.
CO3	To execute the skills of management in the inventory of drugs and evaluate product design, market channel, and pricing.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	3	-	1	2	-	2	-	-	3	-	-	2
CO2	3	-	3	-	1	2	-	2	-	-	3	-	-	2
CO3	3	-	3	-	1	2	-	2	-	-	3	-	-	2

  
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## BP 804 ET: PHARMACEUTICAL REGULATORY SCIENCE (Theory)

### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This course is designed to impart the fundamental knowledge on the regulatory requirements for approval of new drugs, and drug products in regulated markets of India & other countries like US, EU, Japan, Australia, UK etc. It prepares the student to learn in detail on the regulatory requirements, documentation requirements, and registration procedures for marketing the drug products.

**Objectives:** Upon completion of the subject student shall be able to;

- Know about the process of drug discovery and development
- Know the regulatory authorities and agencies governing the manufacture and sale of pharmaceuticals
- Know the regulatory approval process and their registration in Indian and international markets

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	<b>New Drug Discovery and development</b> Stages of drug discovery, Drug development process, pre-clinical studies, non-clinical activities, clinical studies, Innovator and generics, Concept of generics, Generic drug product development.	10
II	<b>Regulatory Approval Process</b> Approval processes and timelines involved in Investigational New Drug (IND), New Drug Application (NDA), Abbreviated New Drug Application (ANDA). Changes to an approved NDA / ANDA. <b>Regulatory authorities and agencies</b> Overview of regulatory authorities of India, United States, European Union,	10

  
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	Australia, Japan, Canada (Organization structure and types of applications)	
<b>III</b>	<b>Registration of Indian drug product in overseas market</b> Procedure for export of pharmaceutical products, Technical documentation, Drug Master Files (DMF), Common Technical Document (CTD), electronic Common Technical Document (eCTD), ASEAN Common Technical Document (ACTD) research.	<b>10</b>
<b>IV</b>	<b>Clinical trials</b> Developing clinical trial protocols, Institutional Review Board / Independent Ethics committee - formation and working procedures, Informed consent process and procedures, GCP obligations of Investigators, sponsors & Monitors, Managing and Monitoring clinical trials, Pharmacovigilance - safety monitoring in clinical trials	<b>08</b>
<b>V</b>	<b>Regulatory Concepts</b> Basic terminology, guidance, guidelines, regulations, Laws and Acts, Orange book, Federal Register, Code of Federal Regulatory, Purple book	<b>07</b>

### Recommended Books

1. Drug Regulatory Affairs by Sachin Itkar, Dr. N.S. Vyawahare, Nirali Prakashan.
2. The Pharmaceutical Regulatory Process, Second Edition Edited by Ira R. Berry and Robert P. Martin, Drugs and the Pharmaceutical Sciences, Vol. 185. Informa Healthcare Publishers.
3. New Drug Approval Process: Accelerating Global Registrations By Richard A Guarino, MD, 5th edition, Drugs and the Pharmaceutical Sciences, Vol. 190.
4. Guidebook for drug regulatory submissions / Sandy Weinberg. By John Wiley & Sons. Inc.
5. FDA Regulatory Affairs: a guide for prescription drugs, medical devices, and biologics / edited by Douglas J. Pisano, David Mantus.
6. Generic Drug Product Development, Solid Oral Dosage forms, Leon Shargel and Isader Kaufer, Marcel Dekker series, Vol. 143
7. Clinical Trials and Human Research: A Practical Guide to Regulatory Compliance By Fay A. Rozovsky and Rodney K. Adams
8. Principles and Practices of Clinical Research, Second Edition Edited by John I. Gallin and Frederick P. Ognibene

  
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9. Drugs: From Discovery to Approval, Second Edition By Rick Ng

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand the regulatory requirements for approval of new drugs, and drug products in regulated markets of India & other countries like the US, EU, Japan, Australia, UK, etc.
CO2	To apply the knowledge of regulatory requirements, documentation requirements, and registration procedures for marketing approval of the drug products.
CO3	To implement the knowledge of pharmacovigilance in the safety monitoring of clinical trials.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	-	-	1	-	-	1	-	-	1	1	-	-
CO2	3	-	-	-	1	-	-	1	-	-	1	1	-	-
CO3	3	-	-	-	1	-	-	1	-	-	1	1	-	-

  
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**BP 808 ET: CELL AND MOLECULAR BIOLOGY (Elective subject)**

  
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### Teaching and Examination Scheme:

Teaching Scheme			Credits C	Marks			Duration of End Semester Examination
L	T	P		Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** Cell biology is a branch of biology that studies cells – their physiological properties, their structure, the organelles they contain, interactions with their environment, their life cycle, division, death and cell function. This is done both on a microscopic and molecular level. Cell biology research encompasses both the great diversity of single-celled organisms like bacteria and protozoa, as well as the many specialized cells in multi-cellular organisms such as humans, plants, and sponges.

**Objectives:** Upon completion of the subject student shall be able to;

- Summarize cell and molecular biology history.
- Summarize cellular functioning and composition.
- Describe the chemical foundations of cell biology.
- Summarize the DNA properties of cell biology.
- Describe protein structure and function.
- Describe cellular membrane structure and function.
- Describe basic molecular genetic mechanisms.
- Summarize the Cell Cycle

### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	a) Cell and Molecular Biology: Definitions theory and basics and Applications. b) Cell and Molecular Biology: History and Summation. c) Properties of cells and cell membrane. d) Prokaryotic versus Eukaryotic e) Cellular Reproduction f) Chemical Foundations – an Introduction and Reactions (Types)	10
II	a) DNA and the Flow of Molecular Information b) DNA Functioning c) DNA and RNA	10

  
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	d) Types of RNA e) Transcription and Translation	
<b>III</b>	a) Proteins: Defined <b>and</b> Amino Acids b) Protein Structure c) Regularities in Protein Pathways d) Cellular Processes e) Positive Control and significance of Protein Synthesis	<b>10</b>
<b>IV</b>	a) Science of Genetics b) Transgenics and Genomic Analysis c) Cell Cycle analysis d) Mitosis and Meiosis e) Cellular Activities and Checkpoints	<b>08</b>
<b>V</b>	a) Cell Signals: Introduction b) Receptors for Cell Signals c) Signaling Pathways: Overview d) Misregulation of Signaling Pathways e) Protein-Kinases: Functioning	<b>07</b>

### Recommended Books

1. W.B. Hugo and A.D. Russel: Pharmaceutical Microbiology, Blackwell Scientific publications, Oxford London.
2. Prescott and Dunn., Industrial Microbiology, 4th edition, CBS Publishers & Distributors, Delhi.
3. Pelczar, Chan Kreig, Microbiology, Tata McGraw Hill edn.
4. Malcolm Harris, Balliere Tindall and Cox: Pharmaceutical Microbiology.
5. Rose: Industrial Microbiology.
6. Probisher, Hinsdill et al: Fundamentals of Microbiology, 9th ed. Japan
7. Cooper and Gunn's: Tutorial Pharmacy, CBS Publisher and Distribution.
8. Pepler: Microbial Technology.
9. Edward: Fundamentals of Microbiology.
10. N.K. Jain: Pharmaceutical Microbiology, Vallabh Prakashan, Delhi
11. Bergeys manual of systematic bacteriology, Williams and Wilkins- A Waverly company

  
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12. B.R. Glick and J.J. Pasternak: Molecular Biotechnology: Principles and Applications of Recombinant DNA: ASM Press Washington D.C.
13. RA Goldshy et. al.,: Kuby Immunology.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand and learn the basic physiology of the cell, its functioning, and the mechanism of communication.
CO2	To understand and learn the signaling cascade operating within the cell, with special reference to Receptors, ion channels, transporters, and other transducer mechanisms.
CO3	To understand and learn the basic concept of genetics and the pattern of inheritance.

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	-	1	-	-	-	-	-	-	3	1	-	-
CO2	3	-	-	1	-	-	-	-	-	-	3	1	-	-
CO3	3	-	-	1	-	-	-	-	-	-	3	1	-	-

  
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**BP 809 ET. COSMETIC SCIENCE (Theory)**

#### Teaching and Examination Scheme:

Teaching Scheme			Credits	Marks			Duration of End Semester Examination
L	T	P	C	Sessional	End Semester Exam	Total	
3	1	0	4	25	75	100	3 hours

**Scope:** This subject is designed to impart the basic knowledge of cosmetic sciences

**Objectives :** Upon completion of the course the student shall be able to,

- Appreciate the applications of various commonly used cosmetic excipients.
- Appreciate and demonstrate the various formulation methods used in cosmetic formulations
- Appreciate and demonstrate the importance of analytical methods used in cosmetics.

#### COURSE CONTENT

UNIT	CONTENT	No. of Hrs.
I	Classification of cosmetic and cosmeceutical products ; Definition of cosmetics as per Indian and EU regulations, Evolution of cosmeceuticals from cosmetics, cosmetics as quasi and OTC drugs. <b>Cosmetic excipients:</b> Surfactants, rheology modifiers, humectants, emollients, preservatives. Classification and application. <b>Skin:</b> Basic structure and function of skin. <b>Hair:</b> Basic structure of hair. Hair growth cycle. <b>Oral Cavity:</b> Common problem associated with teeth and gums.	10
II	<b>Principles of formulation and building blocks of skin care products:</b> Face wash, Moisturizing cream, Cold Cream, Vanishing cream and their advantages and disadvantages. Application of these products in formulation of cosmeceuticals. <b>Antiperspirants &amp; deodorants-</b> Actives & mechanism of action. <b>Principles of formulation and building blocks of Hair care products:</b> Conditioning shampoo, Hair conditioner, anti-dandruff shampoo. Hair oils. Chemistry and formulation of Para-phenylene diamine based hair dye. Principles of formulation and building blocks of oral care products: Toothpaste for bleeding gums, sensitive teeth. Teeth whitening, Mouthwash.	10
III	Sun protection, Classification of Sunscreens and SPF.	10

  
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	<b>Role of herbs in cosmetics:</b> Skin Care: Aloe and turmeric ; Hair care: Henna and amla Oral care: Neem and clove. <b>Analytical cosmetics:</b> BIS specification and analytical methods for shampoo, skincream and toothpaste.	
IV	Principles of Cosmetic Evaluation: Principles of sebumeter, corneometer. Measurement of TEWL, Skin Color, Hair tensile strength, Hair combing properties Soaps, and syndet bars. Evolution and skin benefits.	08
V	Oily and dry skin, causes leading to dry skin, skin moisturisation. Basic understanding of the terms Comedogenic, dermatitis. Cosmetic problems associated with Hair and scalp: Dandruff, Hair fall causes Cosmetic problems associated with skin: blemishes, wrinkles, acne, prickly heat and body odor. Antiperspirants and Deodorants- Actives and mechanism of action	07

### Recommended Books

1. Harry's Cosmeticology, Wilkinson, Moore, Seventh Edition, George Godwin.
2. Cosmetics – Formulations, Manufacturing and Quality Control, P.P. Sharma, 4<sup>th</sup> Edition, Vandana Publications Pvt. Ltd., Delhi.
3. Text book of cosmeticology by Sanju Nanda & Roop K. Khar, Tata Publishers.

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Theory		
	Internal Assessment	Mid Term Assessment	End Term Assessment
Marks	10	15	75
Total Marks	100		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To understand and learn the principles for the formulation of creams, deodorants, and hair products.
CO2	To interpret various oral problems and the various cosmetics associated with oral care.
CO3	To apply the knowledge of natural sources for the preparation of cosmetics and to evaluate the efficacy of herbal products over synthetic ones.

  
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## CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	-	-	1	-	-	-	-	-	-	3	1	1	1
CO2	3	-	-	1	-	-	-	-	-	-	3	1	1	1
CO3	3	-	-	1	-	-	-	-	-	-	3	1	1	1

  
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## BP 813 PW: PROJECT WORK

### Assessment Pattern- Internal and External

The performance of students is evaluated as follows:

Components	Practical		
	Continuous Assessment	Mid Term Assessment	End Term Assessment
Marks	-	25	125
Total Marks	150		

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To design comprehensive project work in the discipline of interest.
CO2	To perform a rigorous literature search and required experiments.
CO3	To compile a project report, end semester presentation of the project work, and publication (if any).

### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	3	2	3	3	-	-	1	3	-	-	3	3	-	-
CO2	3	2	3	3	-	-	1	3	-	-	3	3	-	-
CO3	3	2	3	3	-	-	1	3	-	-	3	3	-	-

  
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## VALUE ADDED COURSES

### Course 1: National Service Scheme (NSS)

#### Assessment Pattern

The performance of students enrolled in the NSS unit will be regularly monitored and evaluated by the institute, NSS Himachal Pradesh University, and the NSS Programme Officer of the institute. The volunteers who complete a total tenure of three years or 240 h of community service, along with participation at one compulsory special NSS camp will be considered eligible for the NSS Certification by NSS Himachal Pradesh University, Shimla. Special weightage will be given to volunteers who participate in the state level, national level, RD, or Pre RD camp or perform exceptionally well during the volunteer ship.

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To develop written, oral, and; listening skills of the students with a prime focus on overall personality development.
CO2	To raise a sense of social awareness and work with the community for the betterment of society.
CO3	To motivate students and the local community to work towards social causes like health awareness, blood donation, election awareness, environmental protection, disaster management, etc.

#### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	-	2	1	-	3	-	-	3	-	2	3	-	-	-
CO2	-	2	1	-	3	-	-	3	-	2	3	-	-	-
CO3	-	2	1	-	3	-	-	3	-	2	3	-	-	-

  
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## VALUE ADDED COURSES

### Course 2: National Cadet Corps (NCC)

#### Assessment Pattern

The performance of the cadets enrolled in the NCC unit will be regularly monitored and evaluated by the institute, 8HP Bn NCC Rampur, and NCC ANO/ CTO. The cadets who complete a total tenure of three years, along with participation at two compulsory special NCC camps will be considered eligible for the NCC Certification by PHHP&C Chandigarh, NCC Gp HQ Shimla, 8 HP Bn NCC, Rampur. Special weightage will be given to volunteers who participate in the state level, national level, RD, or Pre RD camp or perform exceptionally well during the NCC programme.

**Course Outcomes:** On completion of this course, the students are expected to:

CO1	To develop written, oral, listening, and presentation skills of the students with a prime focus on overall personality development
CO2	To raise a sense of social awareness, and patriotism and work for the betterment of society.
CO3	To motivate students and the local community to work towards social causes like health awareness, blood donation, election awareness, environmental protection, disaster management, etc.

#### CO-PO Mapping

Course Outcome	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3
CO1	-	2	1	-	3	-	-	3	-	2	3	-	-	-
CO2	-	2	1	-	3	-	-	3	-	2	3	-	-	-
CO3	-	2	1	-	3	-	-	3	-	2	3	-	-	-

  
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